

# Metal–Organic Frameworks: From Structure to Function

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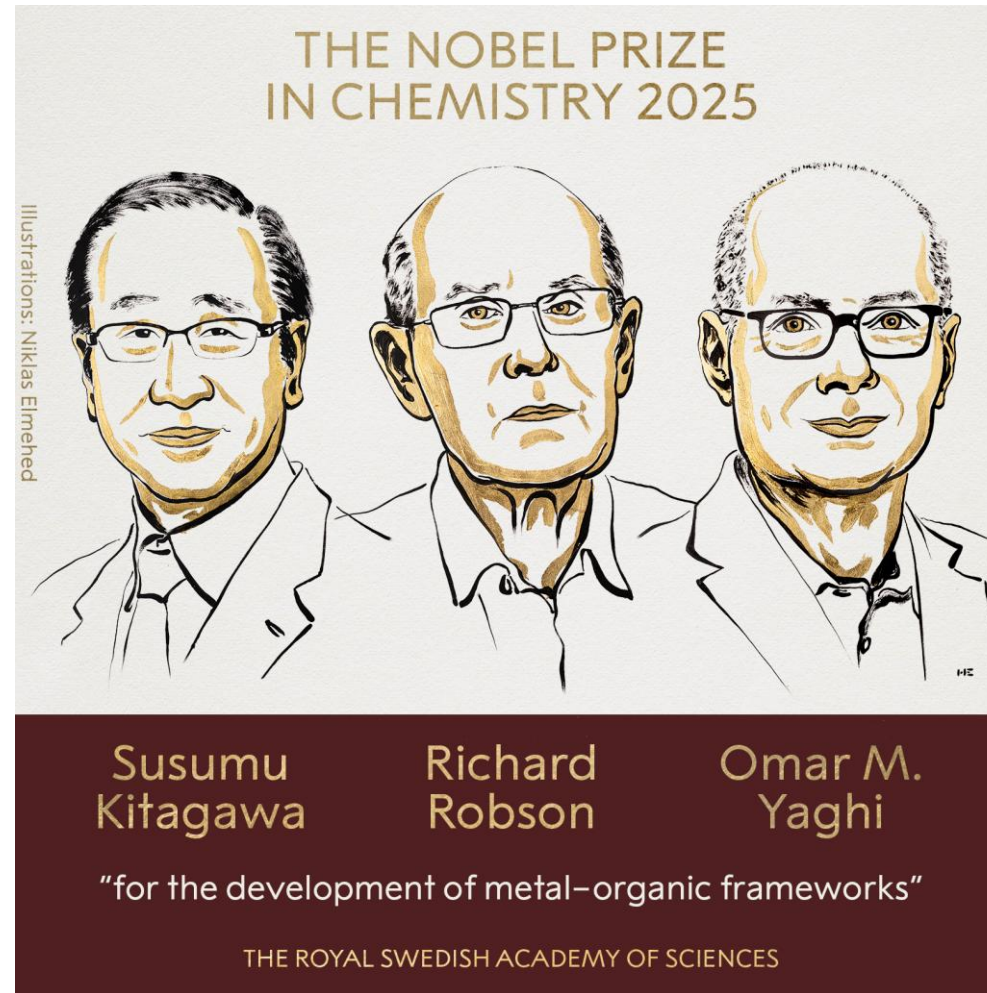
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November 14<sup>th</sup> 2025



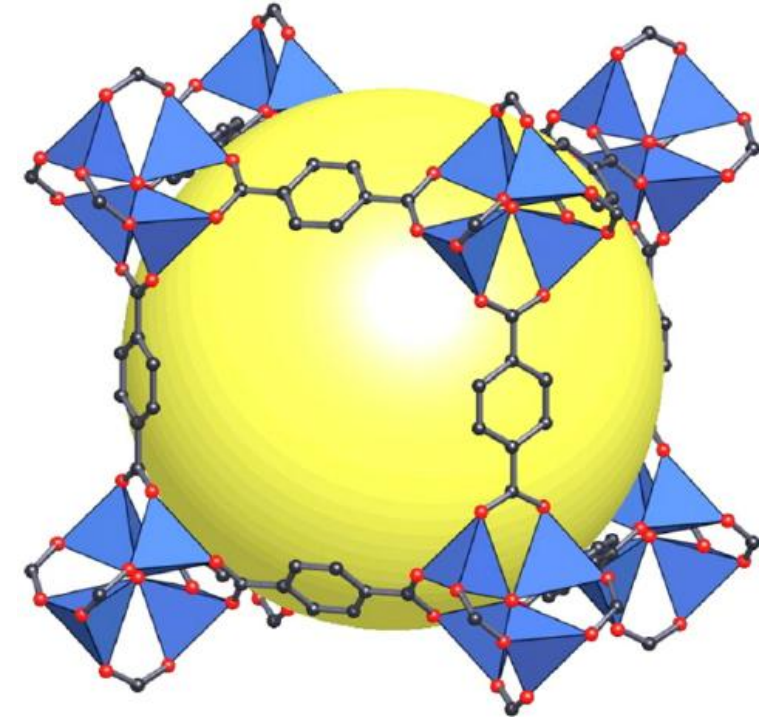
UNIVERSITY OF  
LIVERPOOL

# MOFs won the Nobel Prize in Chemistry!

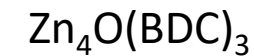


# Metal–organic frameworks (MOFs)

- Metal–organic frameworks (MOFs) are network solids which consist of metal ions or clusters connected to organic linkers
- Scope to modulate their properties via:
  - Metal centres
  - Linker
  - Defect formation
- Their cage-like structure typically leads to a void space within the framework known as a ‘pore’
- This porosity leads to high surface areas



**MOF-5 (IRMOF-1)**

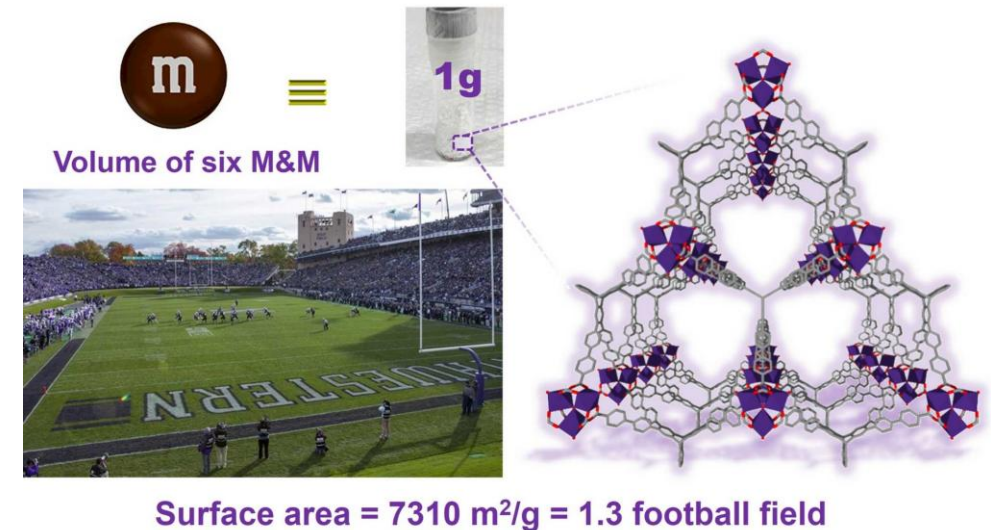


Zn centres with

1,4-benzenedicarboxylate (BDC) linkers

# Metal–organic frameworks (MOFs)

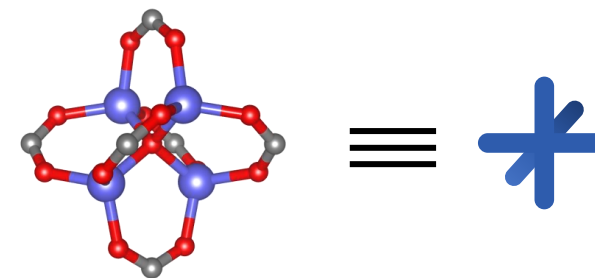
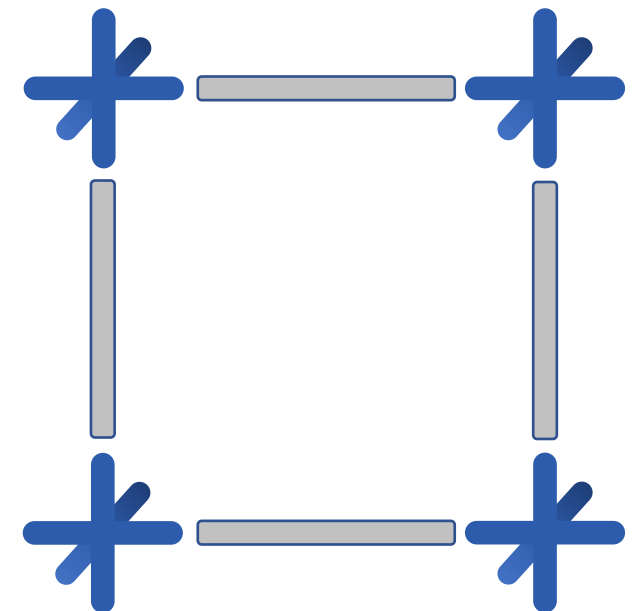
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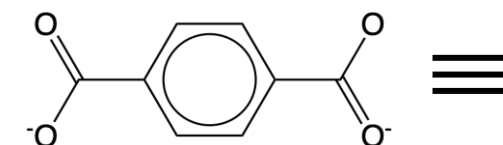
**1g of MOF can have a greater surface area than a football field!**

# Metal–organic frameworks (MOFs)

- Framework is formed by polydentate ligands (organic linkers) coordinating to metal centres
- Carboxylate linkers are most common, but various functionalities, such as imidazoles are possible
- SBUs – small, well-defined clusters of metal ions that act as repeating nodes in the structure



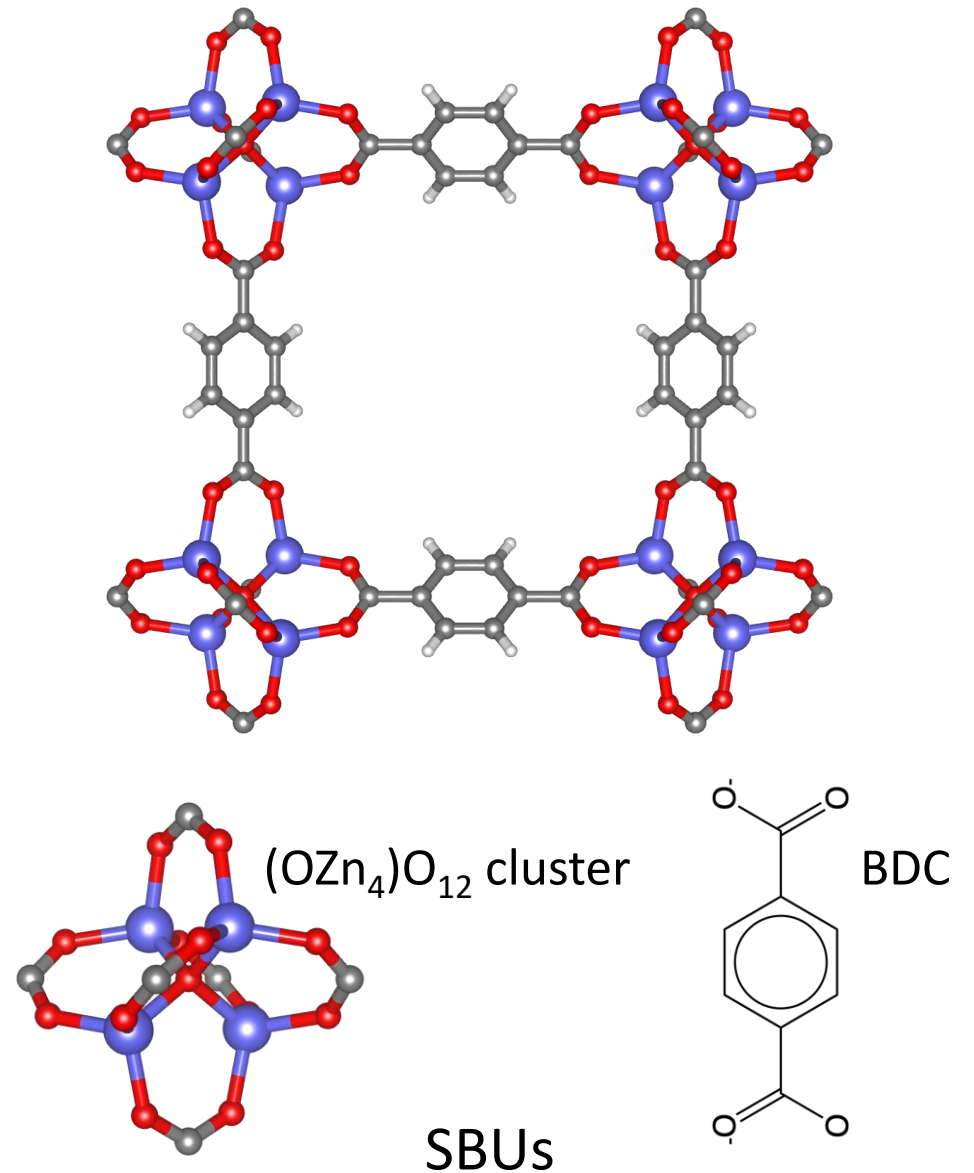
$(\text{OZn}_4)\text{O}_{12}$  cluster



Benzene-1,4-dicarboxylate BDC

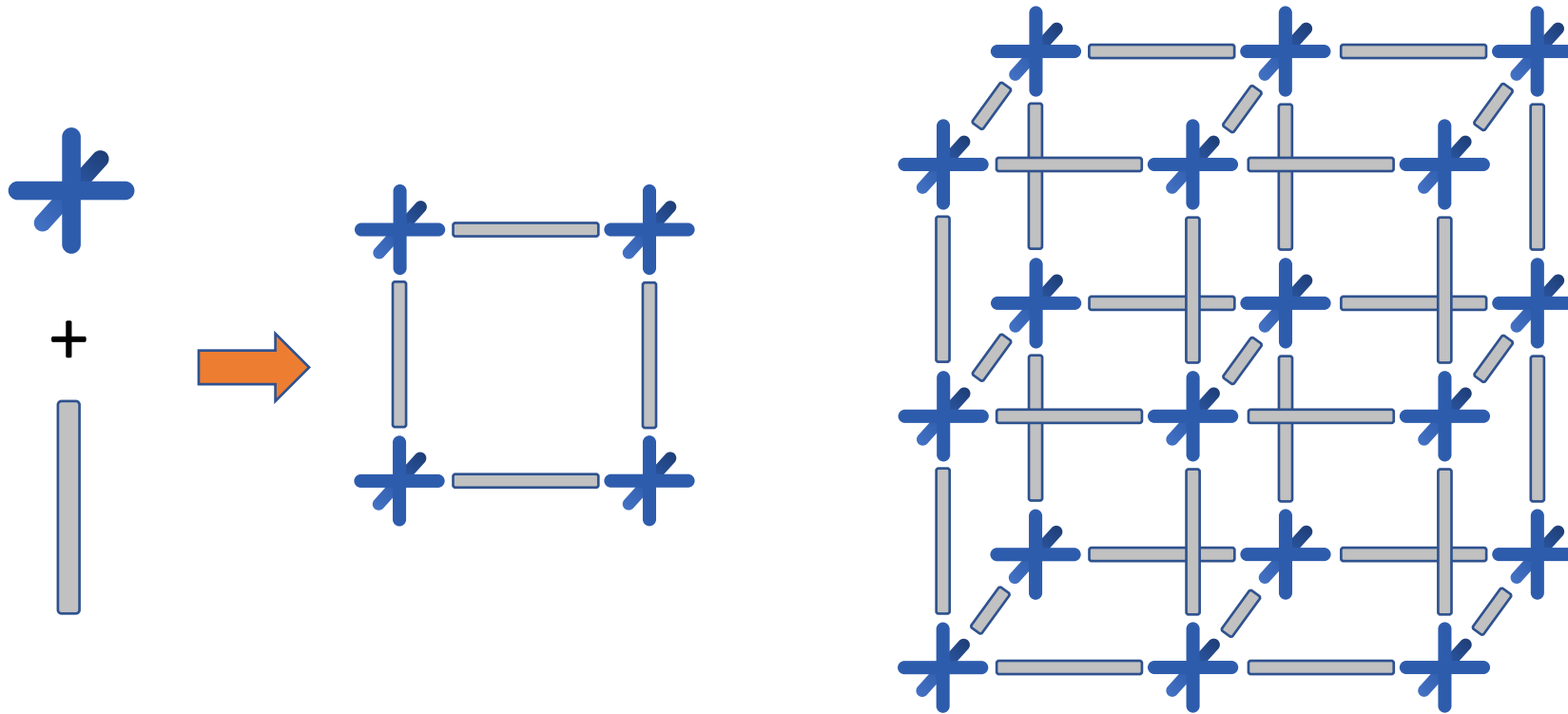
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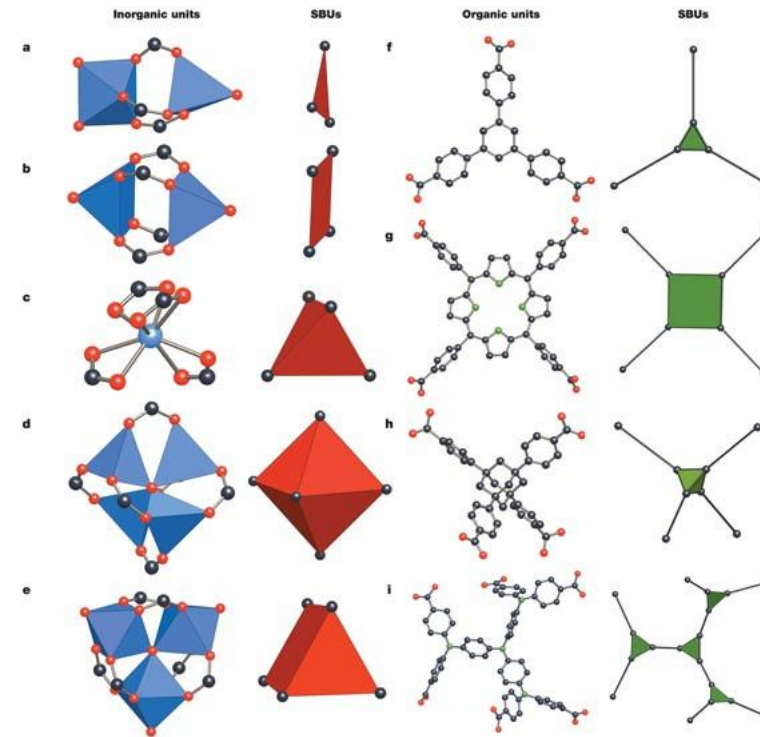
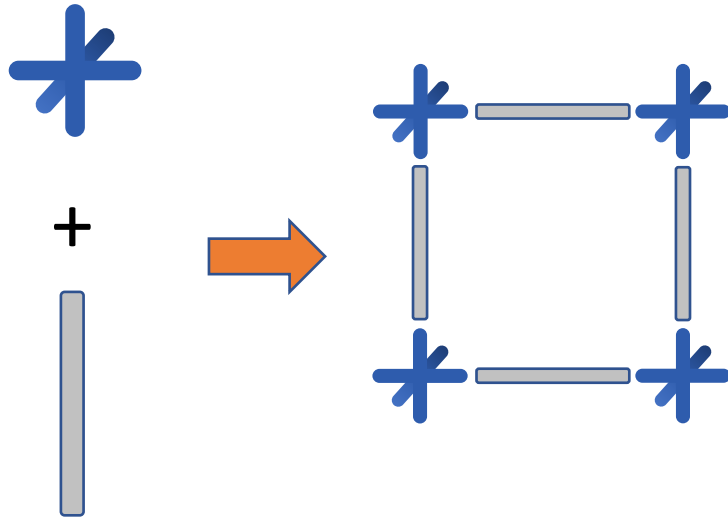


# Reticular chemistry



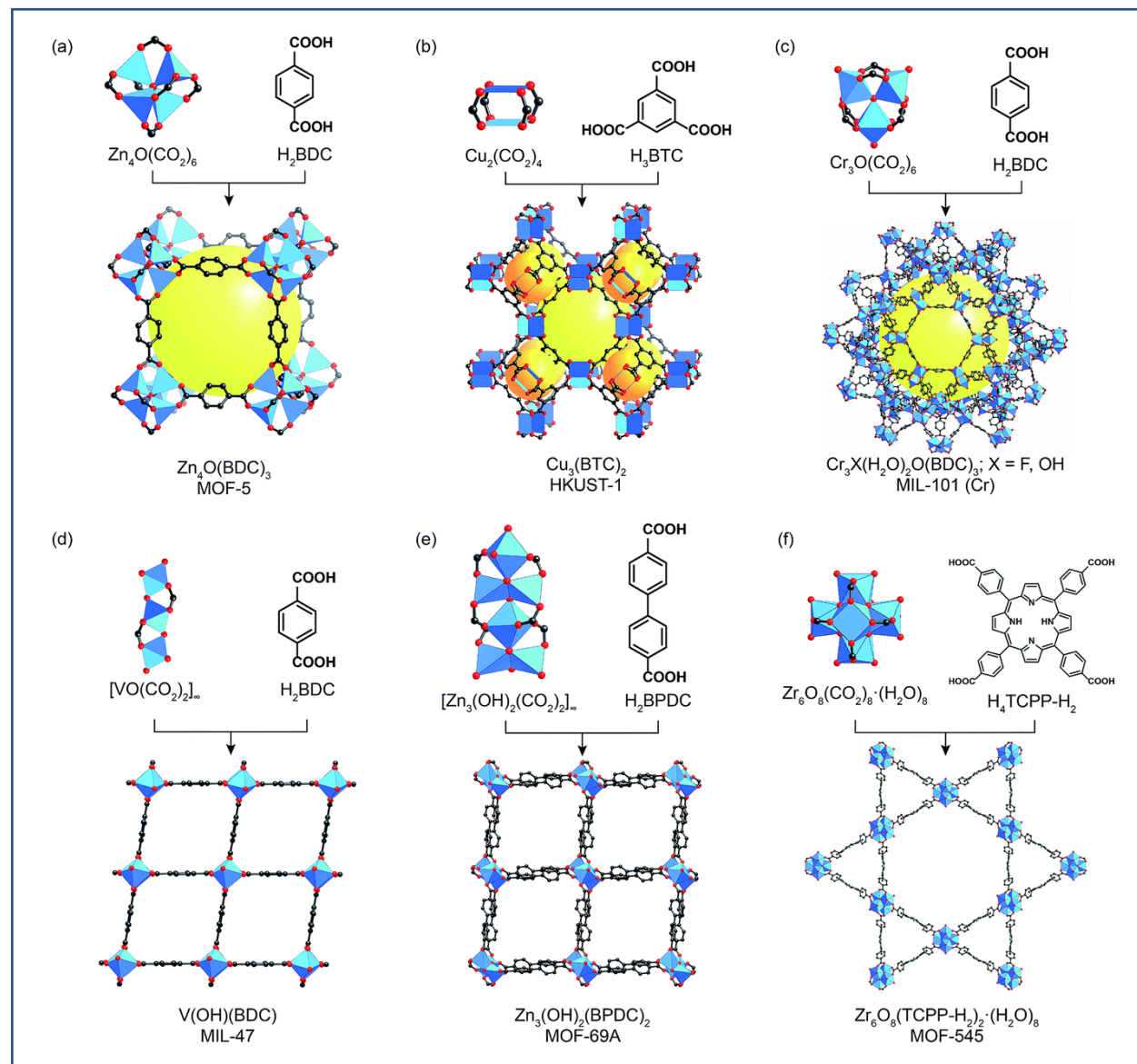
- Chemical components form nodes and linkers in a network
- The geometrical diversity of metal ion and clusters combines with synthetic diversity of organic linkers
- Enables 'controlled' assembly of extended architectures

# Reticular chemistry

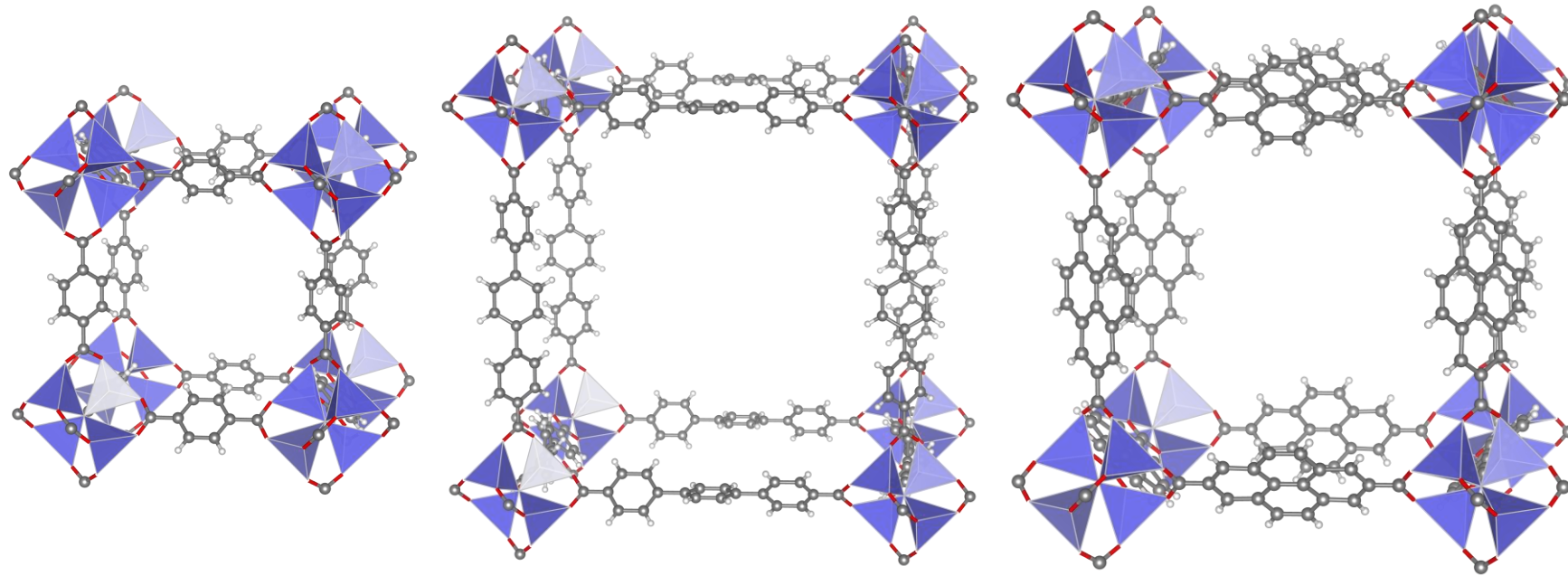


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# Crystal engineering: Isoreticular MOFs

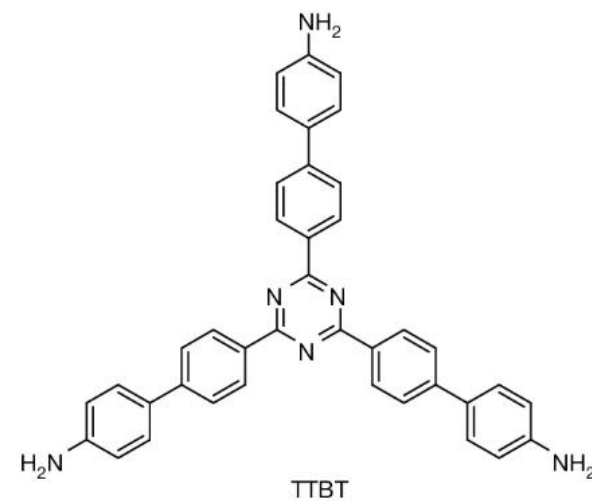
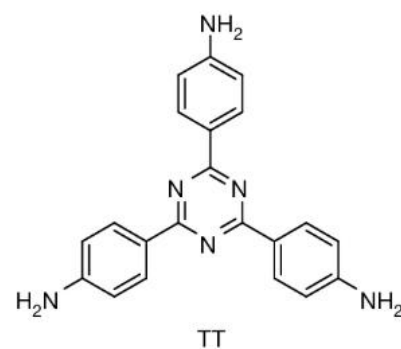
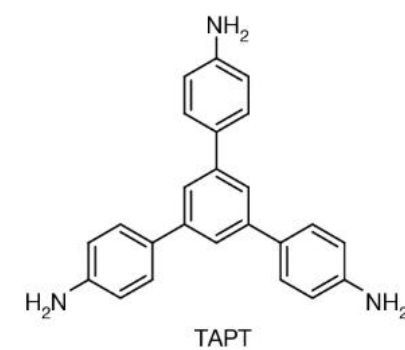
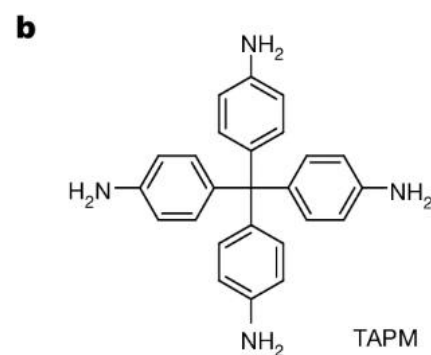
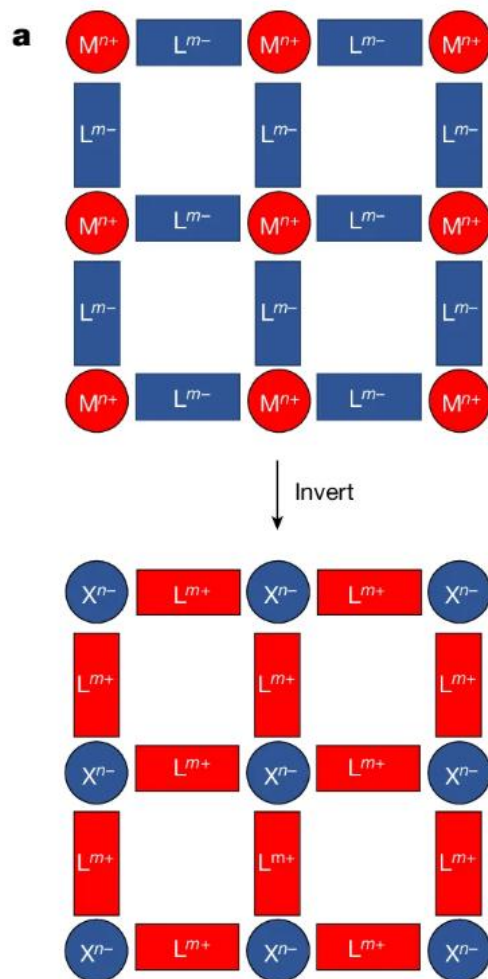


- For some families of MOFs it is possible to change the organic linker and maintain the same framework type
- Relatively strong metal-linker interactions are conserved, but the linker can be swapped
- Modulates the pore size and chemical environment of guests

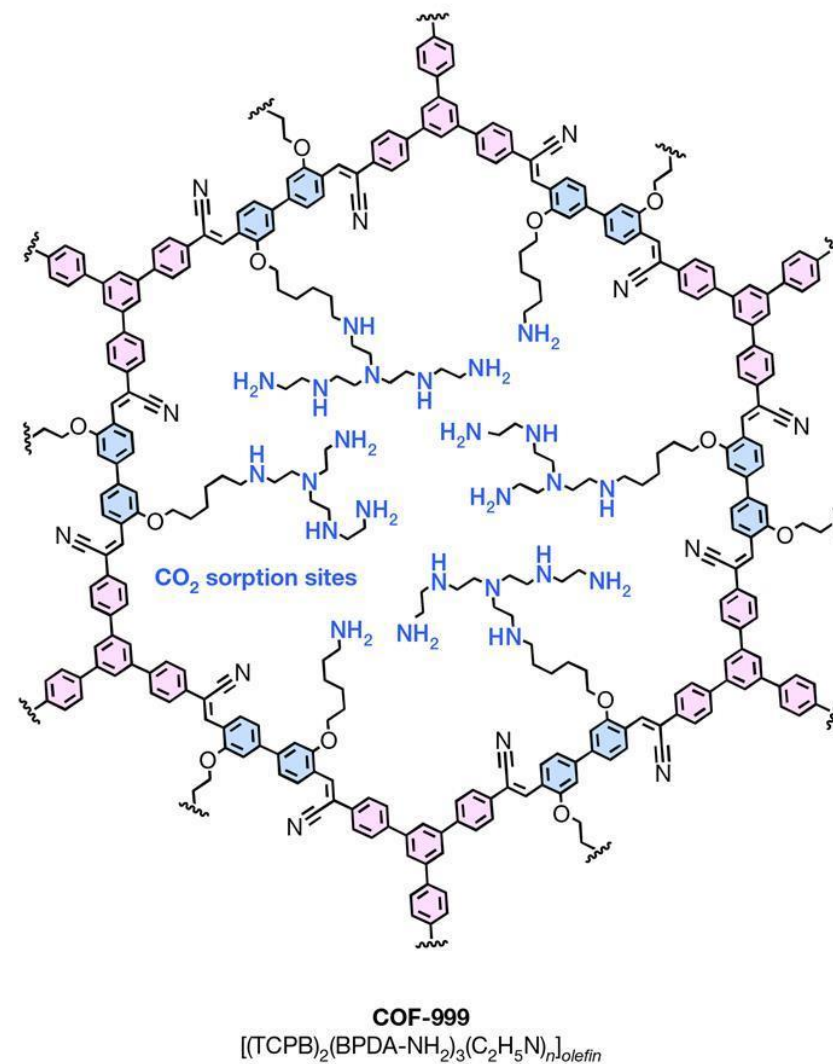
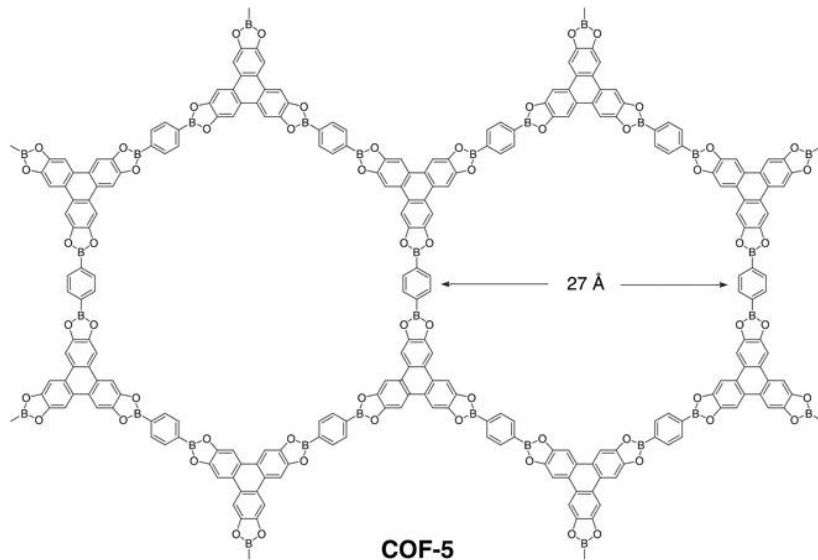
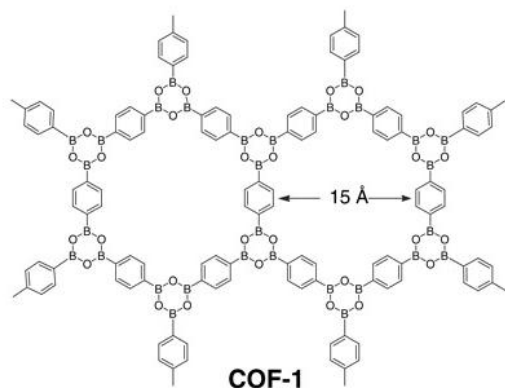
# Porous isorecticular non-metal organic frameworks



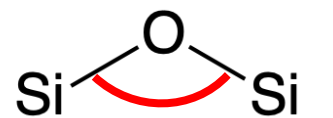
Andy Cooper



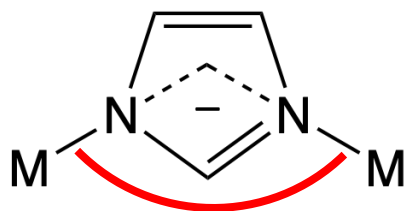
# Covalent organic frameworks (COFs)



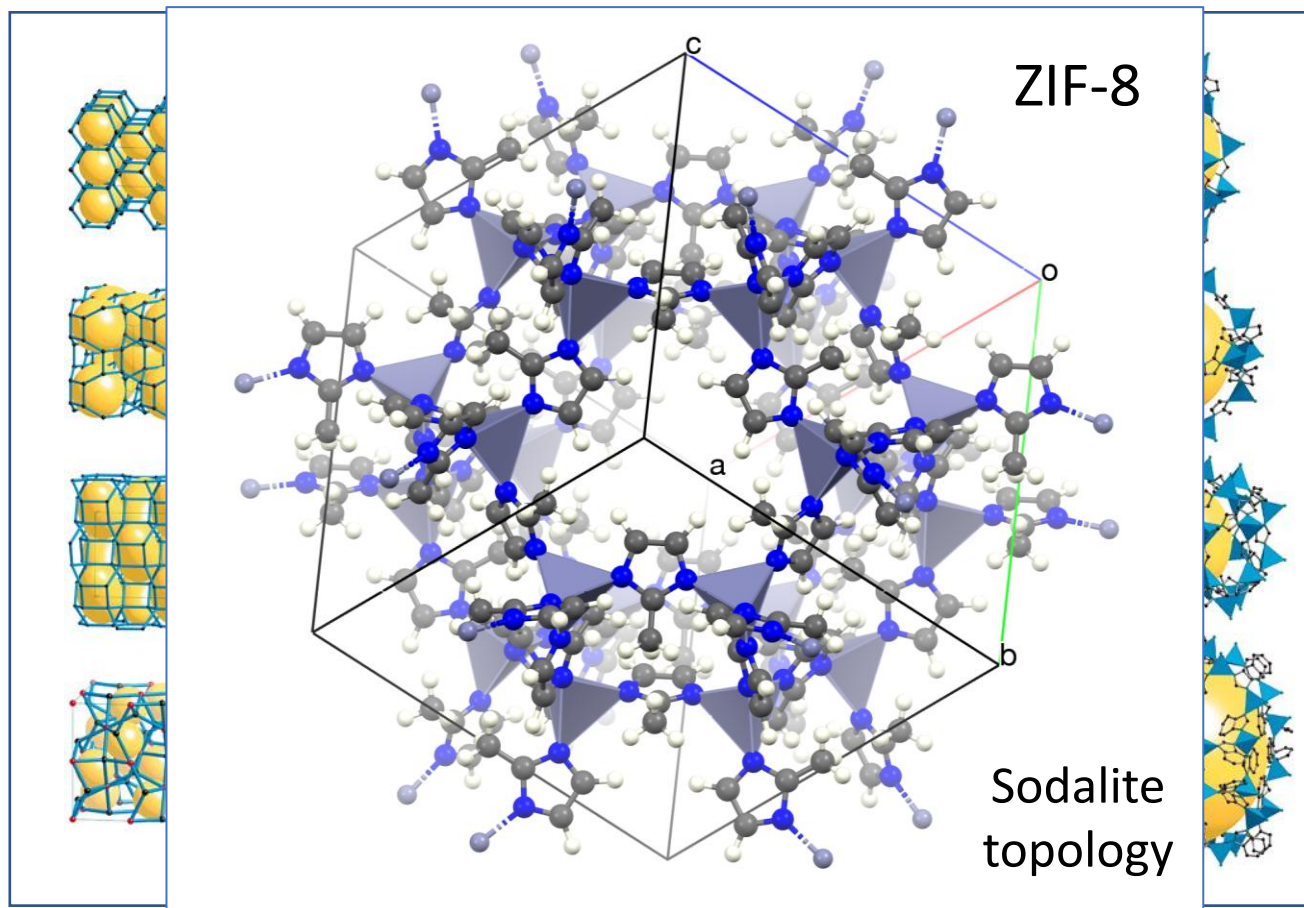
# Zeolitic imidazolate frameworks (ZIFs)



$\angle \text{Si-O-Si} \approx 145^\circ$

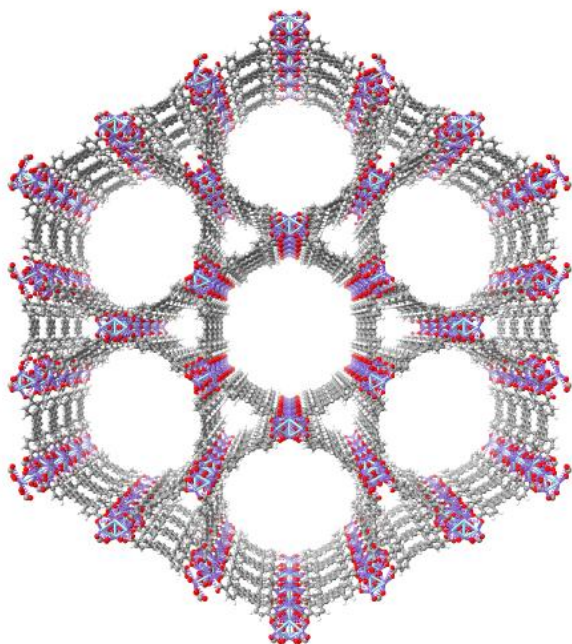


$\angle \text{M-Im-M} \approx 145^\circ$

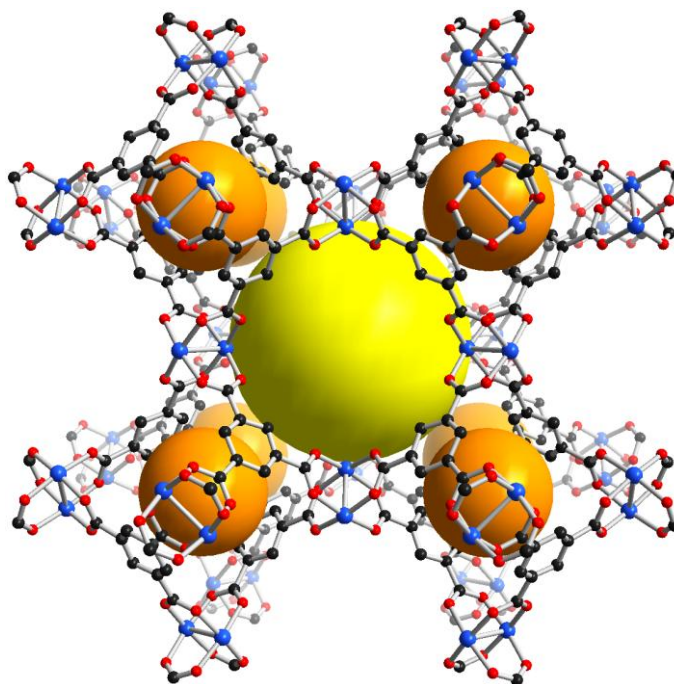




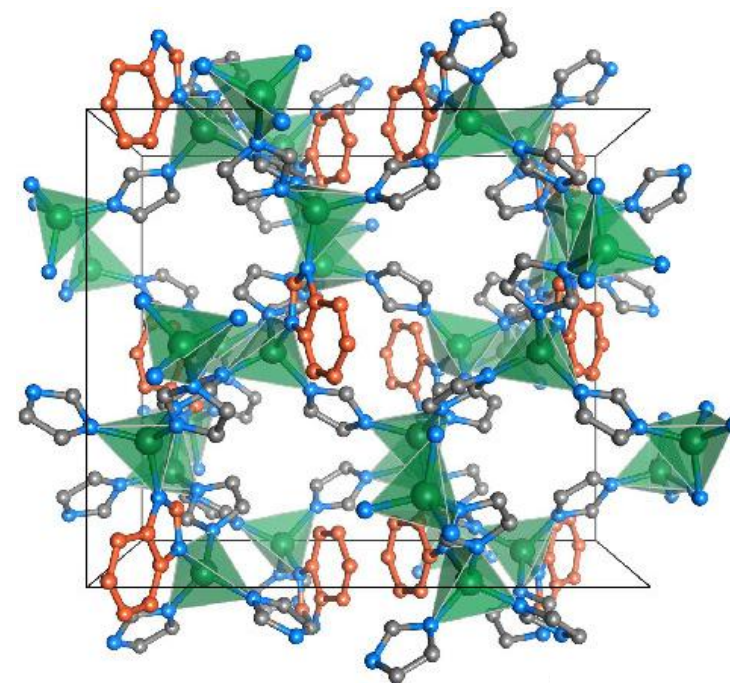
# Some common MOFs



NU-1000  
Zirconium-based  
(yellow)



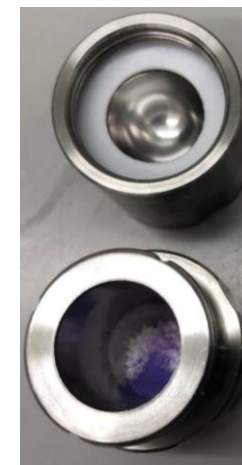
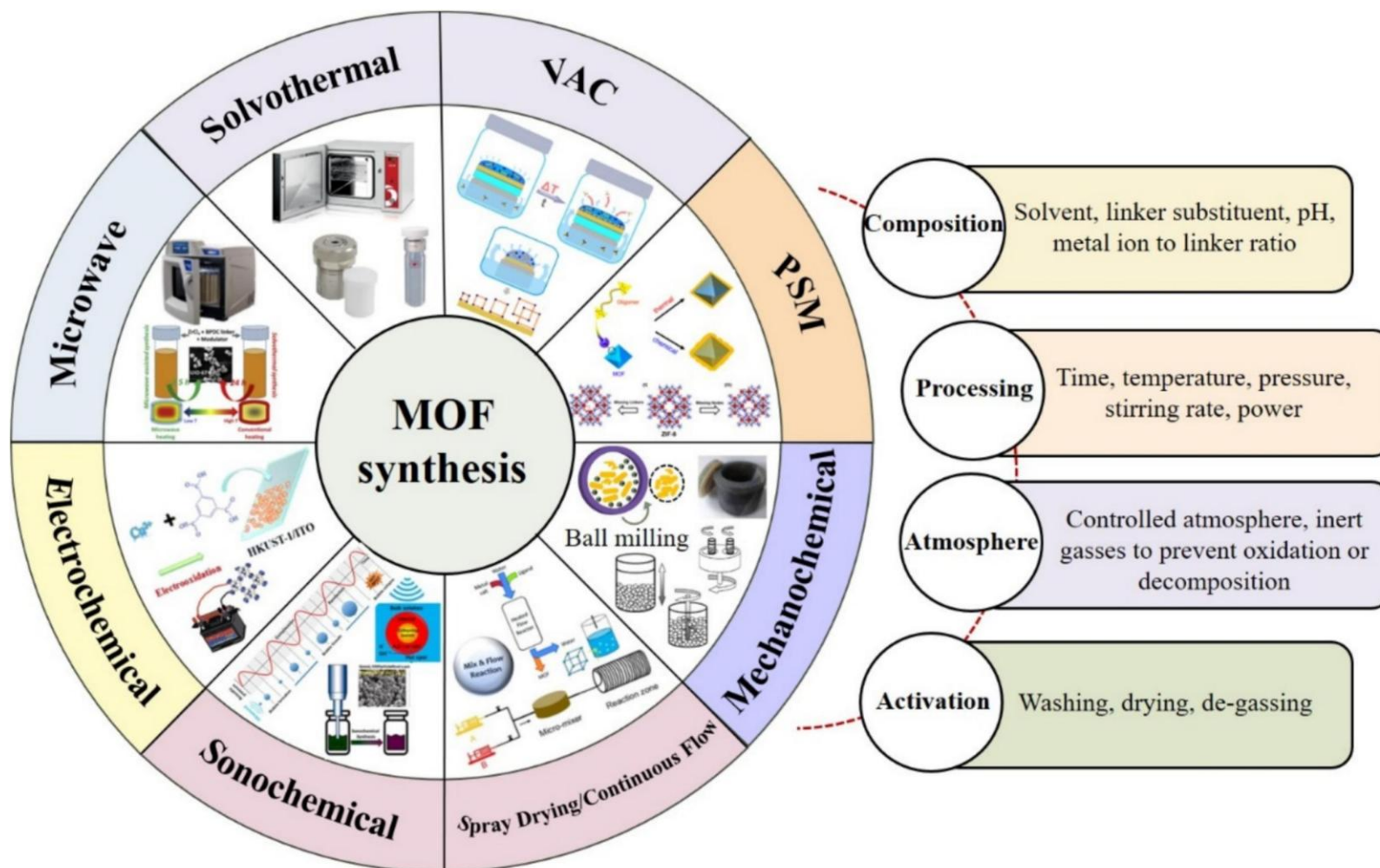
HKUST-1  
Copper-based  
(blue)



ZIF-62 (Zn/Co)  
Zinc or cobalt-based  
(colourless or purple)

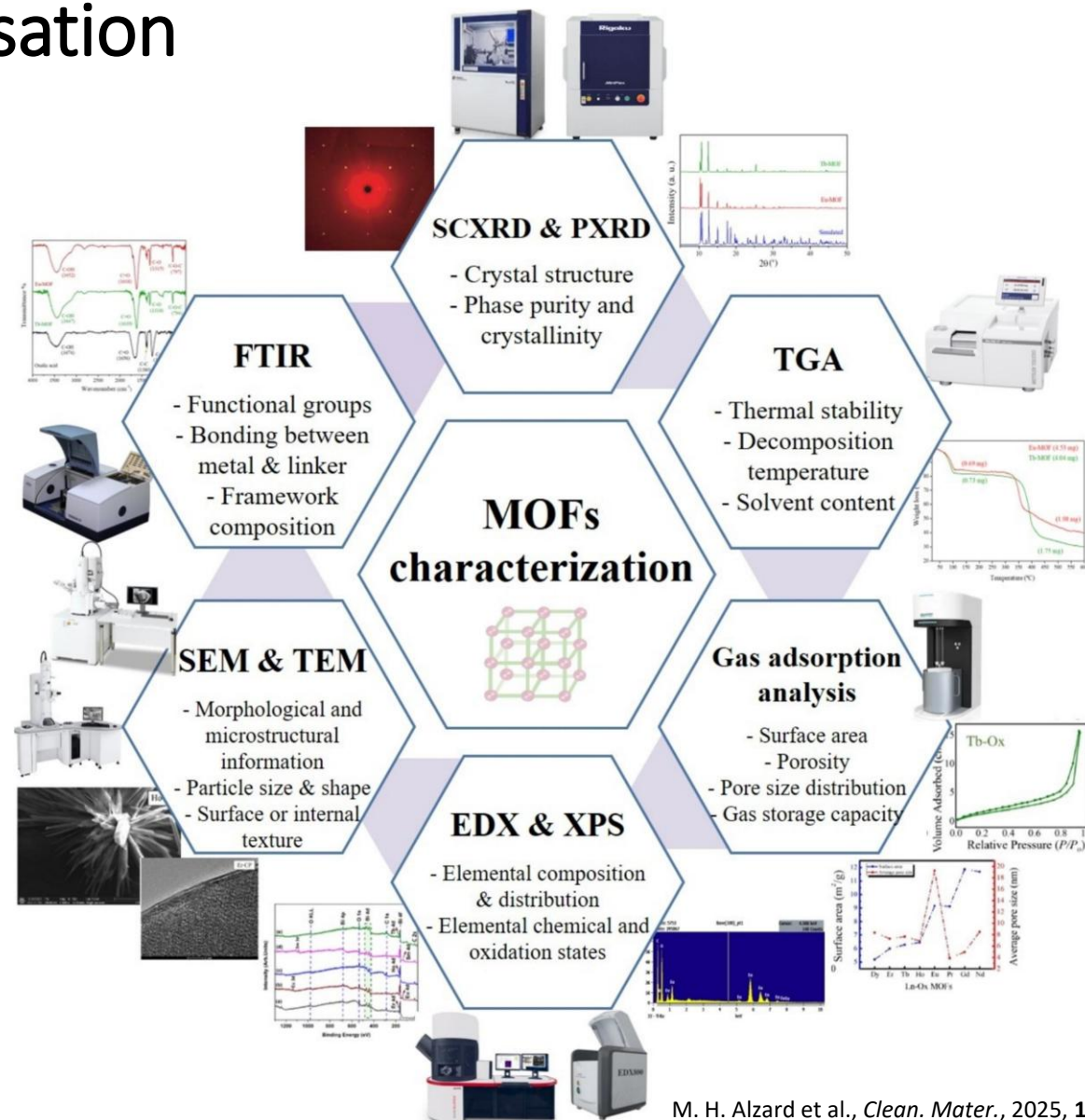


# MOF synthesis



M. H. Alzard et al., *Clean. Mater.*, 2025, **16**, 100314.

# MOF characterisation



M. H. Alzard et al., *Clean. Mater.*, 2025, **16**, 100314.

# Tuning MOF properties

- **Metal centres**

- Structure type – change coordination
- Catalytic properties, *e.g.*, open metal sites

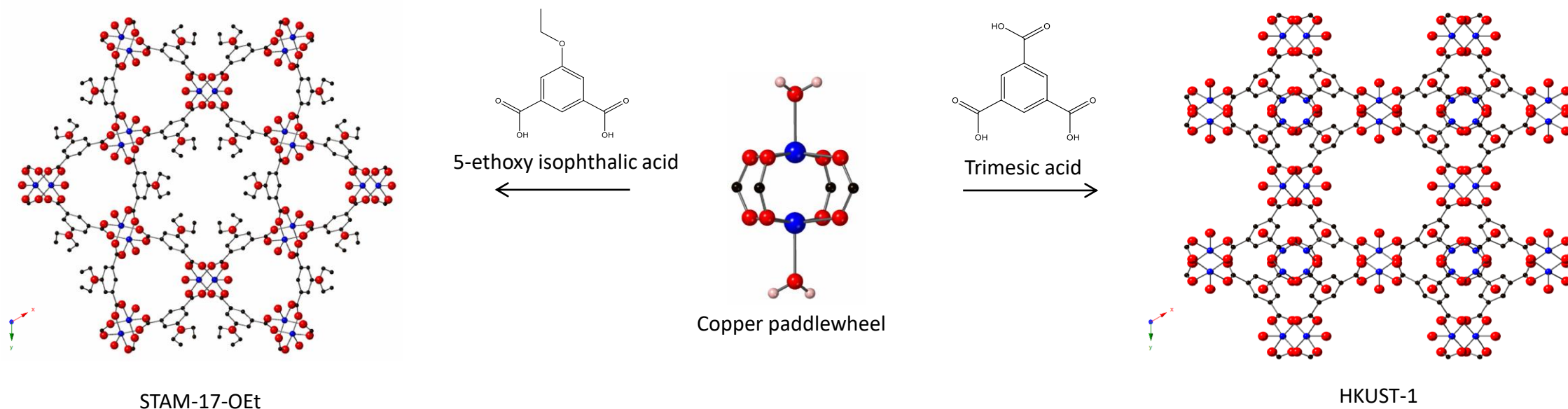
- **Organic linker**

- Essentially endless possibilities
- Structure type – change denticity
- Maintain structure type – change pore dimensions (IRMOFs)
- Chemical environment in pores – hydrophobicity, catalytic, polarity, specific chemical functionalities

- **Synthesis and post-synthetic modifications**

- **Defect engineering**

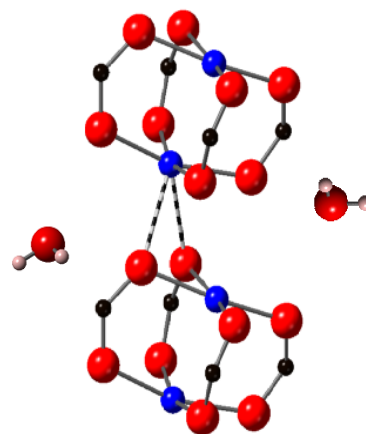
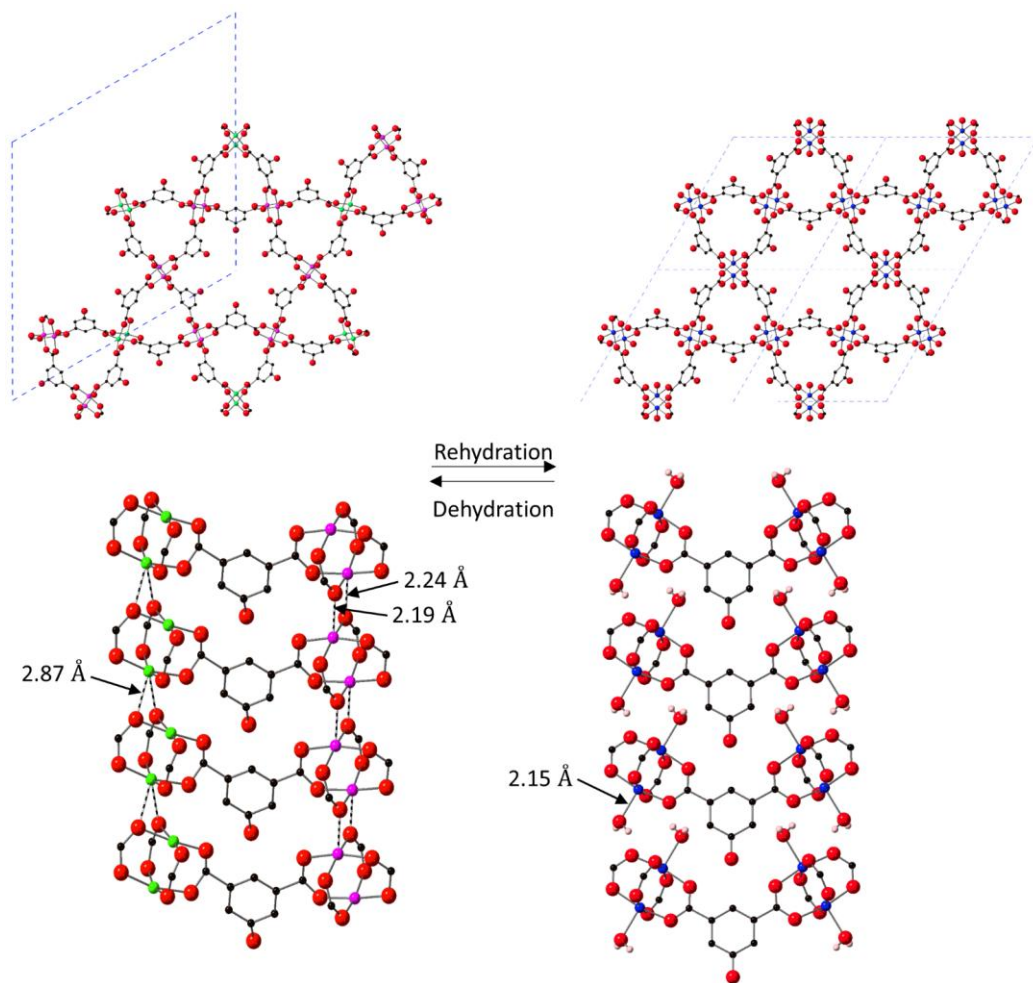
# Water stability in copper MOFs



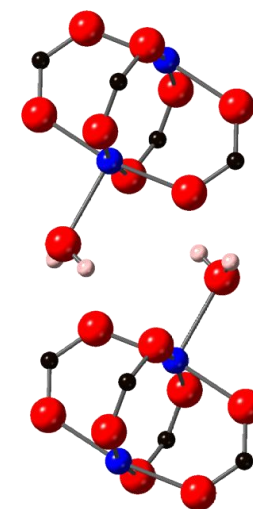
- Copper paddlewheel dimer units present in both structures.
- Similarities in syntheses and structures allow the comparison of properties.
- HKUST-1 is affected by long-term water instability, is STAM-17-OEt similarly affected?



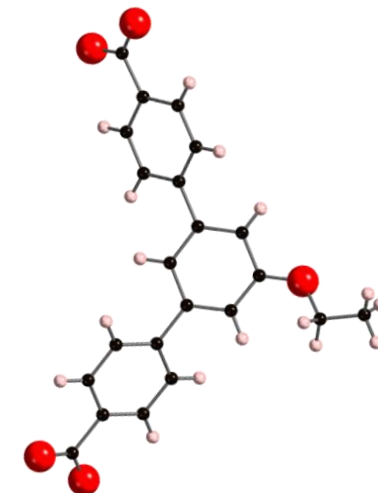
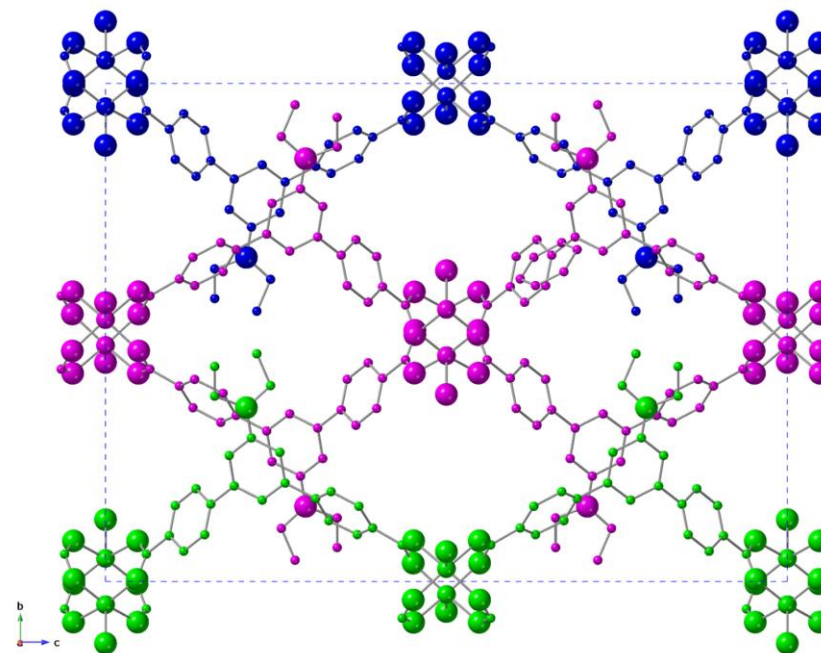
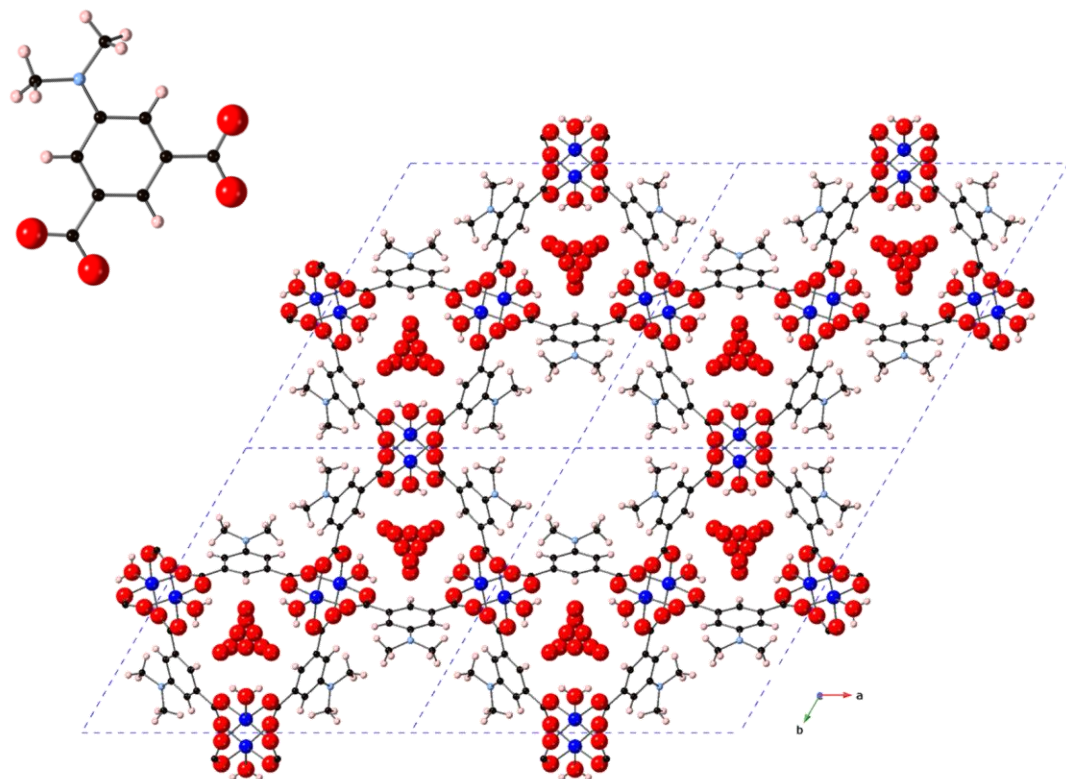
# Hemilability in copper MOFs



“Crumple zone mechanism”

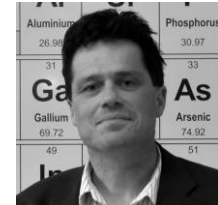


# Structural variations

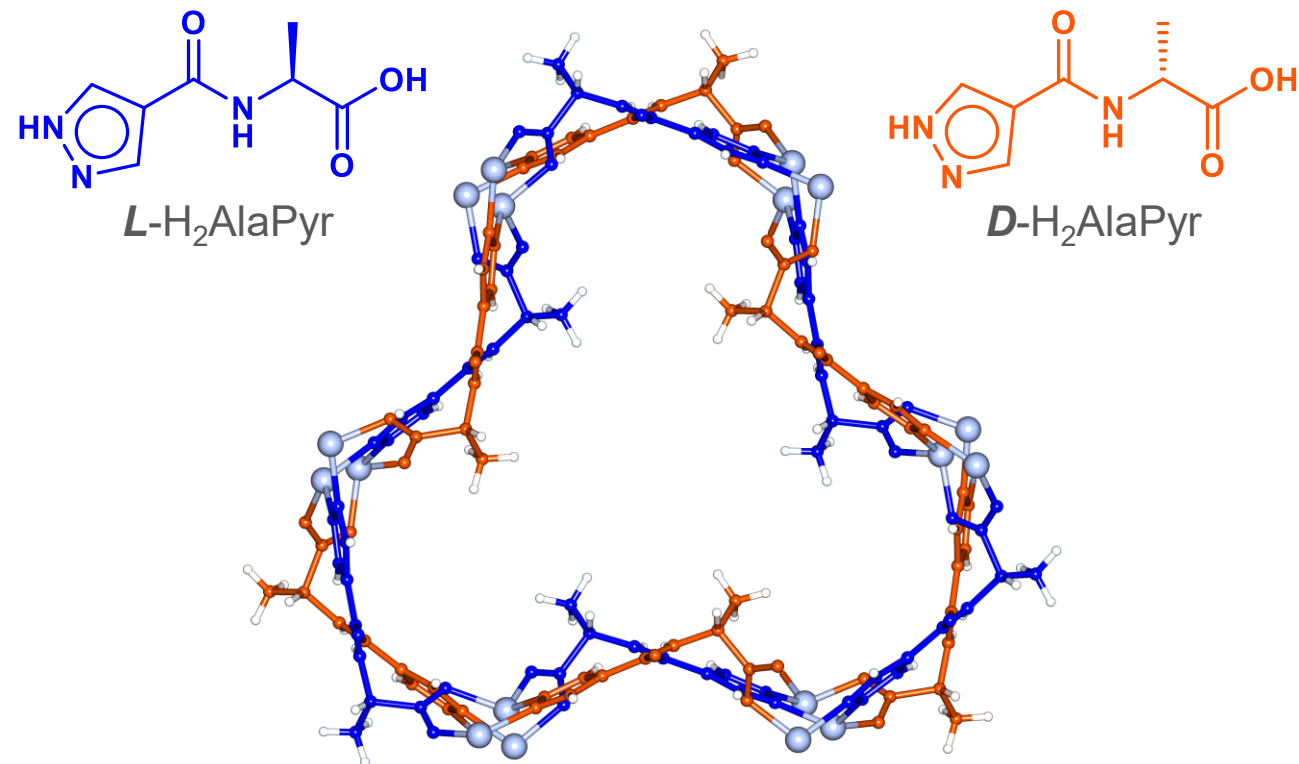
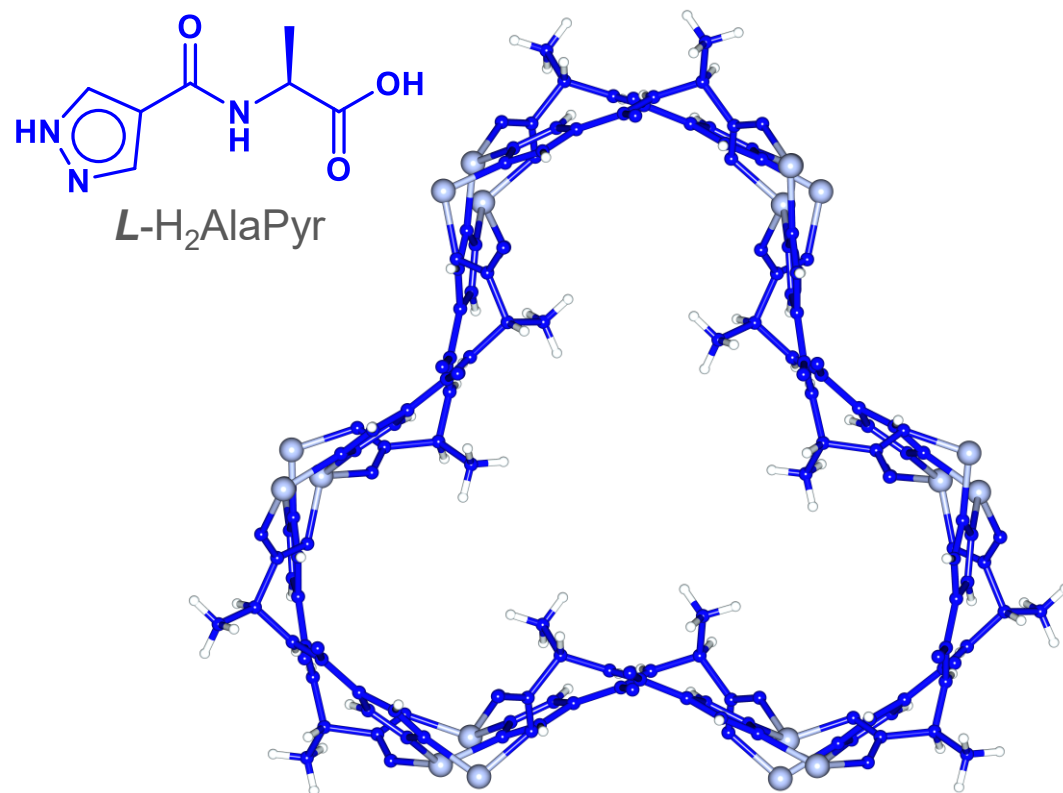




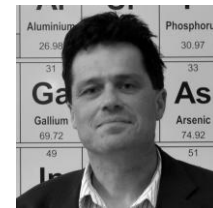
# Ordered sidechains of flexible MOFs



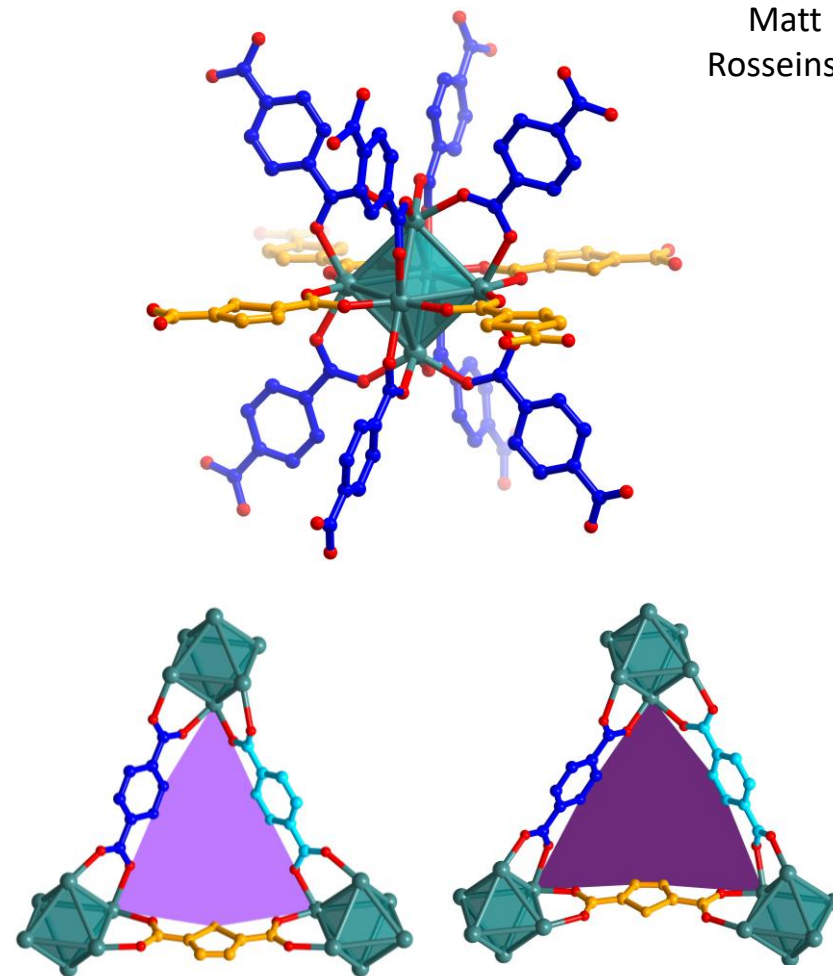
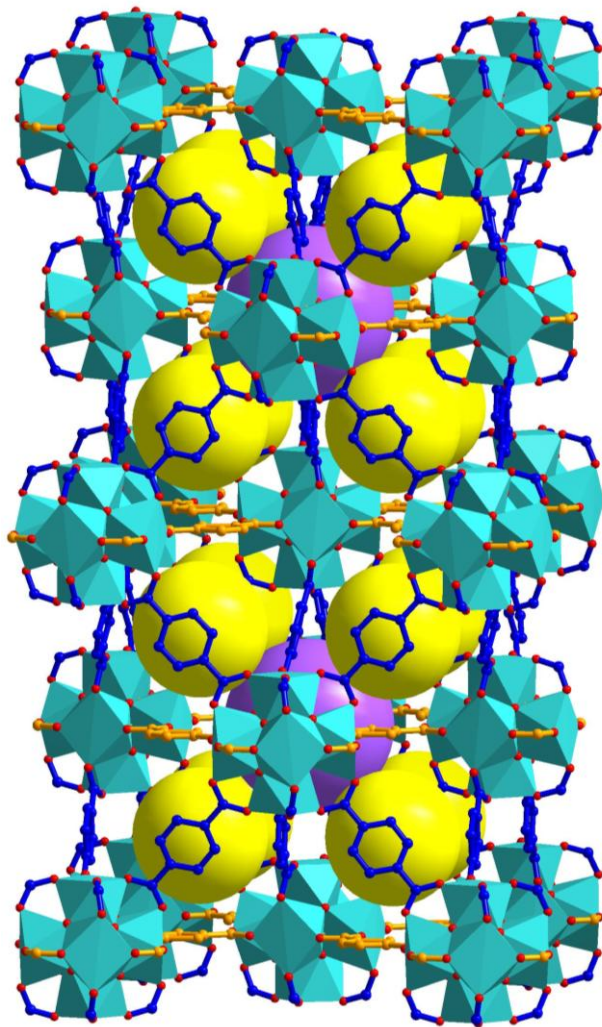
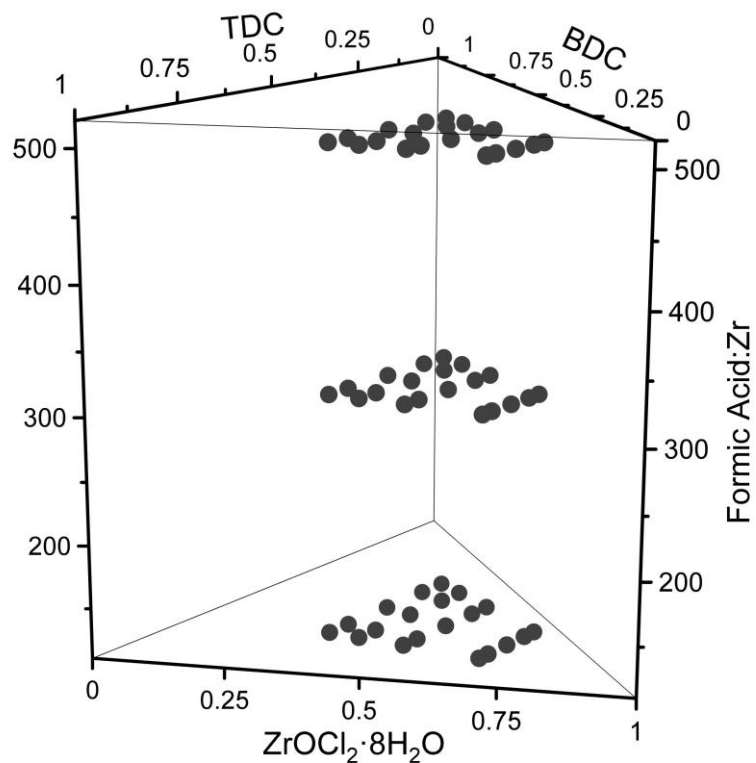
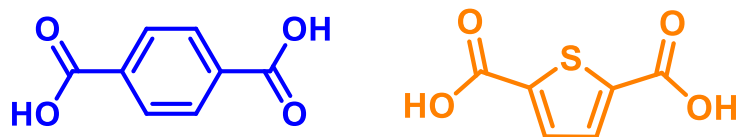
Matt  
Rosseinsky



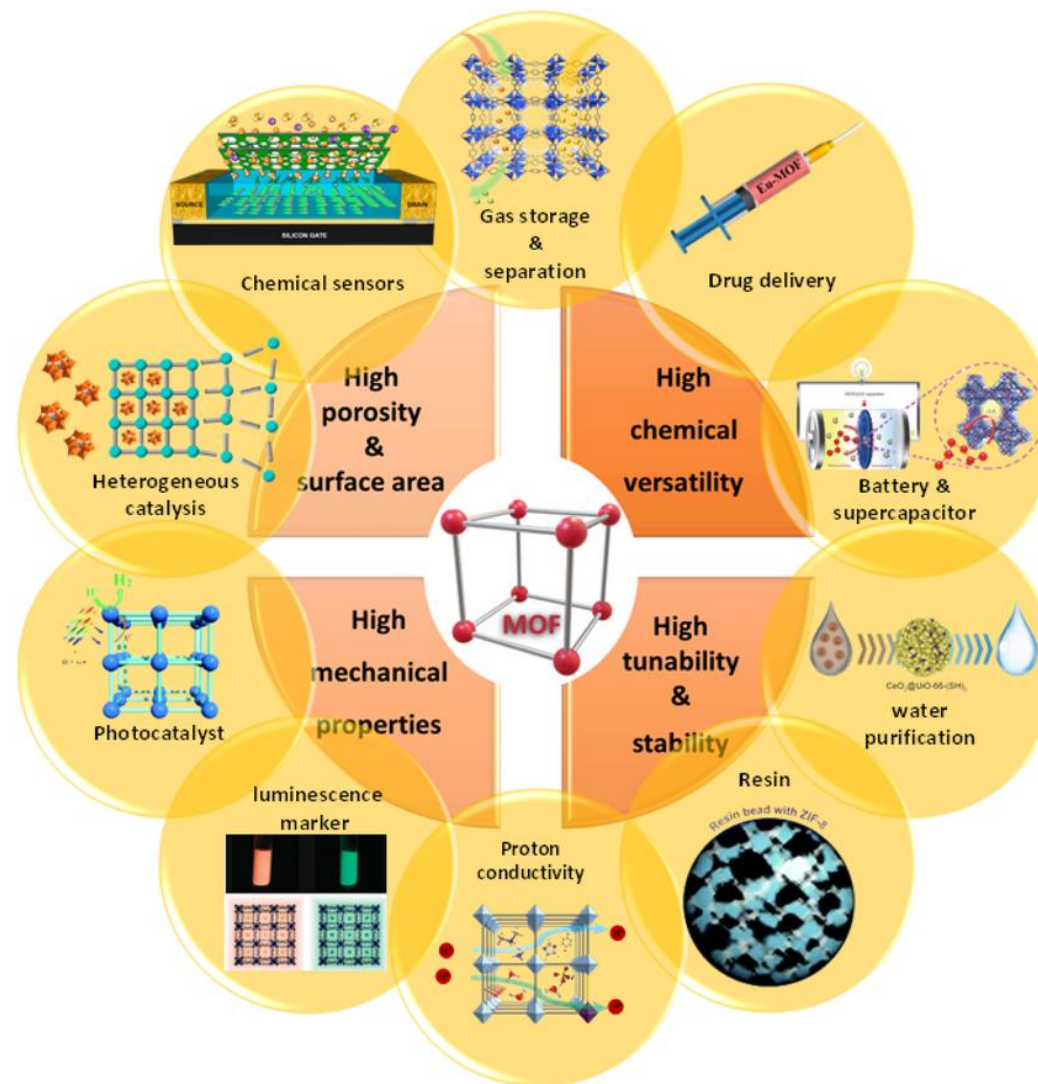
# Ordered straight and bent linkers on Zirconium MOFs



Matt  
Rosseinsky

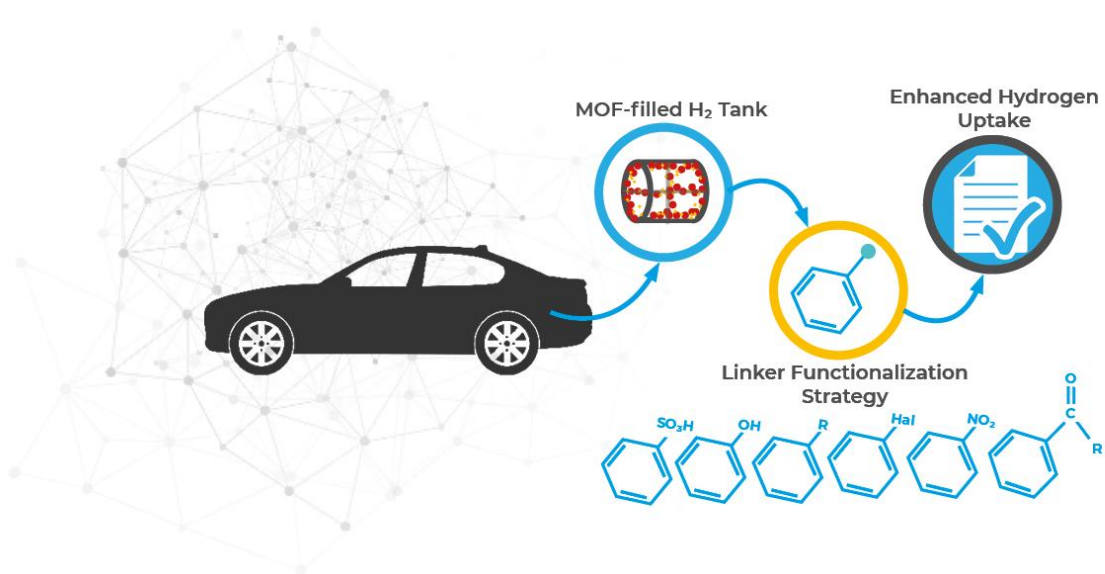


# General applications of MOFs





# Adsorptive applications of MOFs

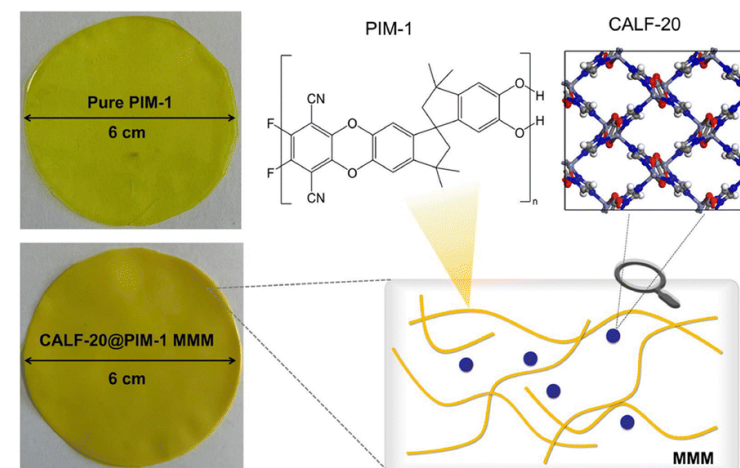
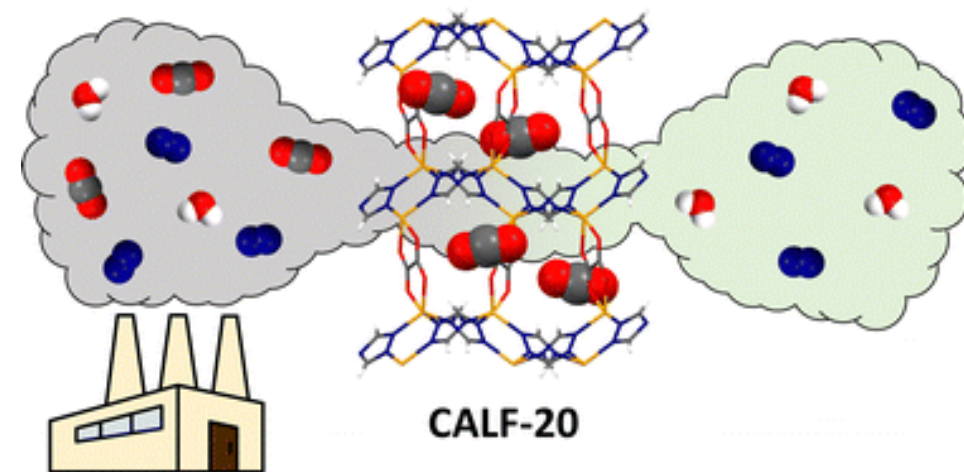
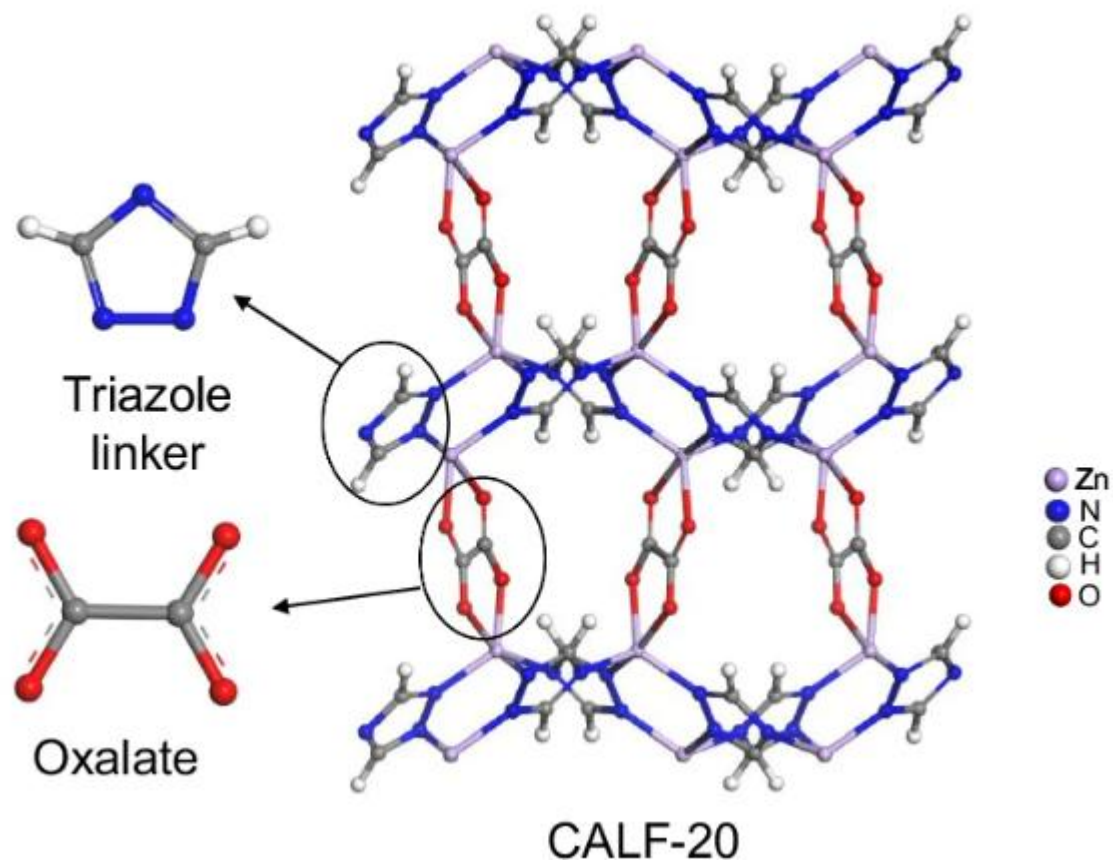


**Vehicular fuel storage**



**Prevention of fruit over-ripening**

# Carbon dioxide capture



# Water harvesting

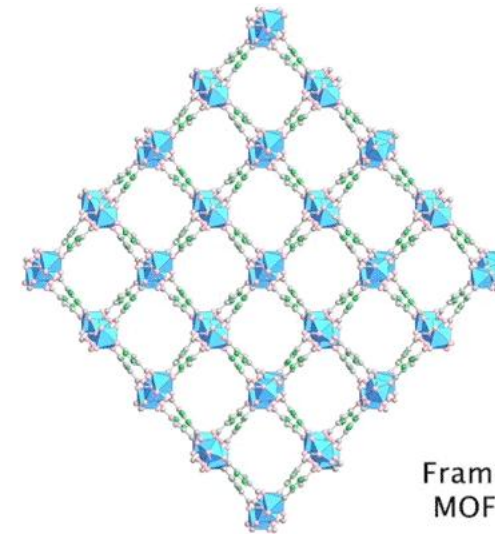
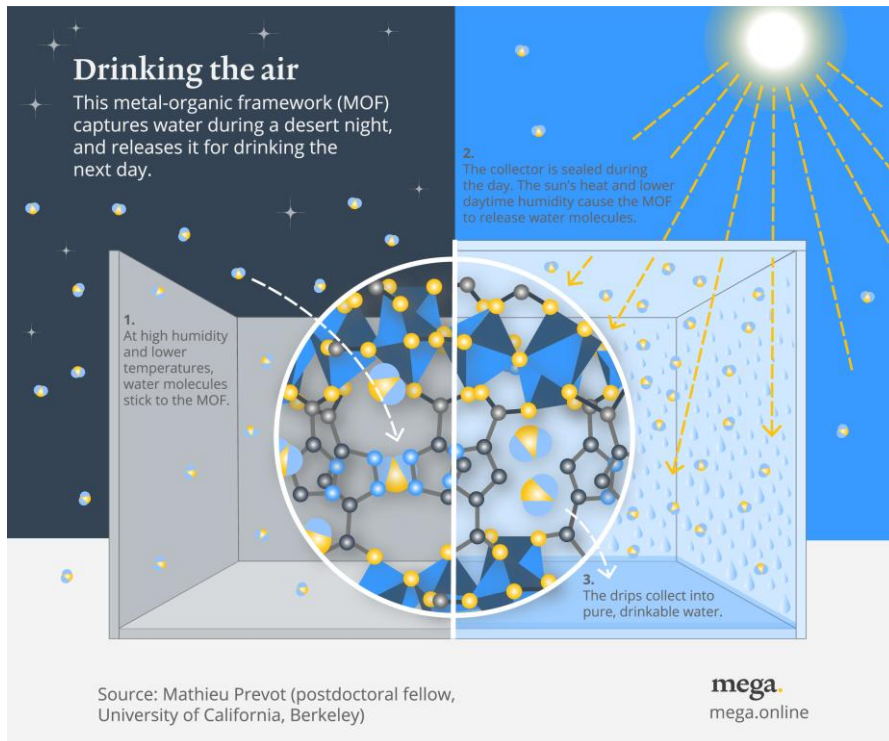
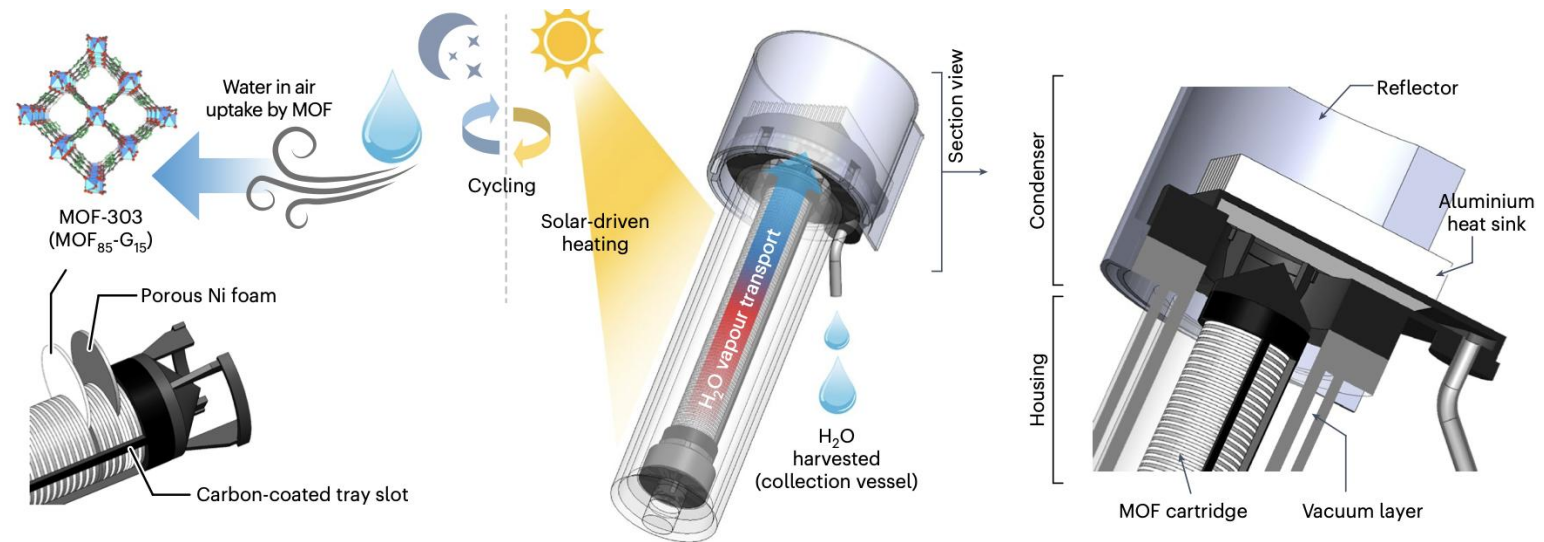
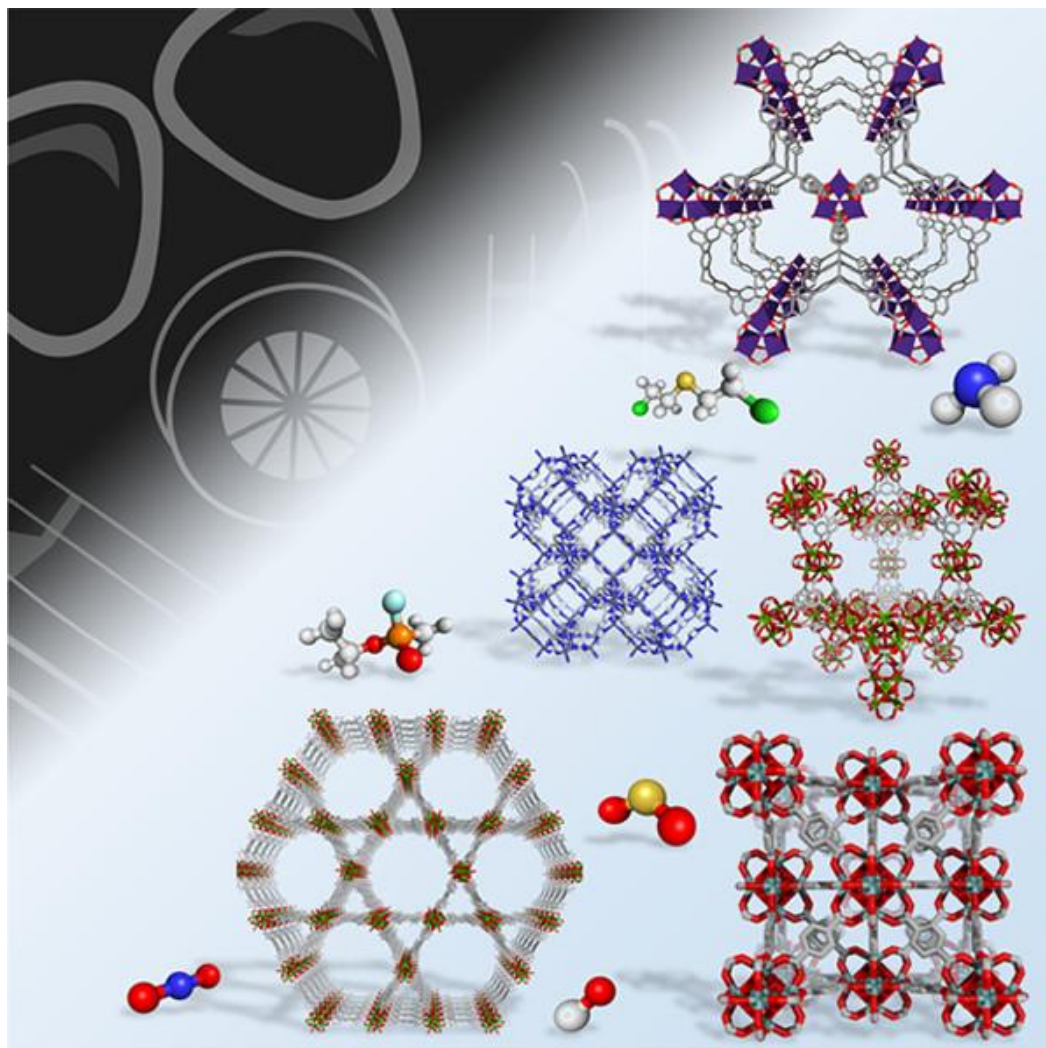


Image source: Lawrence Berkeley National Lab

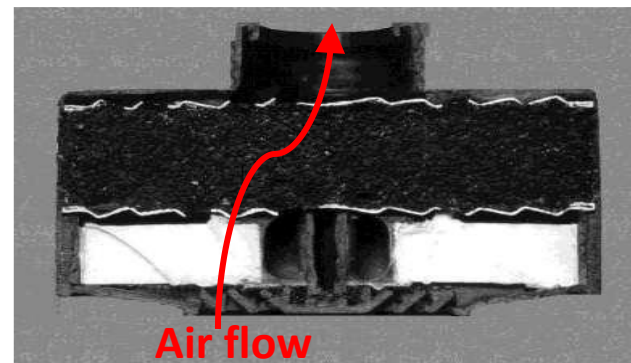




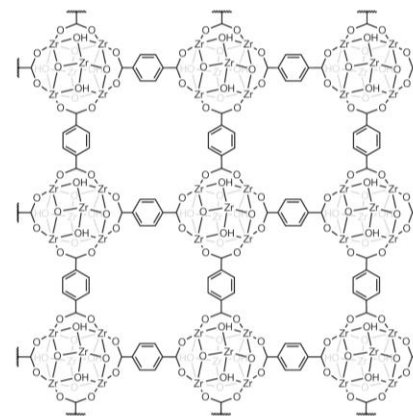
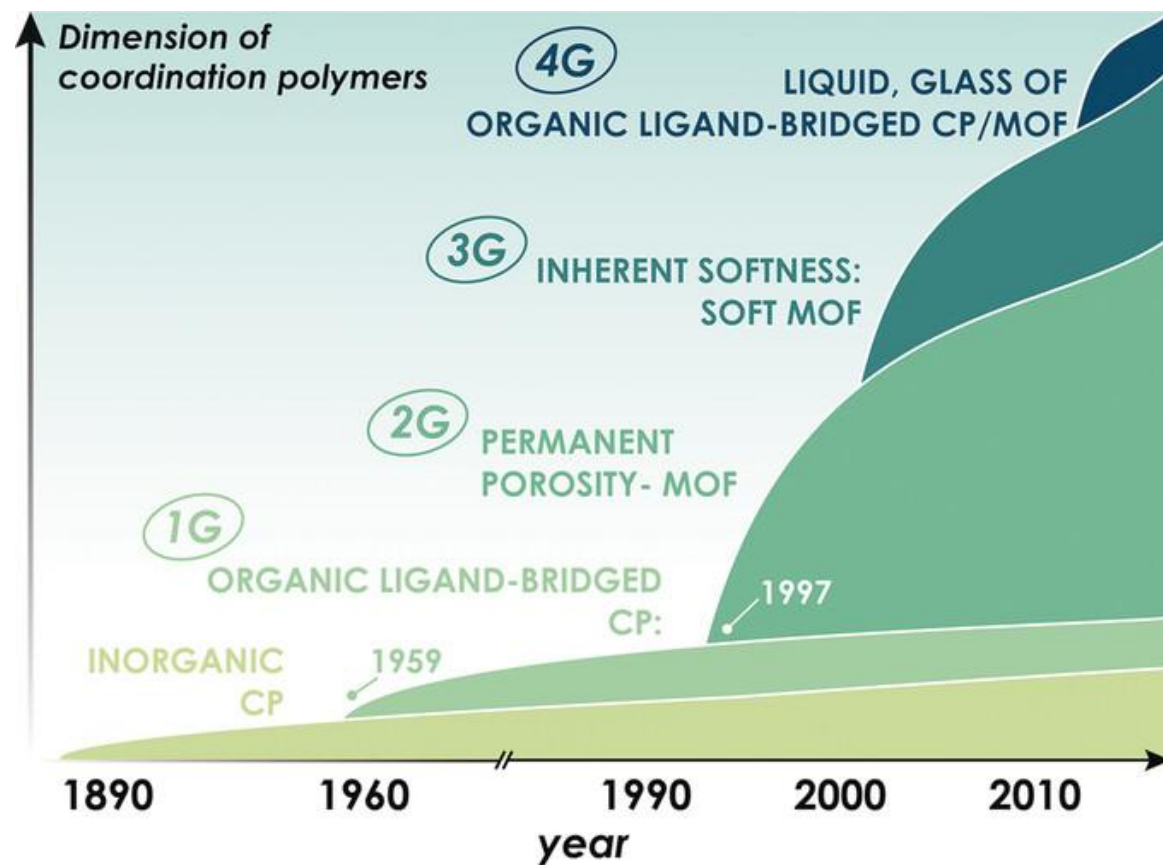
# Toxic gas adsorption



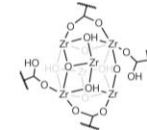
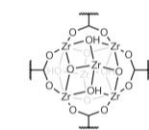
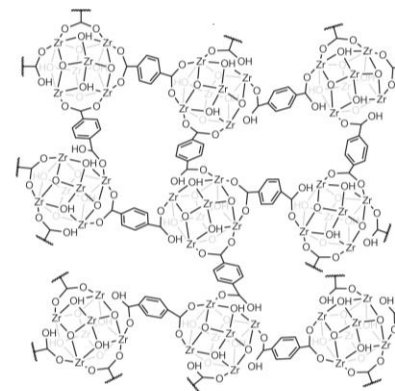
T. Islamoglu et al., *Chem. Rev.*, 2020, **120**, 8130-8160.



# The changing state of MOFs

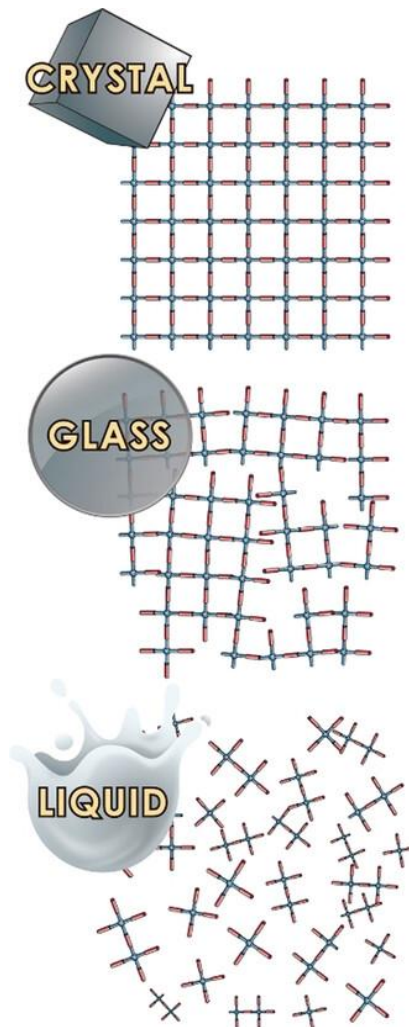


Uniaxial compression or ball milling



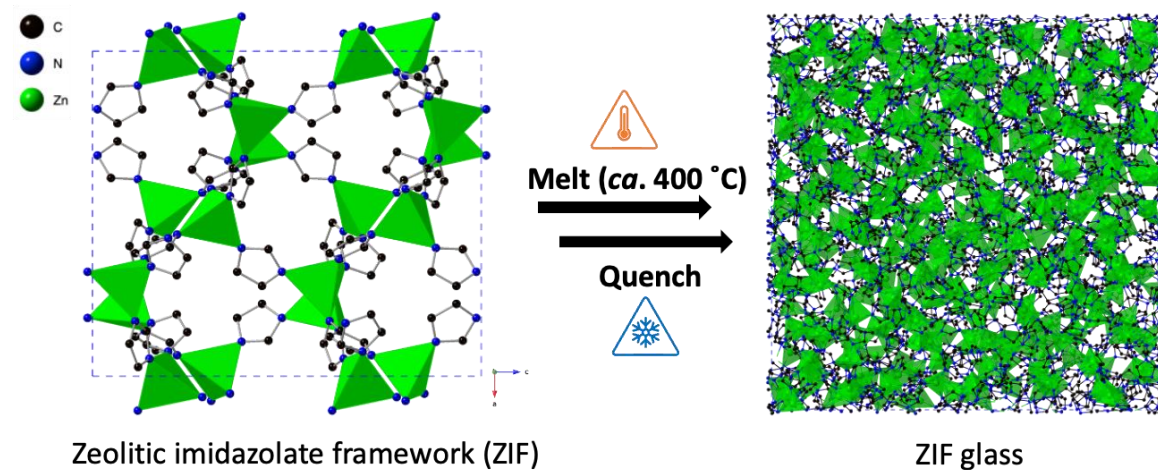
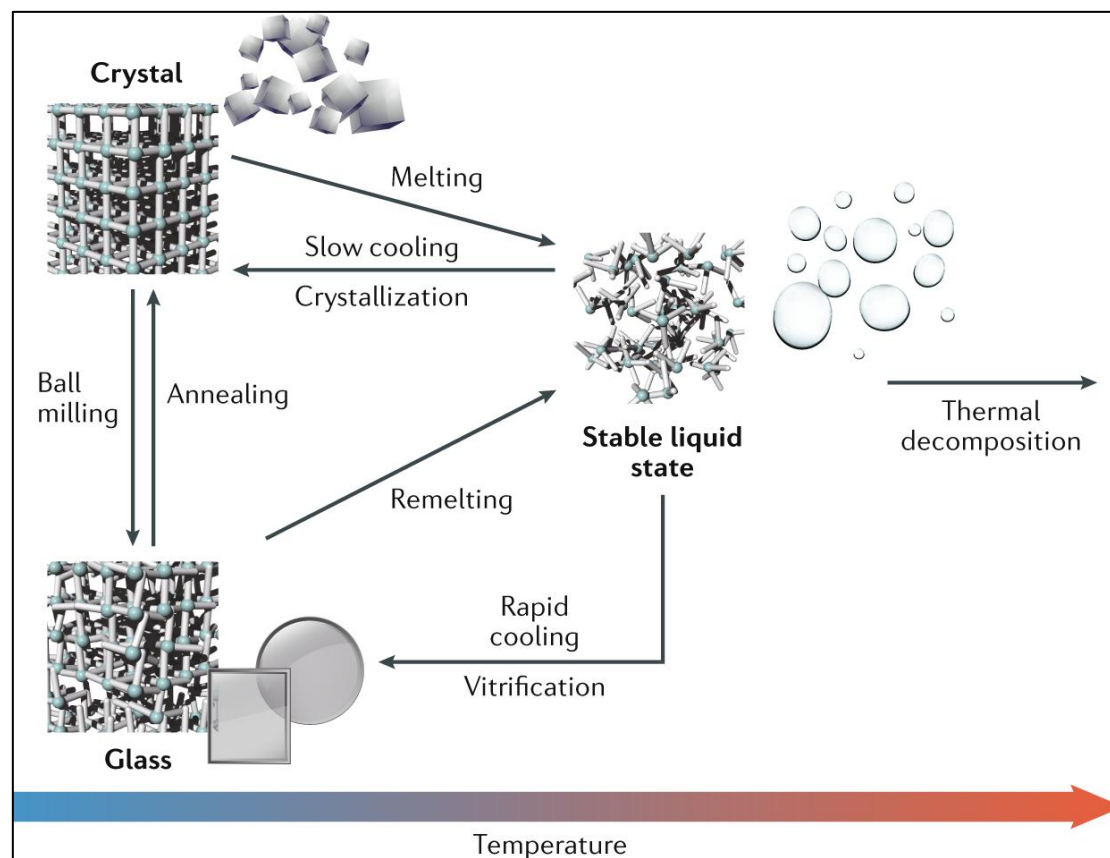
Crystalline

Disordered

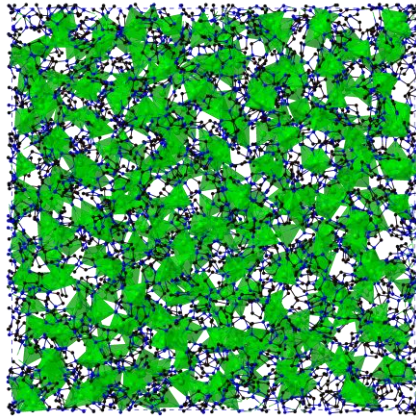




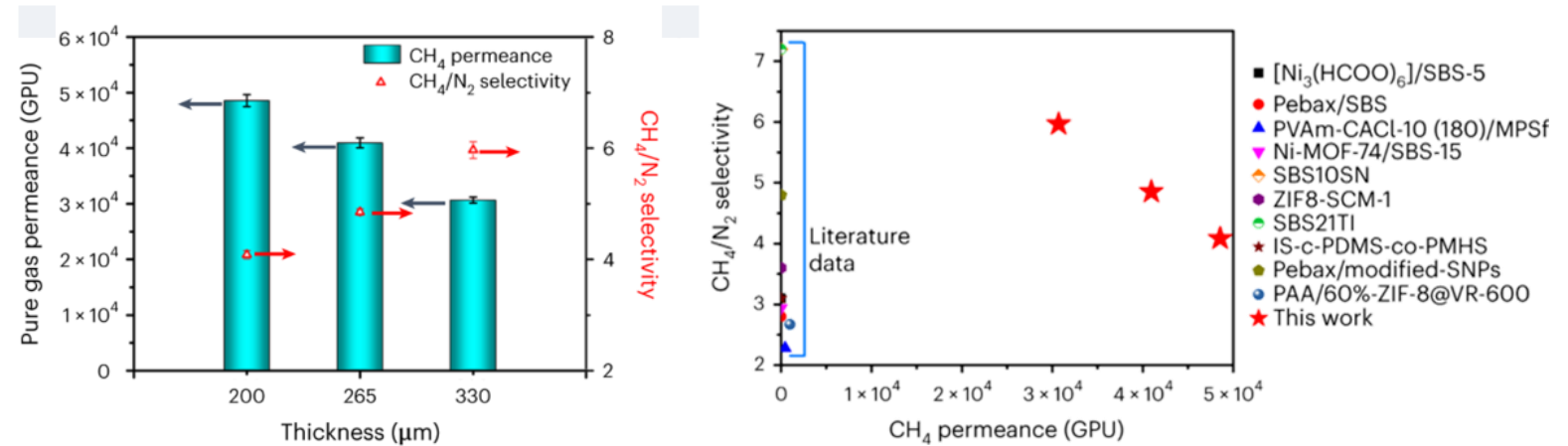
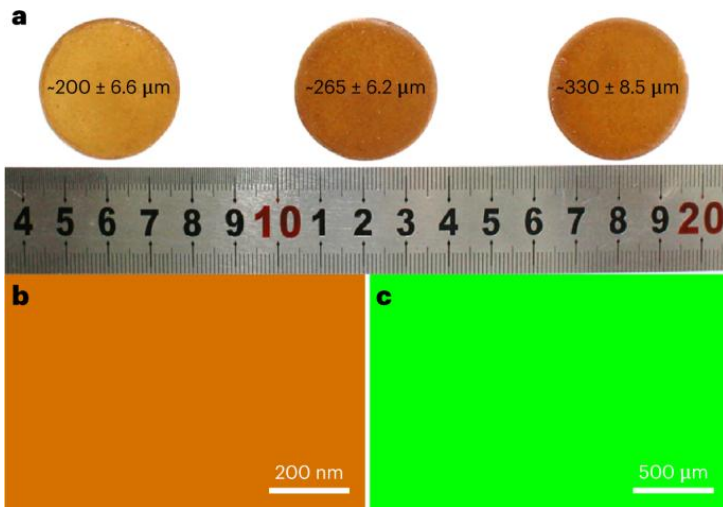
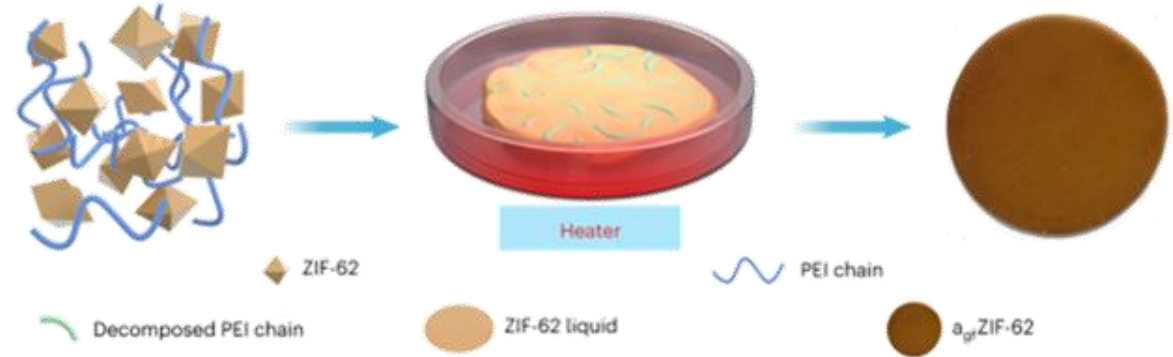
# MOF glasses



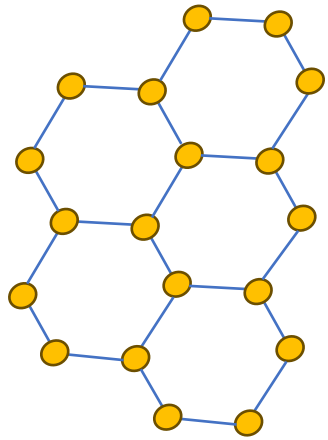
# Membrane materials in chemical separations



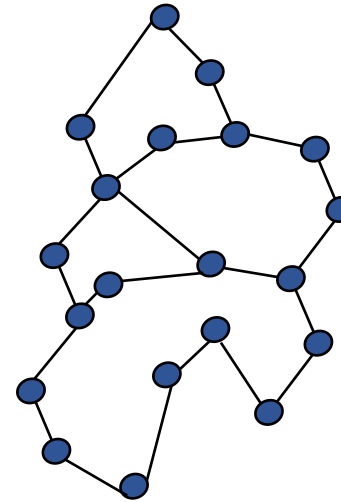
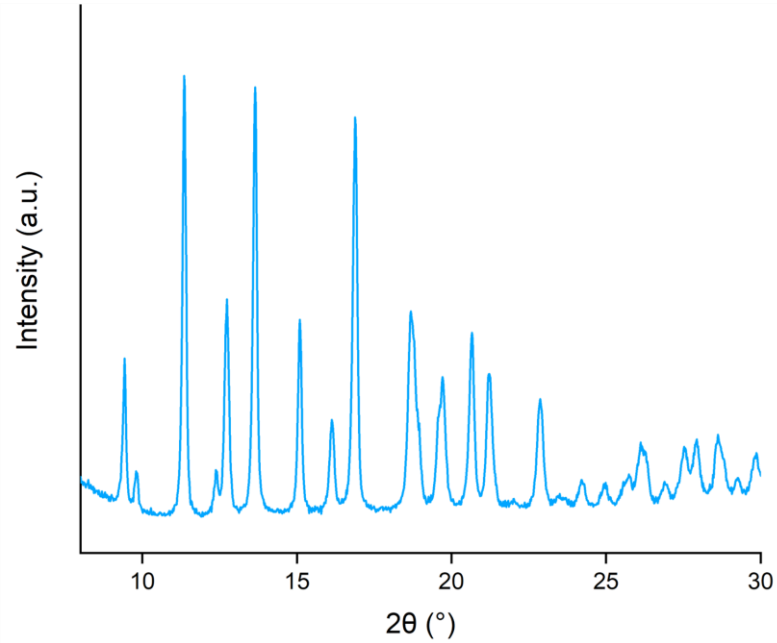
ZIF glass



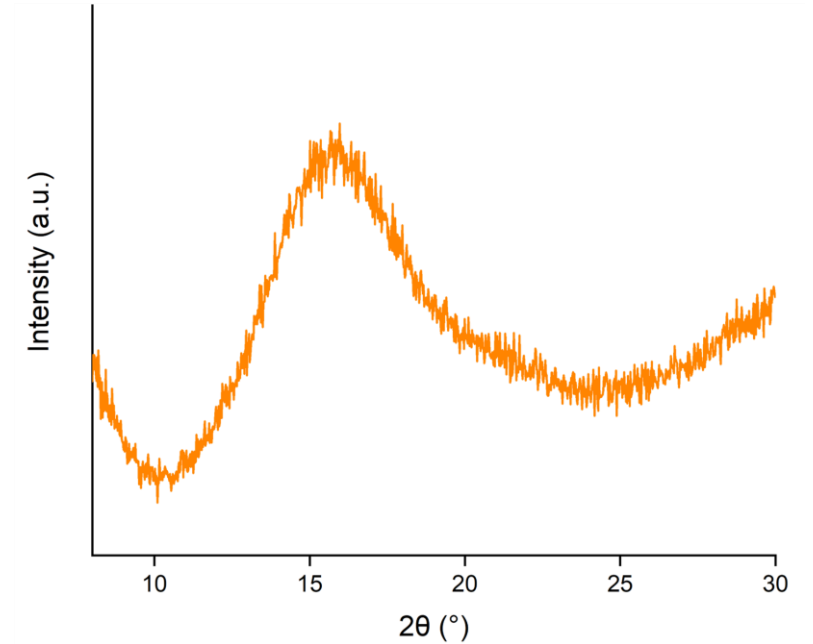
# Crystalline v non-crystalline



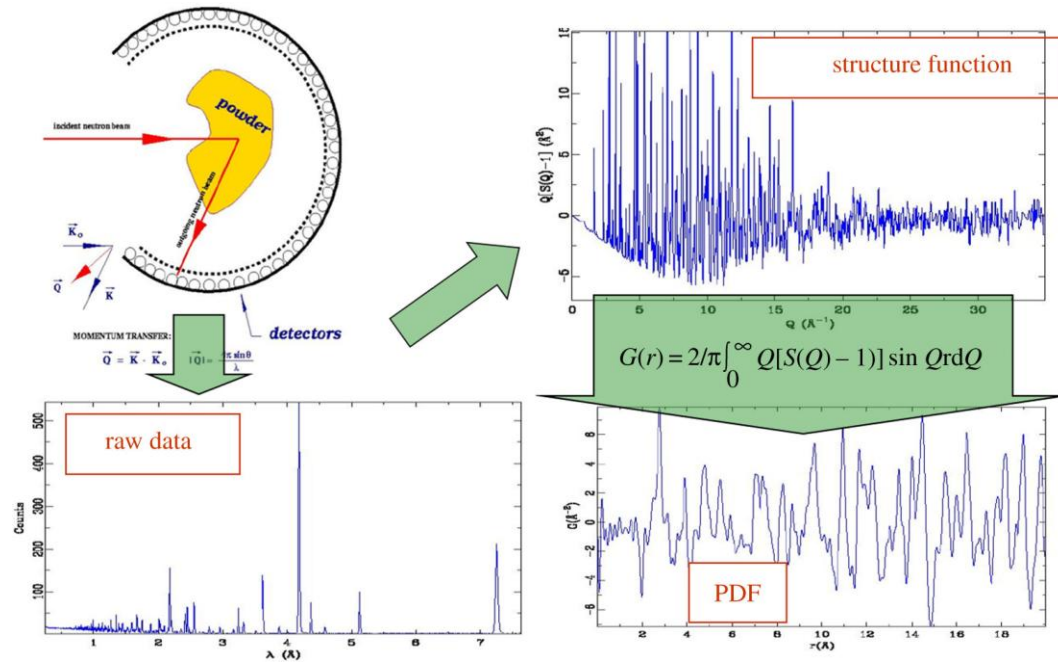
Crystalline



Non-crystalline



# Total scattering and pair distribution function



- **Total scattering** (Bragg + diffuse) experiments are similar to normal diffraction experiments.

- **Diffuse scattering provides information on the short-range structure of materials.**

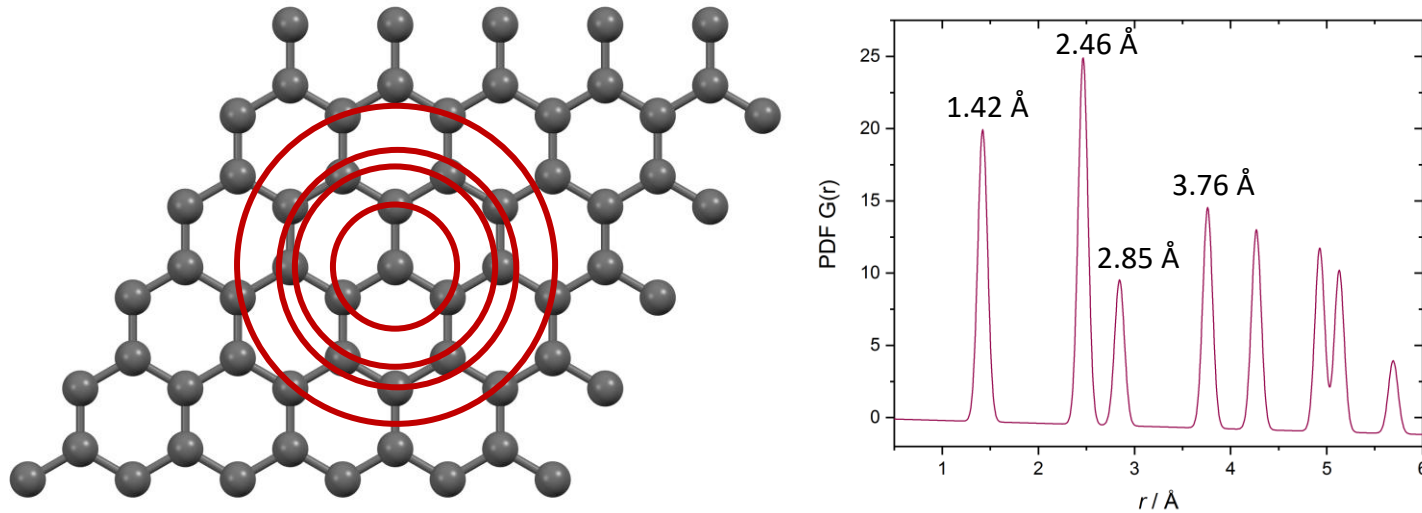
- Total scattering can even be used for amorphous solids, liquids and glasses.

- For disordered crystalline materials, it helps characterise the periodic structure **and** the deviations from long-range order.



# Pair distribution function (PDF)

- Total scattering is mathematically related to the PDF *via* Fourier transform (like the reciprocal and direct lattices)

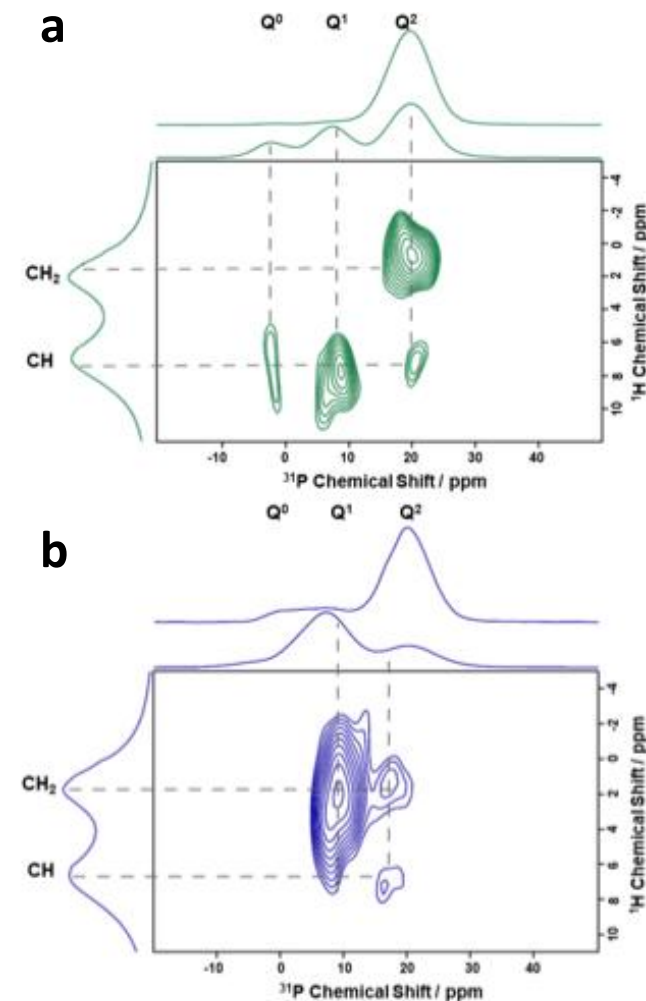
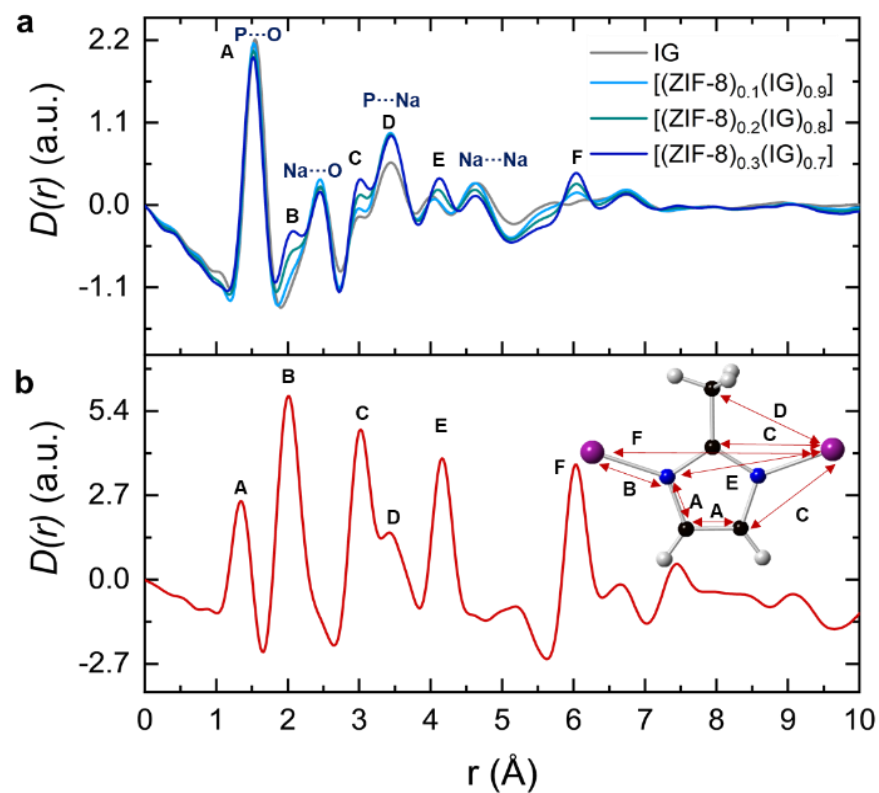
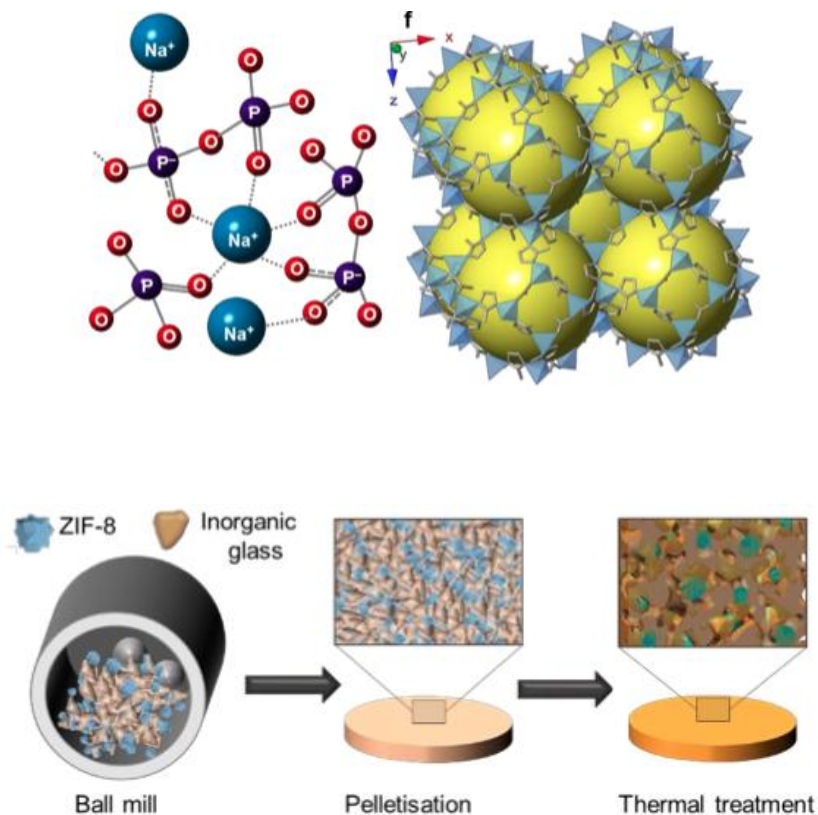


- Peak positions give interatomic separations
- Area provides information about coordination number

# Interfaces in MOF crystal-glass composites



Frédéric Blanc

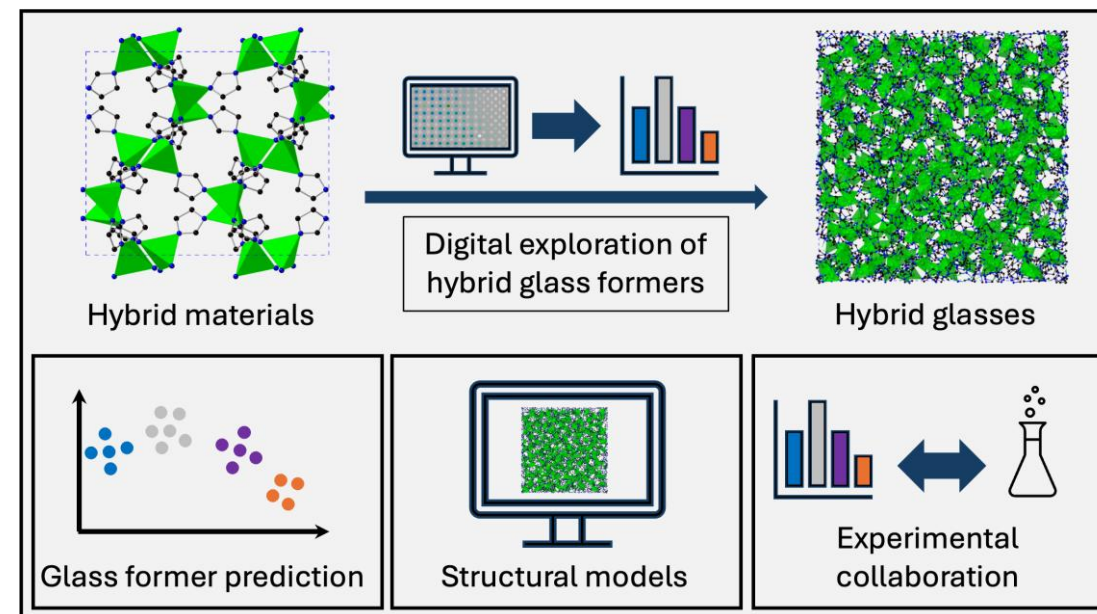
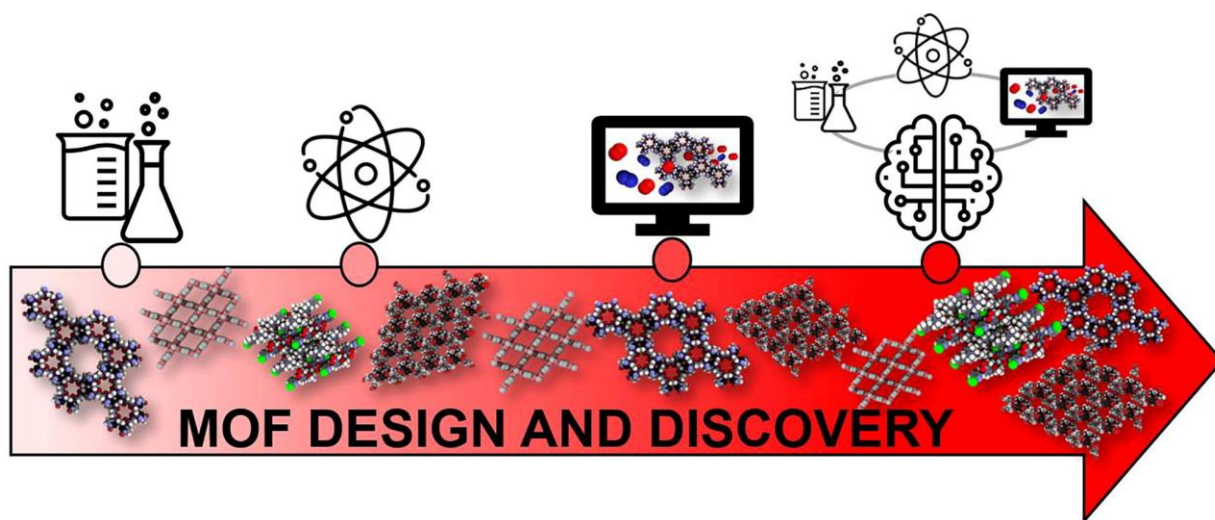


# Summary

- MOFs are framework materials made from metal ions or clusters linked by organic ligands, forming porous 3D networks with exceptionally high surface areas.
- Their structure, pore size, and functionality can be precisely tuned by selecting different metals and linkers, allowing for tailored chemical and physical properties.
- MOFs have wide-ranging uses, including gas storage, separation, catalysis, drug delivery, and sensing – driven by their tuneable porosity and reactivity.
- ✓ Advantages: large surface areas, tuneability, and versatility.
- ✗ Challenges: stability (especially in moisture/heat), scalability, and cost of synthesis.

# Future perspectives: digital discovery

- Digital materials chemistry goes hand in hand with experimental materials chemistry
- AI and machine learning is driving MOF commercialisation with interest from several start-ups and large companies





# Thanks for listening!

## Questions?

