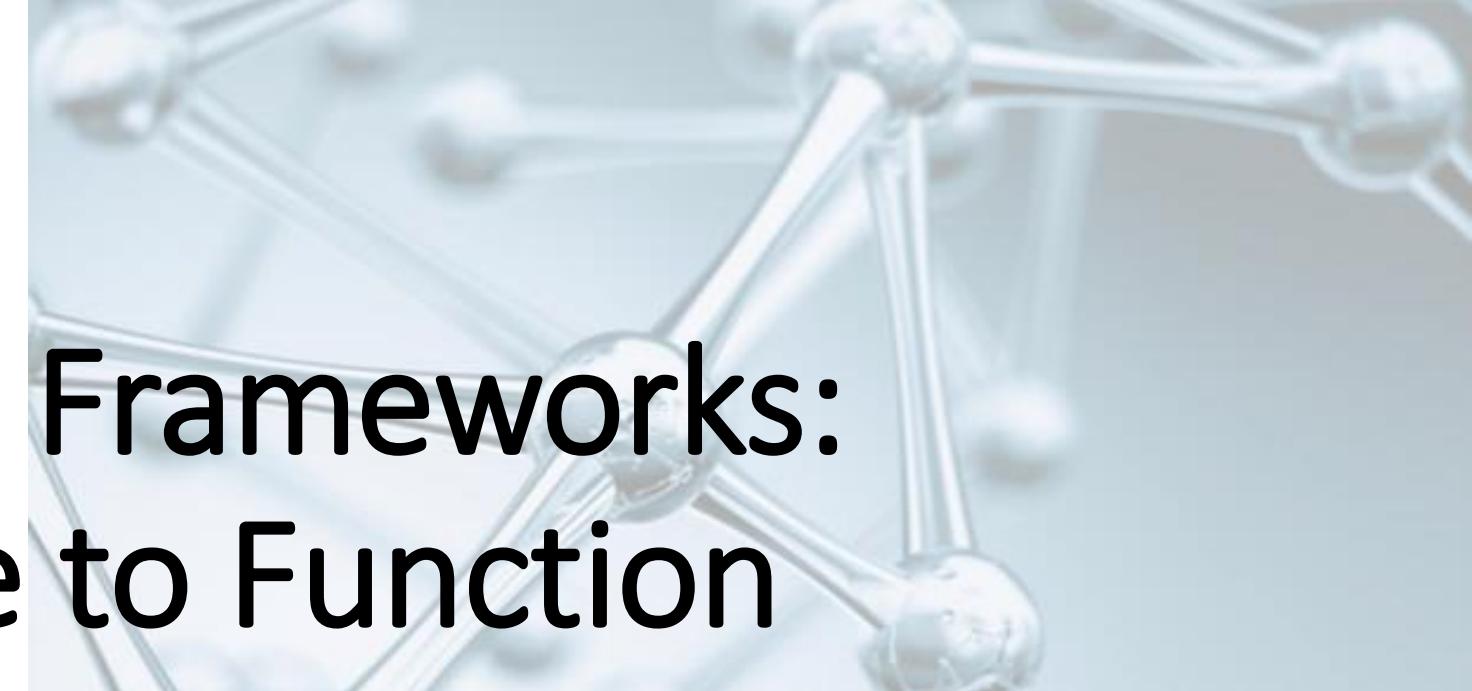


Metal–Organic Frameworks: From Structure to Function



Lauren McHugh

Lecturer in Materials Chemistry

University of Liverpool

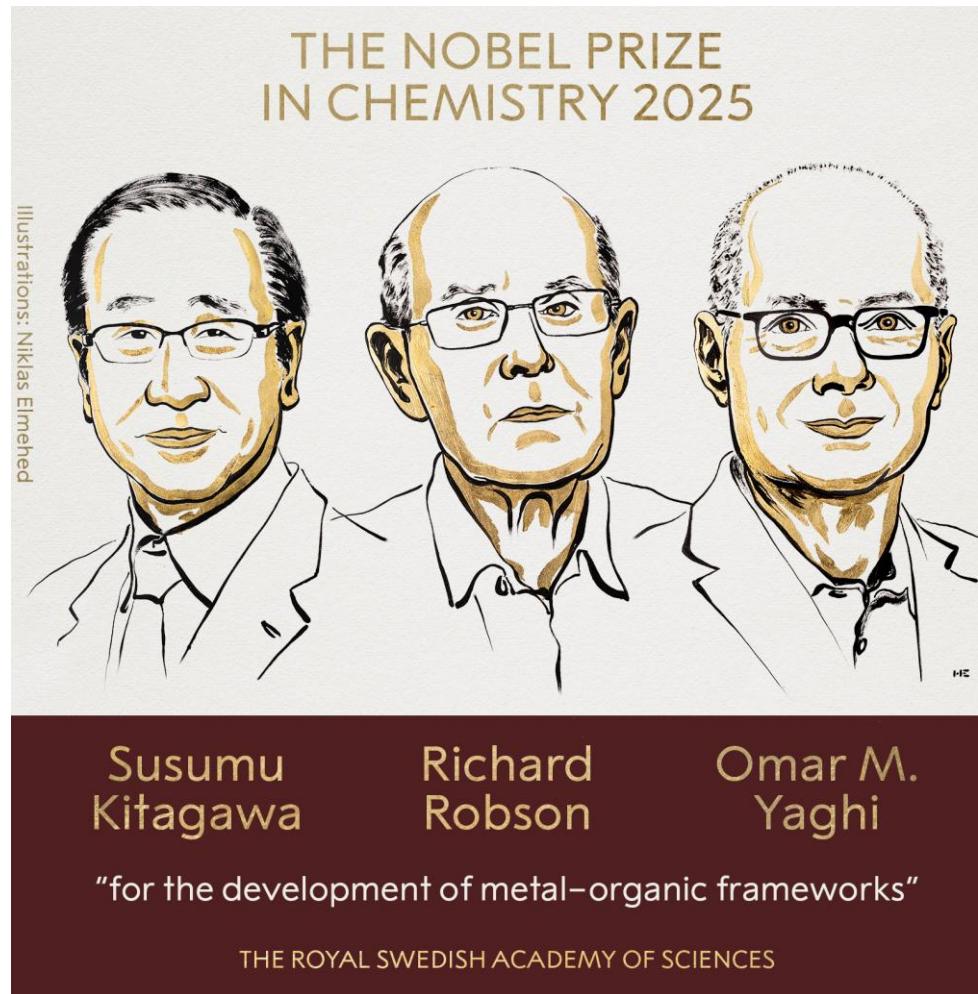
L.N.Mchugh@liverpool.ac.uk

November 14th 2025



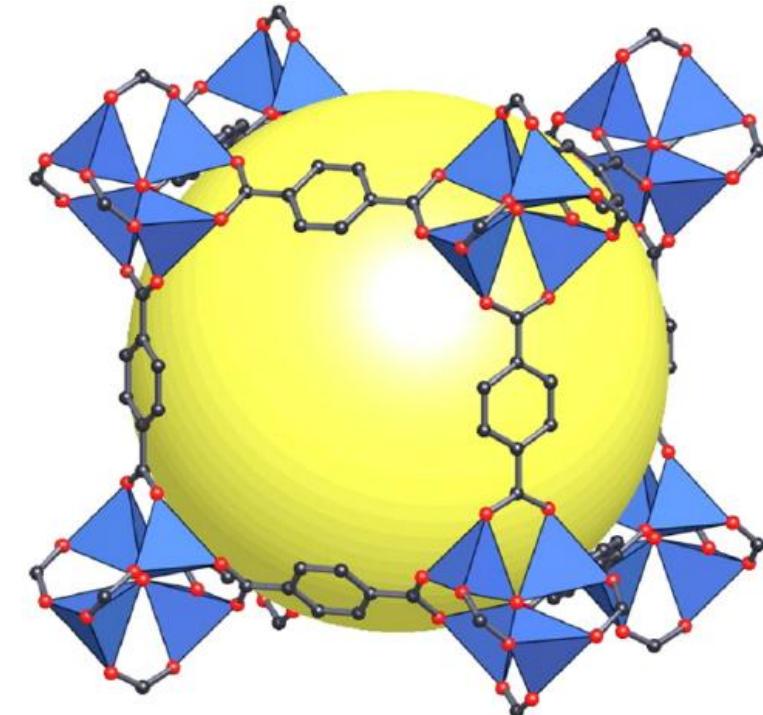
UNIVERSITY OF
LIVERPOOL

MOFs won the Nobel Prize in Chemistry!



Metal–organic frameworks (MOFs)

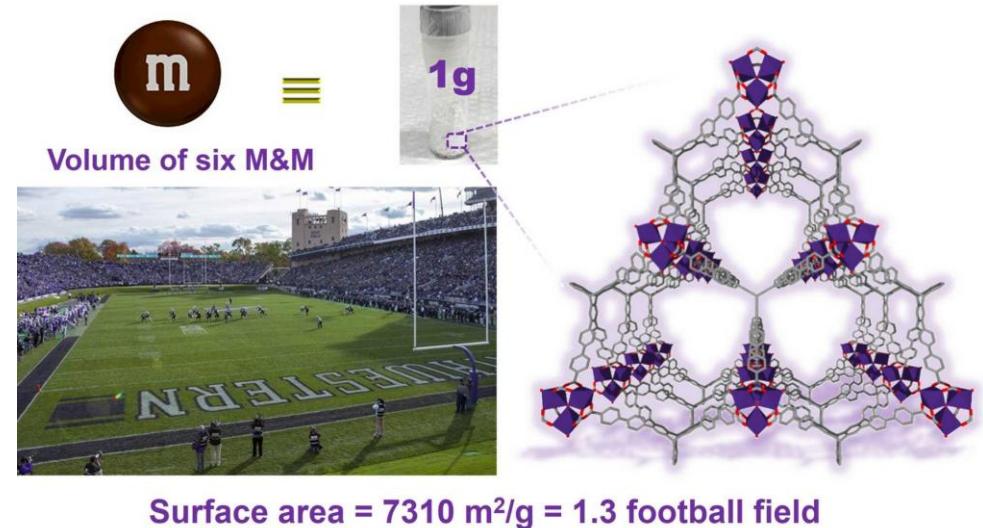
- Metal–organic frameworks (MOFs) are network solids which consist of metal ions or clusters connected to organic linkers
- Scope to modulate their properties via:
 - Metal centres
 - Linker
 - Defect formation
- Their cage-like structure typically leads to a void space within the framework known as a ‘pore’
- This porosity leads to high surface areas



MOF-5 (IRMOF-1)
 $\text{Zn}_4\text{O}(\text{BDC})_3$
Zn centres with
1,4-benzenedicarboxylate (BDC) linkers

Metal–organic frameworks (MOFs)

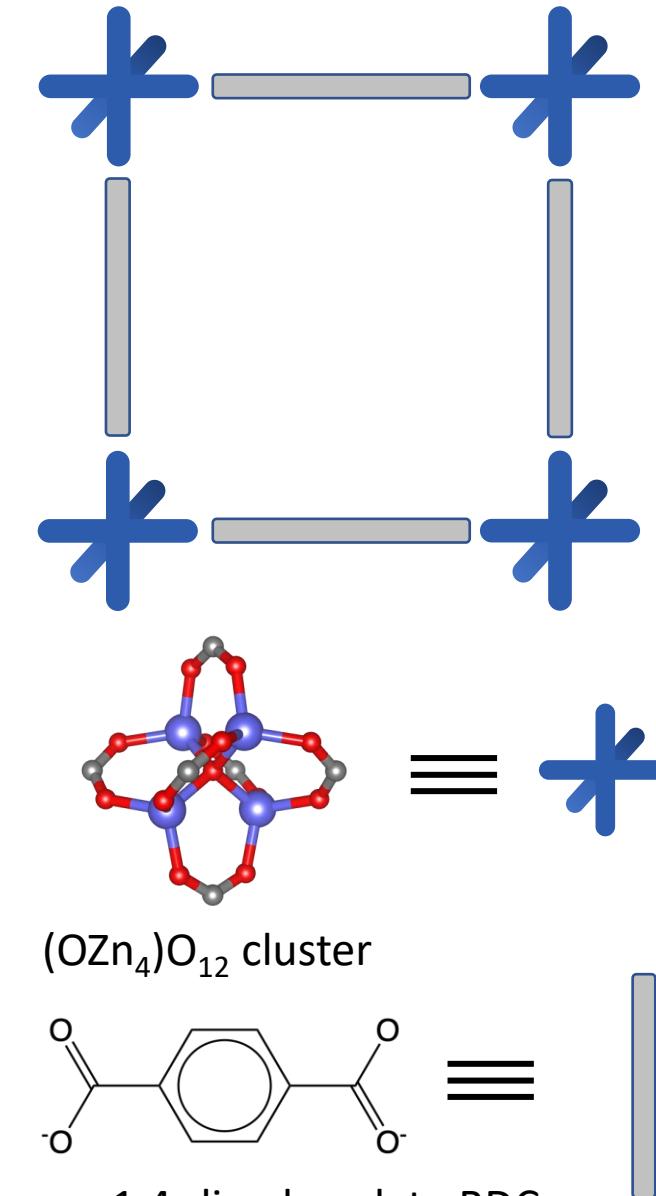
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1g of MOF can have a greater surface area than a football field!

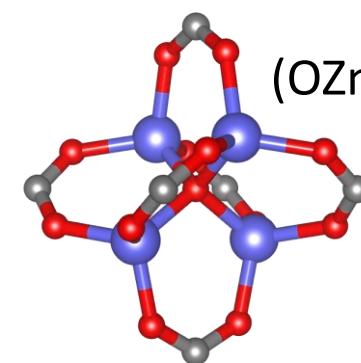
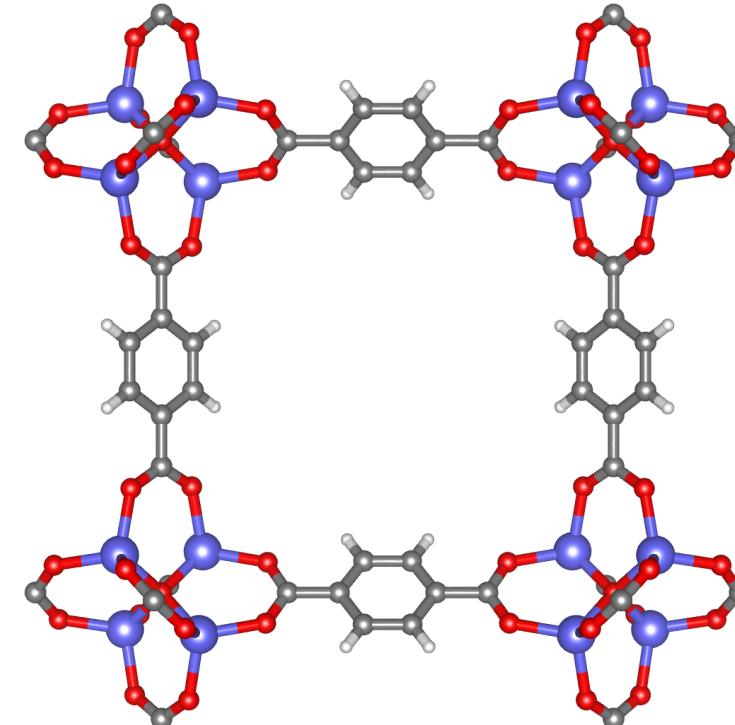
Metal–organic frameworks (MOFs)

- Framework is formed by polydentate ligands (organic linkers) coordinating to metal centres
- Carboxylate linkers are most common, but various functionalities, such as imidazolates are possible
- SBUs – small, well-defined clusters of metal ions that act as repeating nodes in the structure

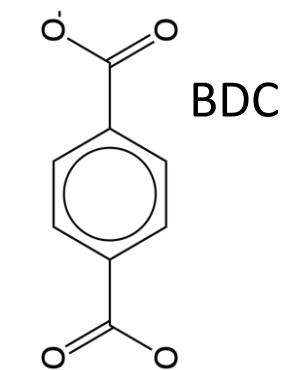


Metal–organic frameworks (MOFs)

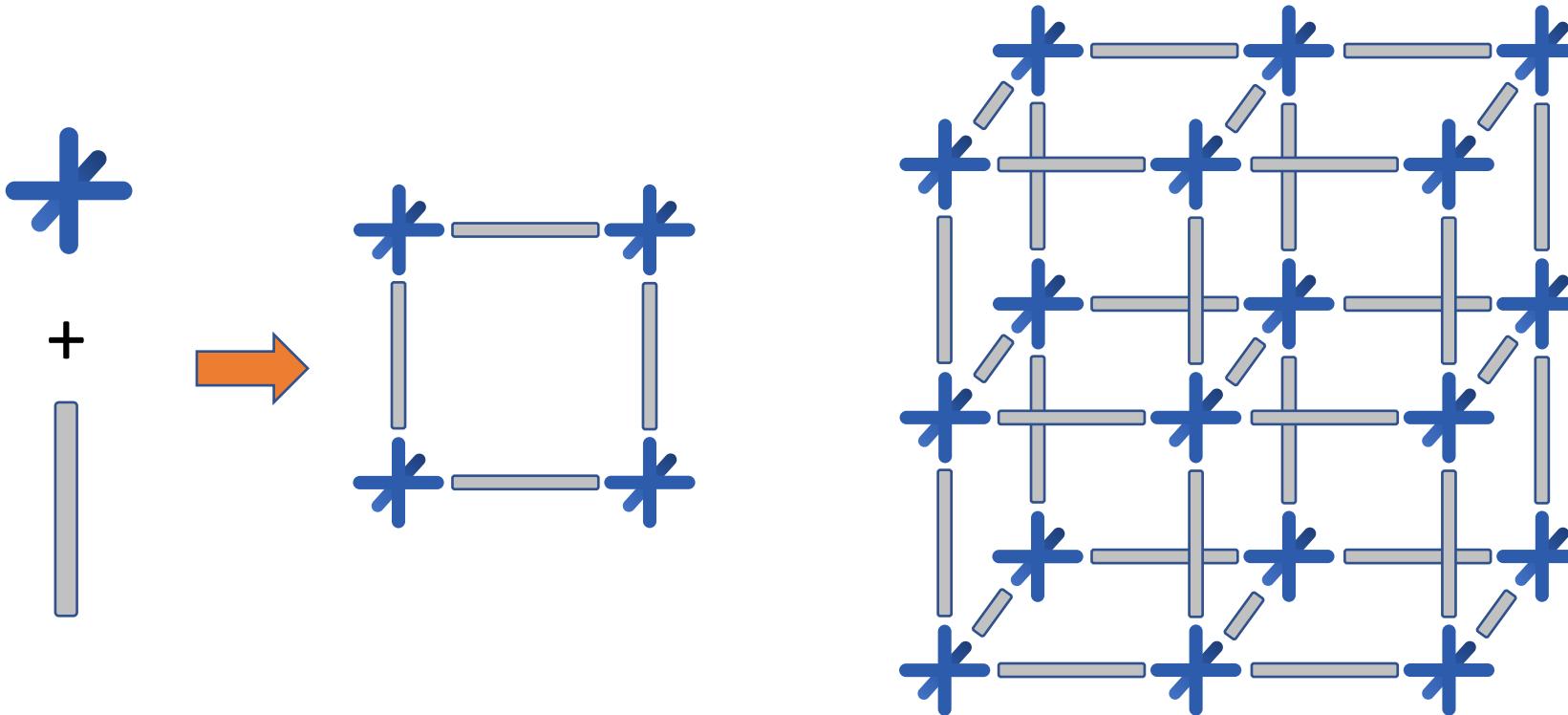
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(OZn₄)O₁₂ cluster
SBUs

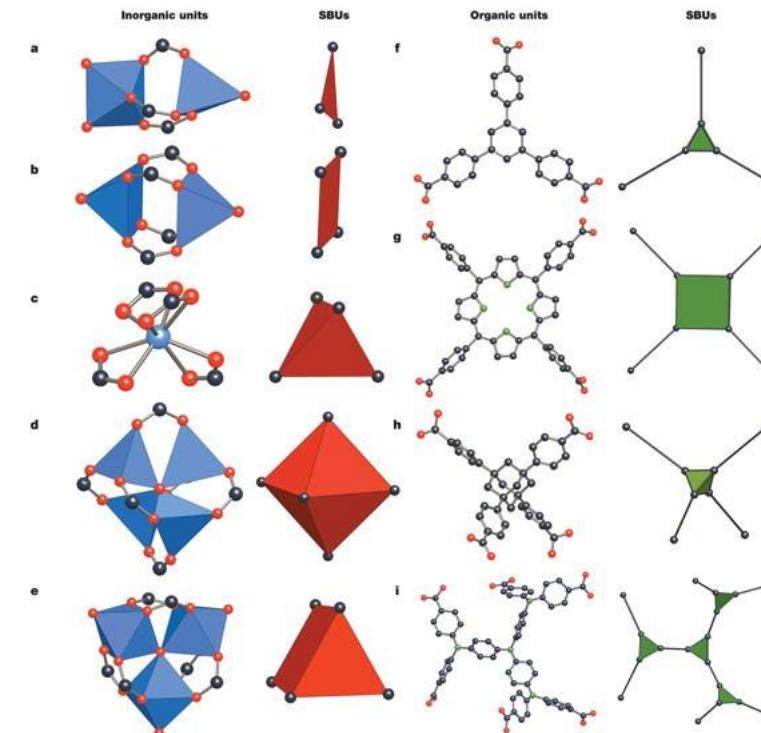
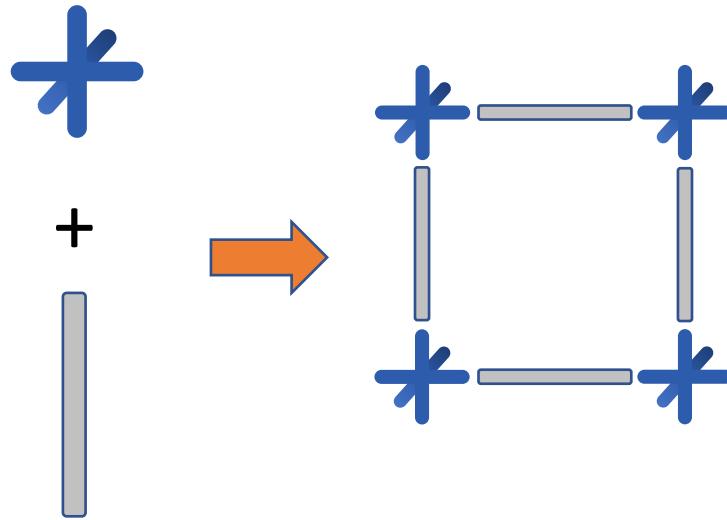


Reticular chemistry

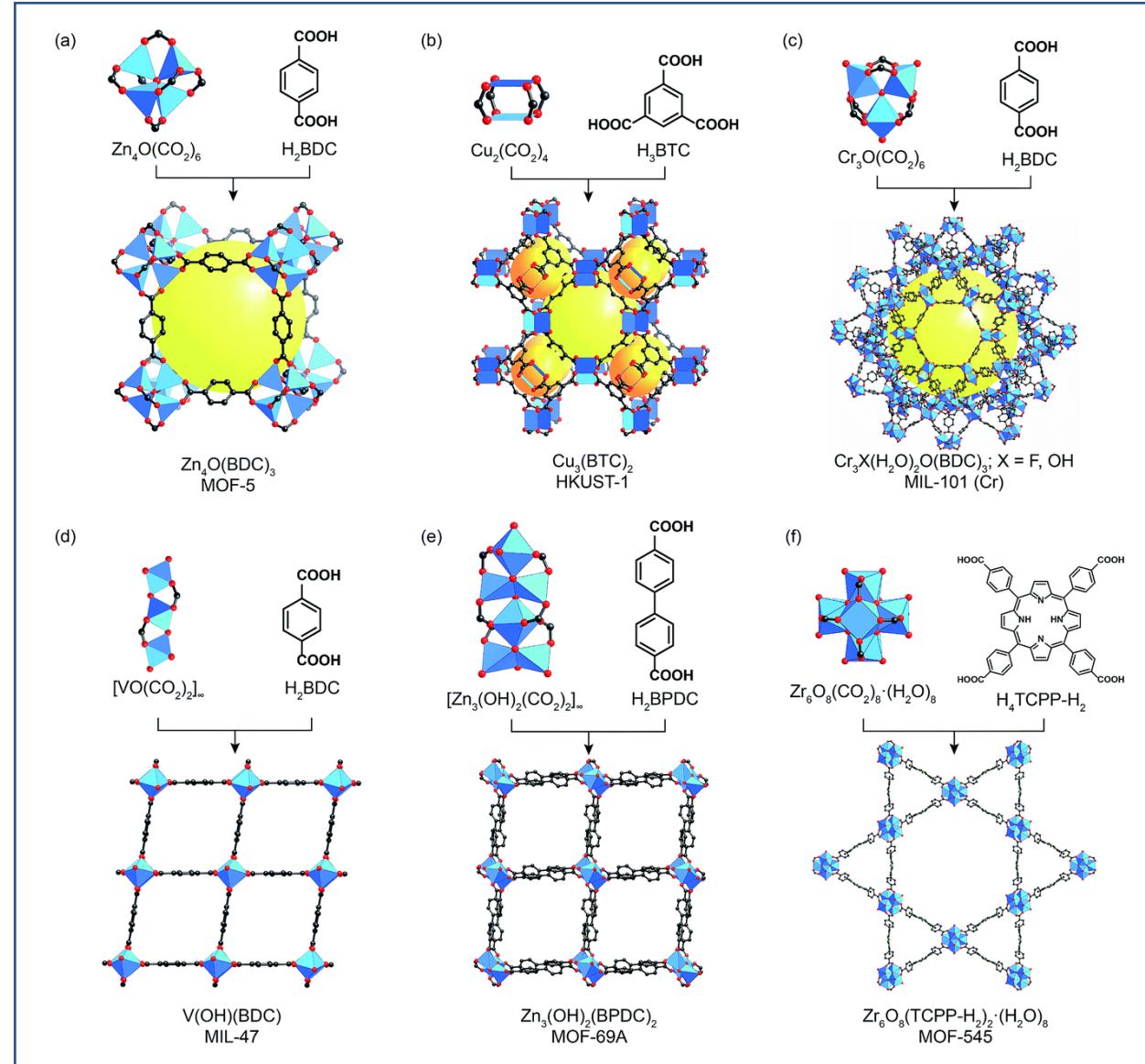


- Chemical components form nodes and linkers in a network
- The geometrical diversity of metal ion and clusters combines with synthetic diversity of organic linkers
- Enables 'controlled' assembly of extended architectures

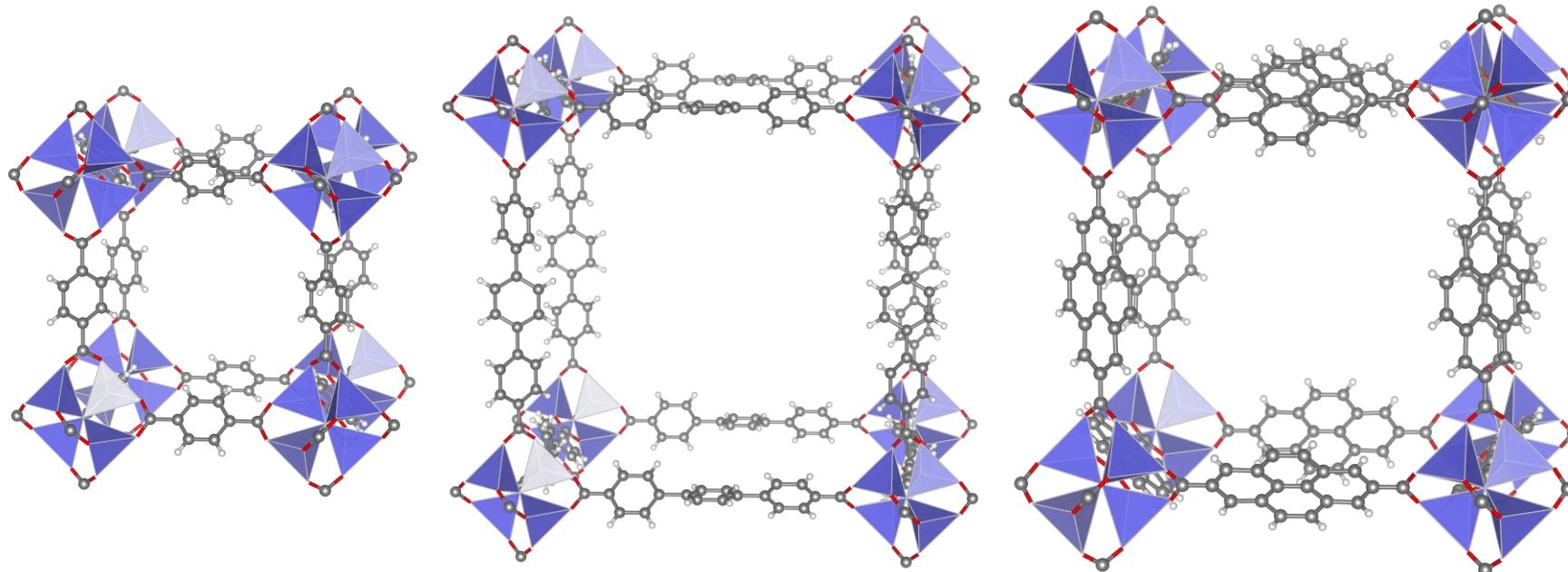
Reticular chemistry



- Chemical components form nodes and linkers in a network
- The geometrical diversity of metal ion and clusters combines with synthetic diversity of organic linkers
- Enables 'controlled' assembly of extended architectures



Crystal engineering: Isoreticular MOFs

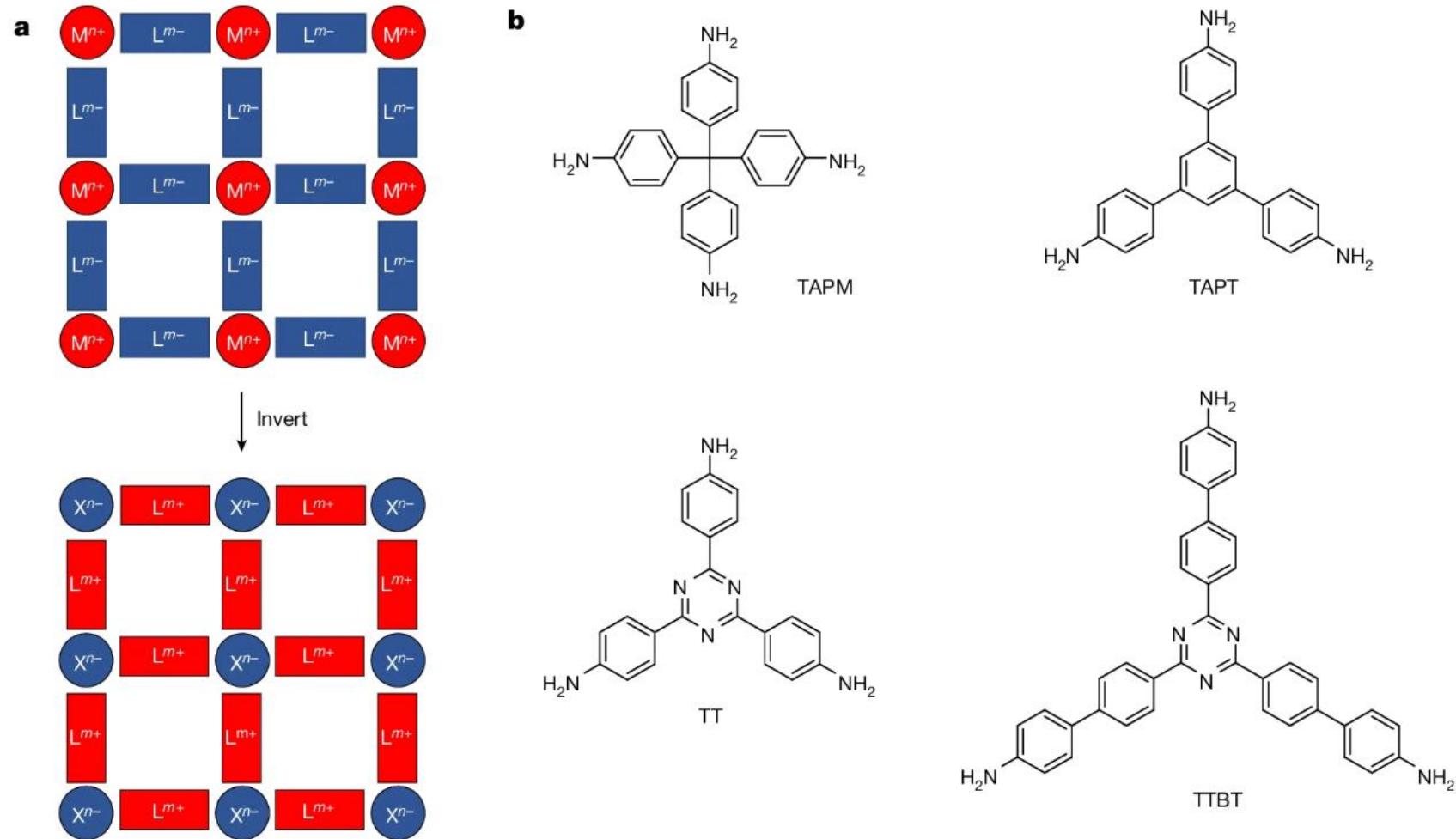


- For some families of MOFs it is possible to change the organic linker and maintain the same framework type
- Relatively strong metal-linker interactions are conserved, but the linker can be swapped
- Modulates the pore size and chemical environment of guests

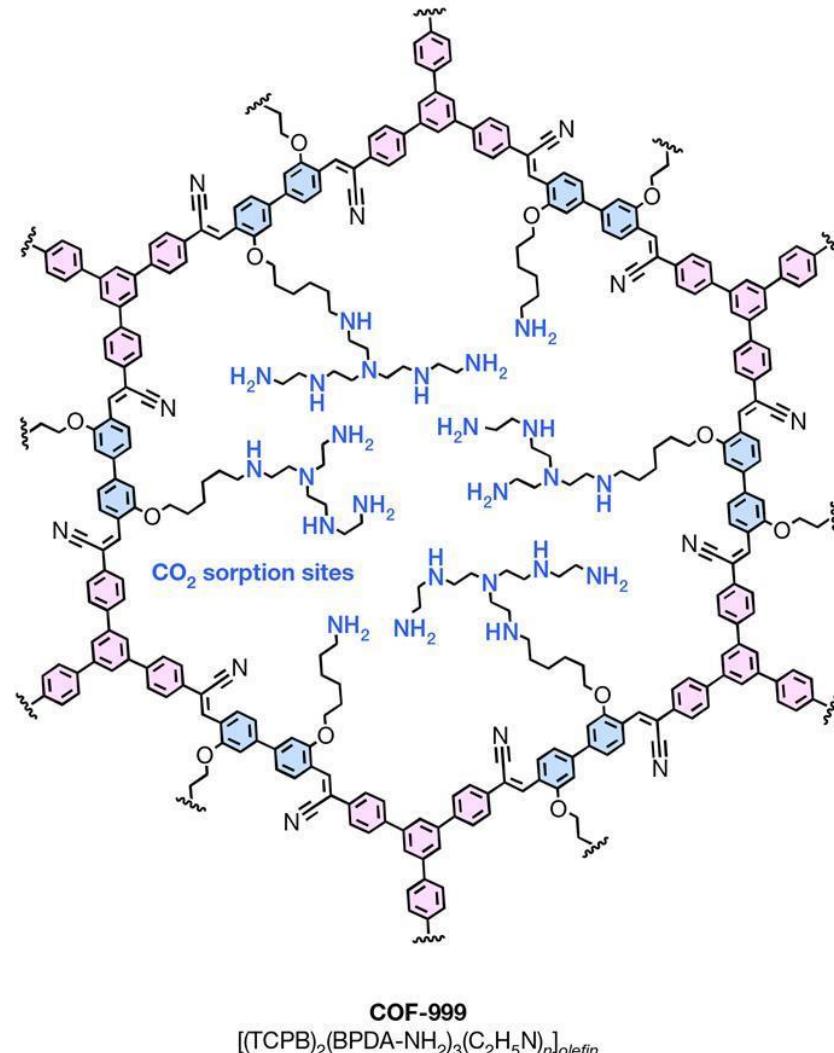
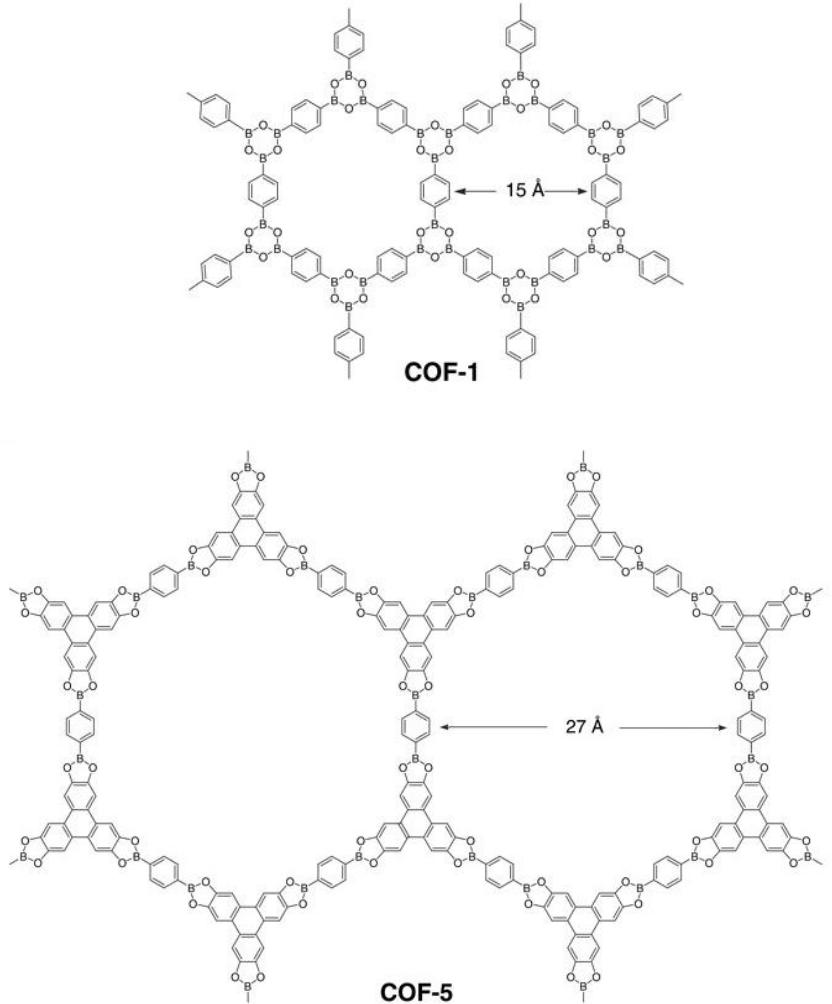
Porous isoreticular non-metal organic frameworks



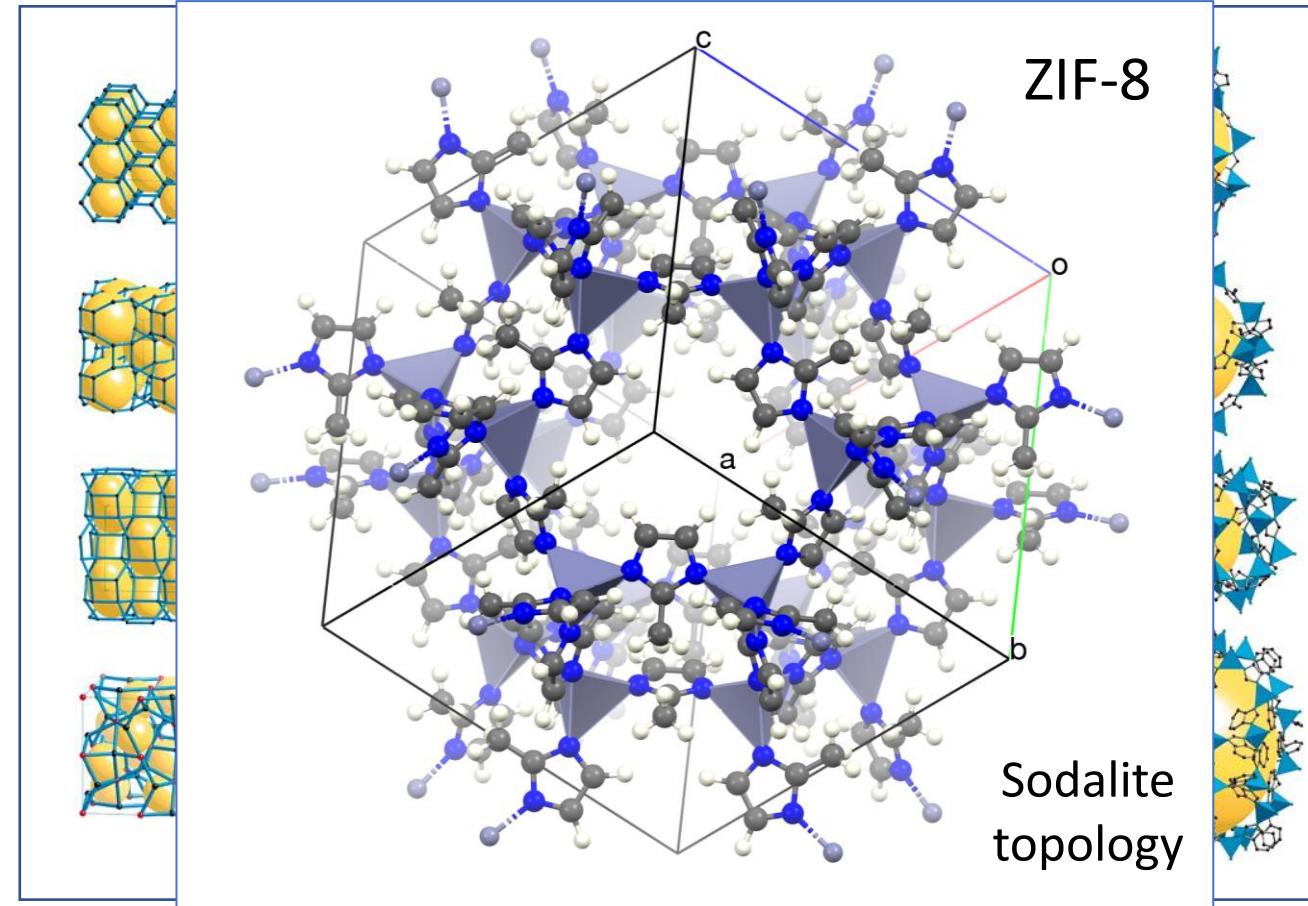
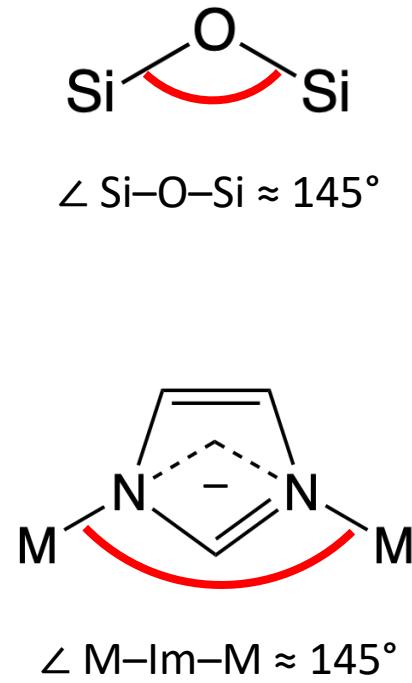
Andy Cooper



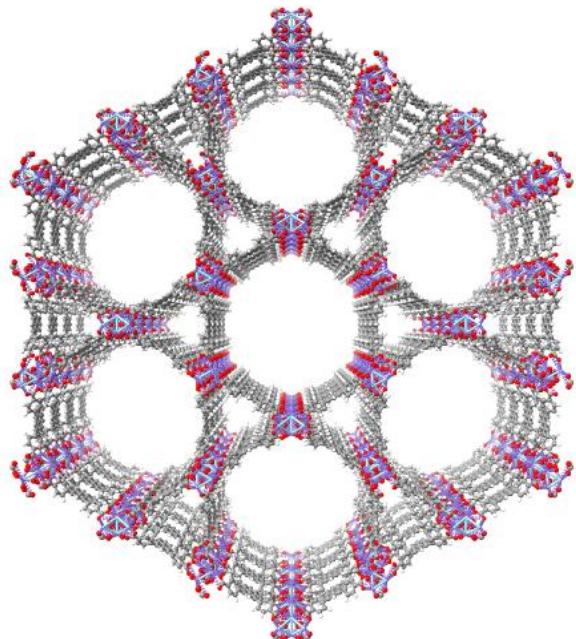
Covalent organic frameworks (COFs)



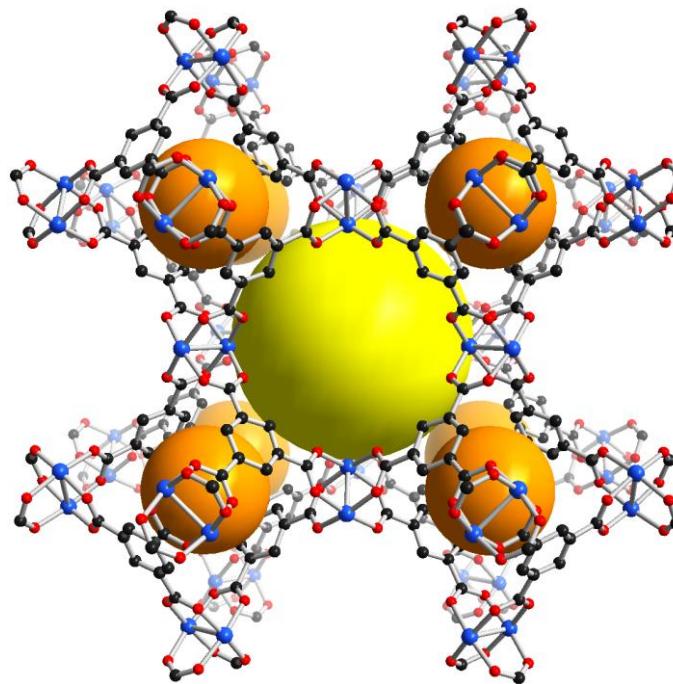
Zeolitic imidazolate frameworks (ZIFs)



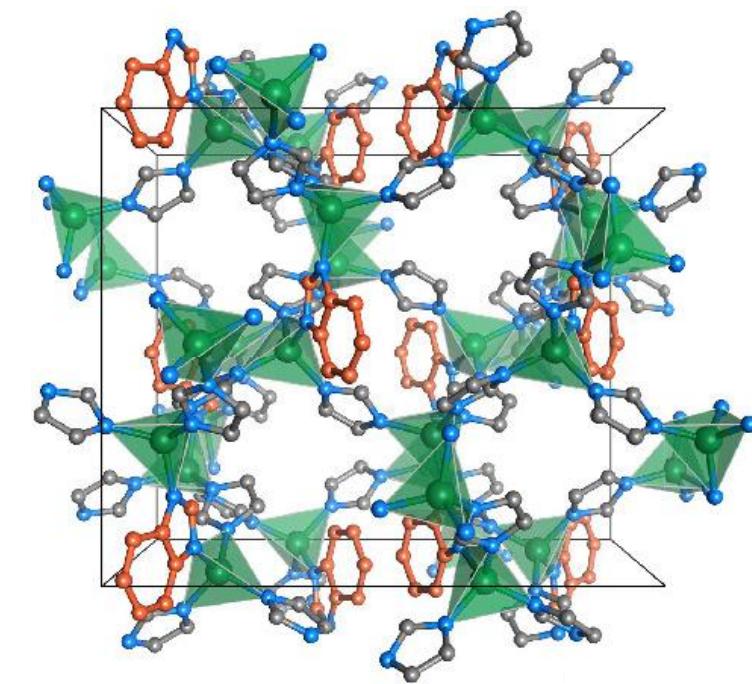
Some common MOFs



NU-1000
Zirconium-based
(yellow)

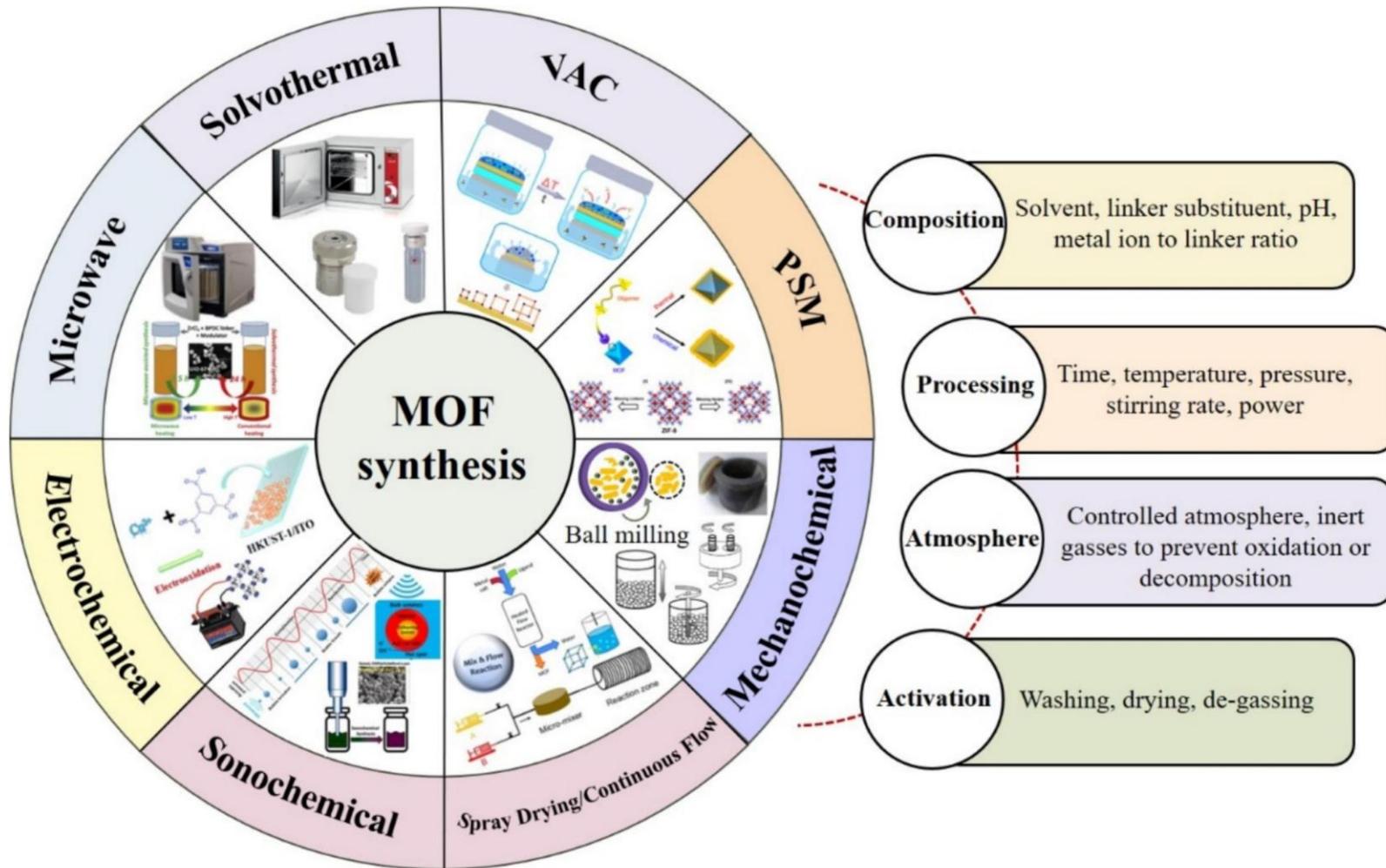


HKUST-1
Copper-based
(blue)

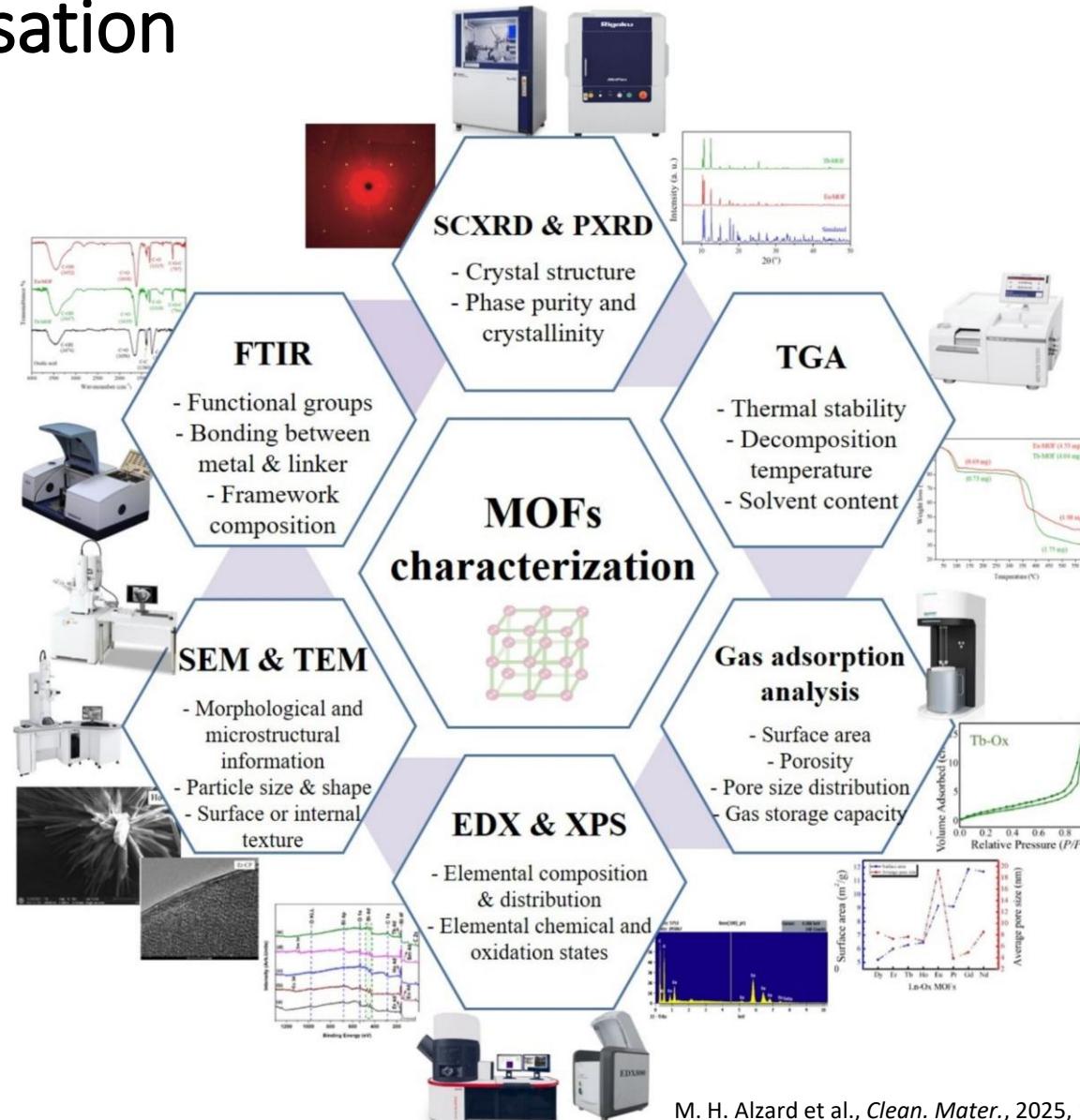


ZIF-62 (Zn/Co)
Zinc or cobalt-based
(colourless or purple)

MOF synthesis



MOF characterisation

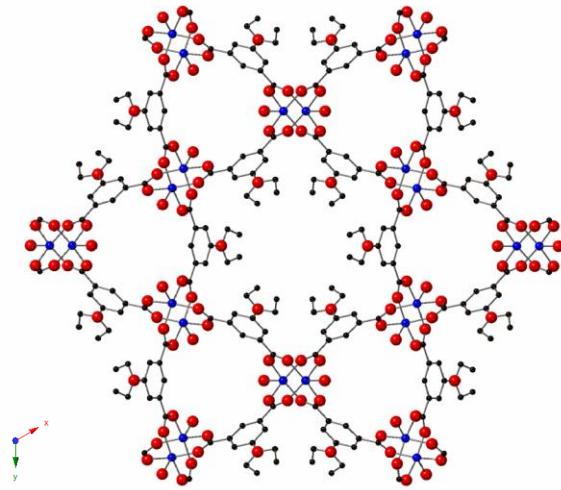


M. H. Alzard et al., *Clean. Mater.*, 2025, **16**, 100314.

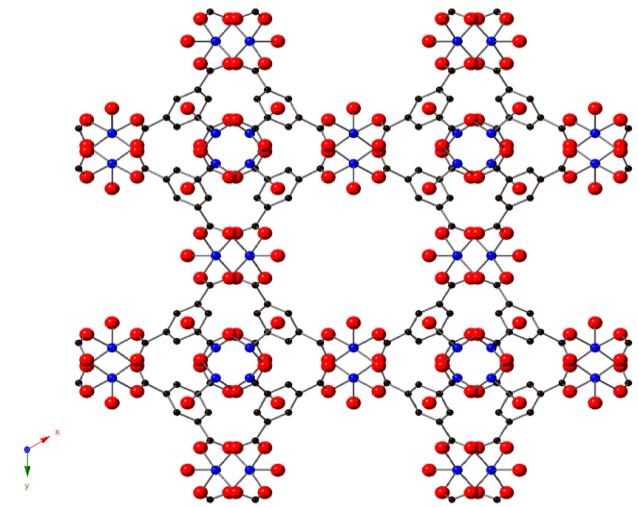
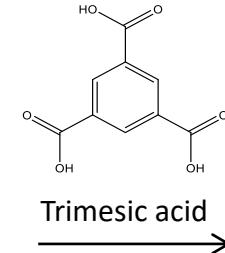
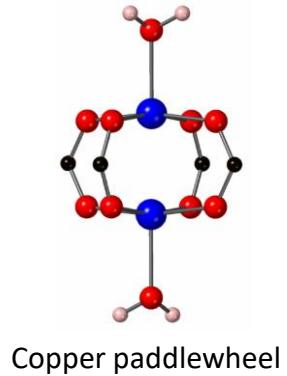
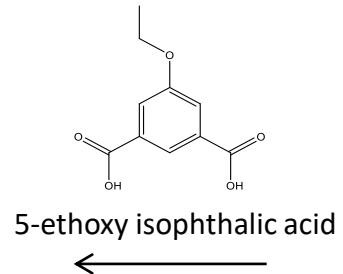
Tuning MOF properties

- Metal centres
 - Structure type – change coordination
 - Catalytic properties, *e.g.*, open metal sites
- Organic linker
 - Essentially endless possibilities
 - Structure type – change denticity
 - Maintain structure type – change pore dimensions (IRMOFs)
 - Chemical environment in pores – hydrophobicity, catalytic, polarity, specific chemical functionalities
- Synthesis and post-synthetic modifications
- Defect engineering

Water stability in copper MOFs



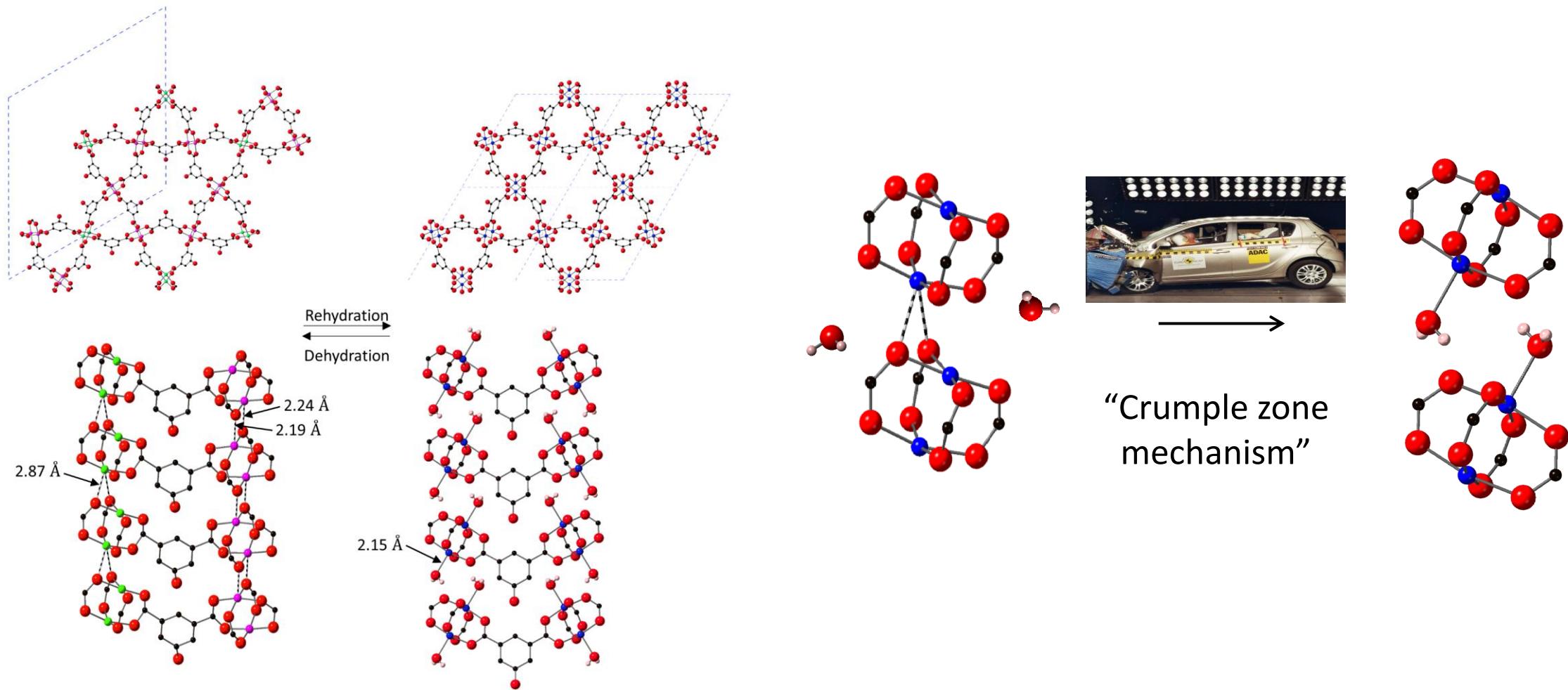
STAM-17-OEt



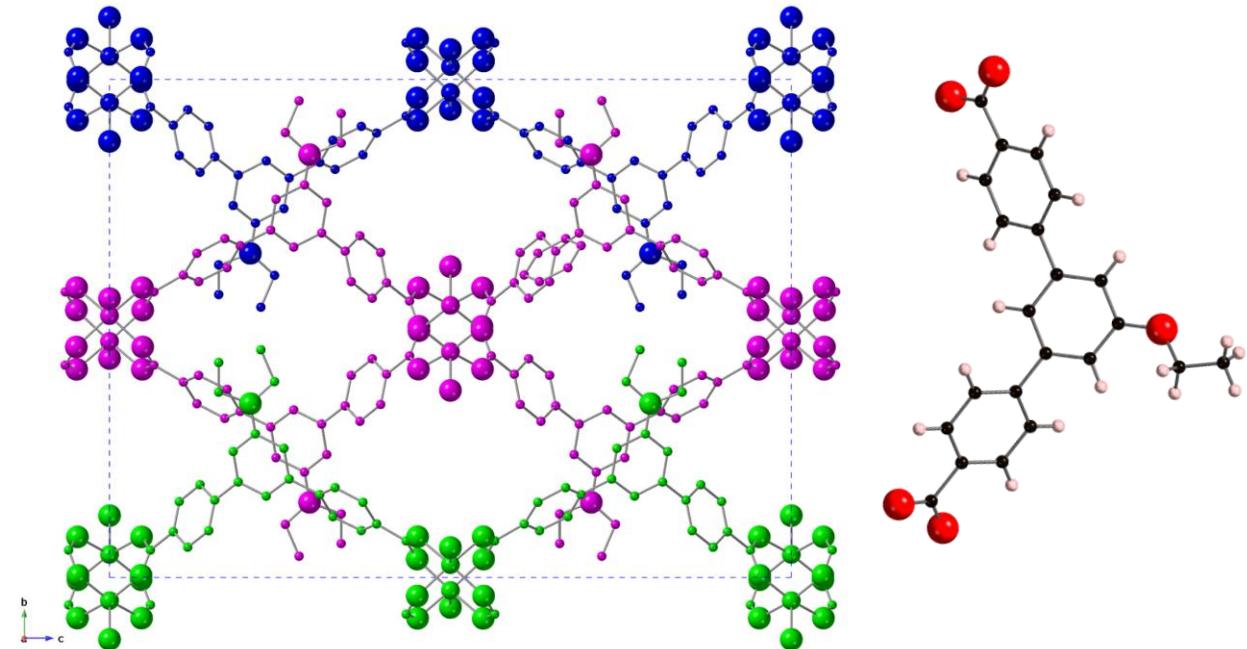
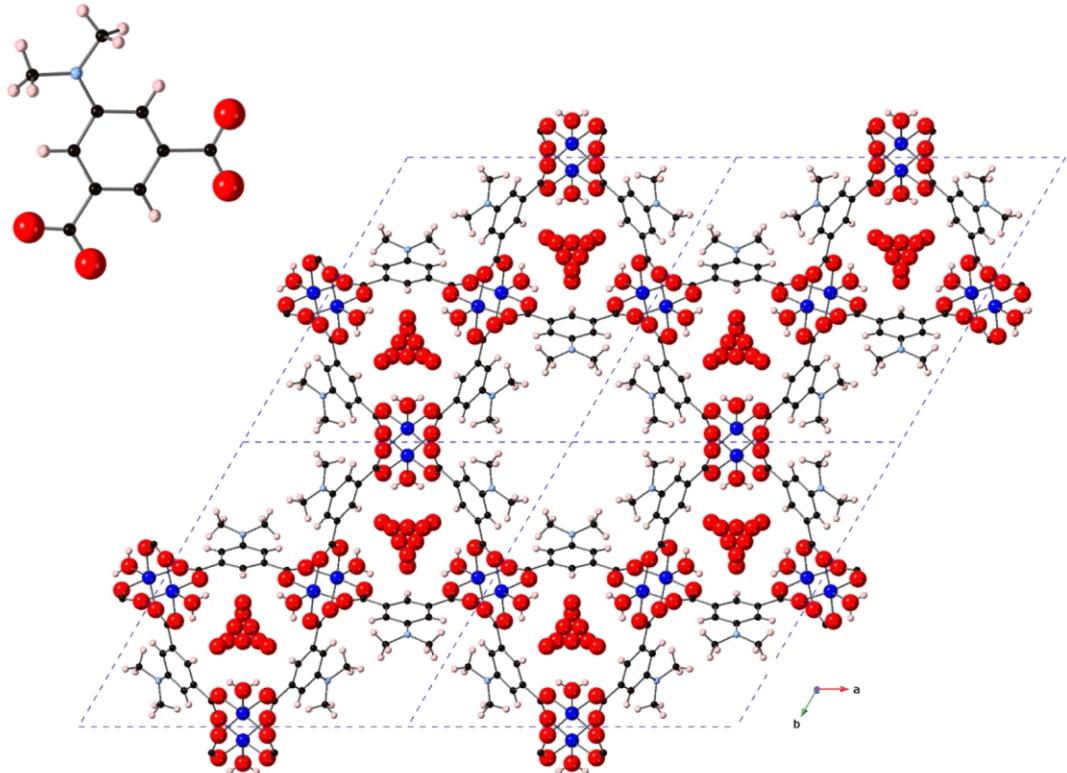
HKUST-1

- Copper paddlewheel dimer units present in both structures.
- Similarities in syntheses and structures allow the comparison of properties.
- HKUST-1 is affected by long-term water instability, is STAM-17-OEt similarly affected?

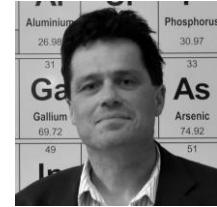
Hemilability in copper MOFs



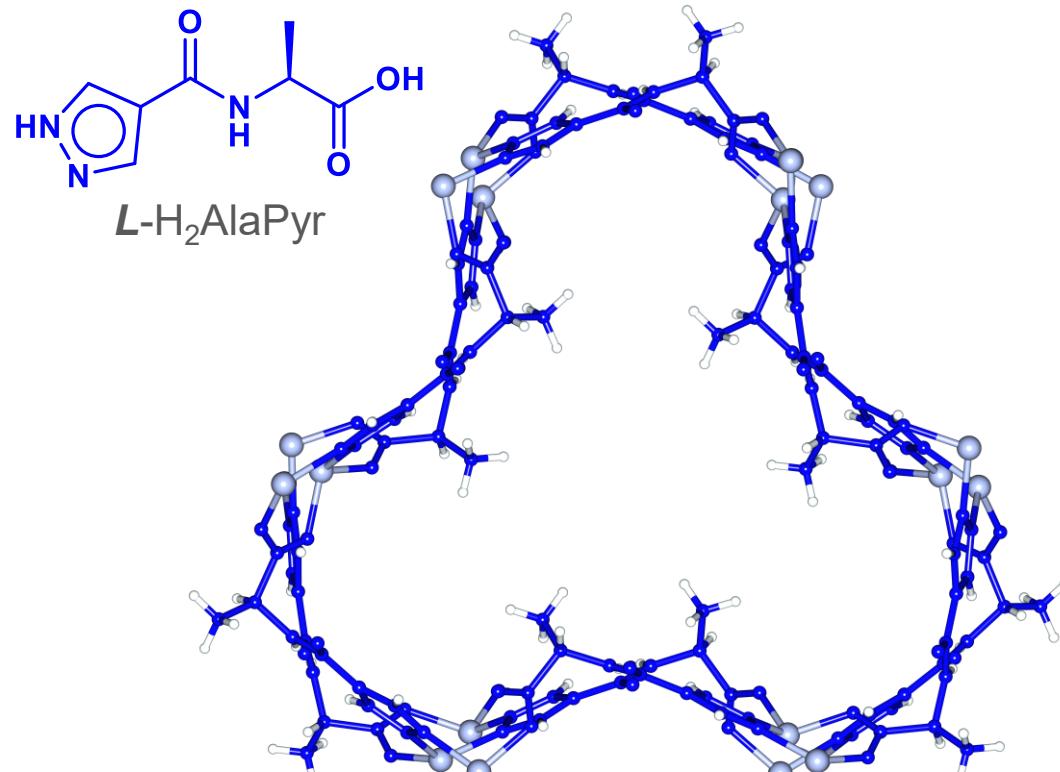
Structural variations



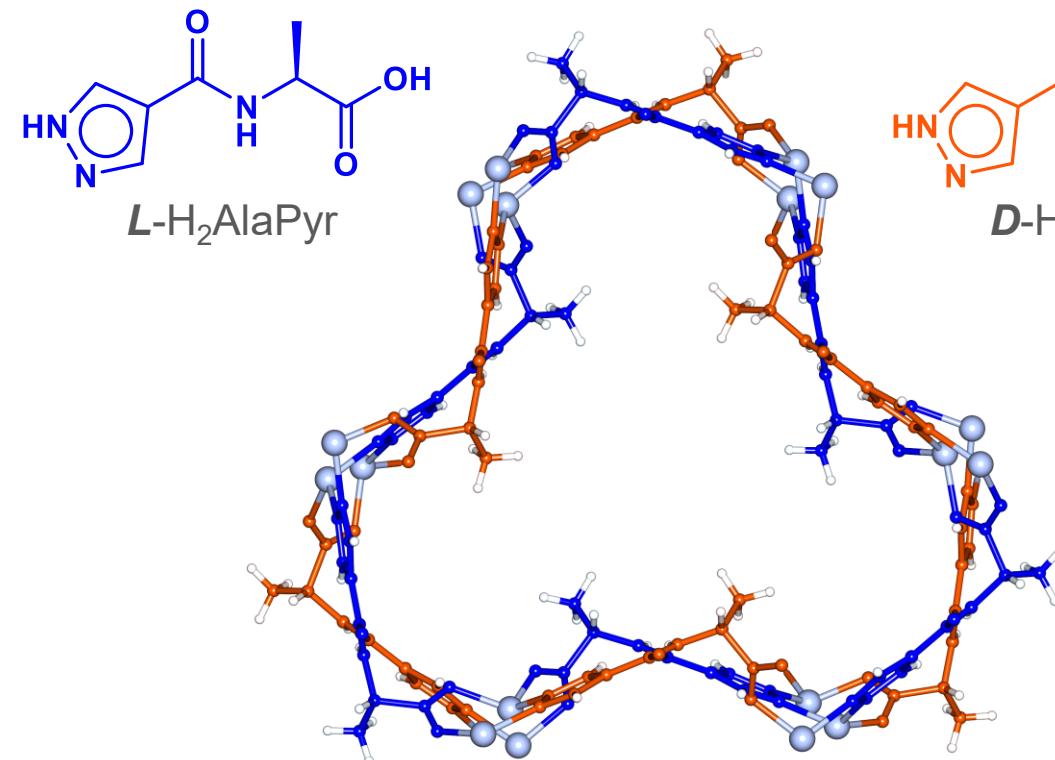
Ordered sidechains of flexible MOFs



Matt
Rosseinsky

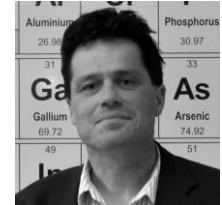
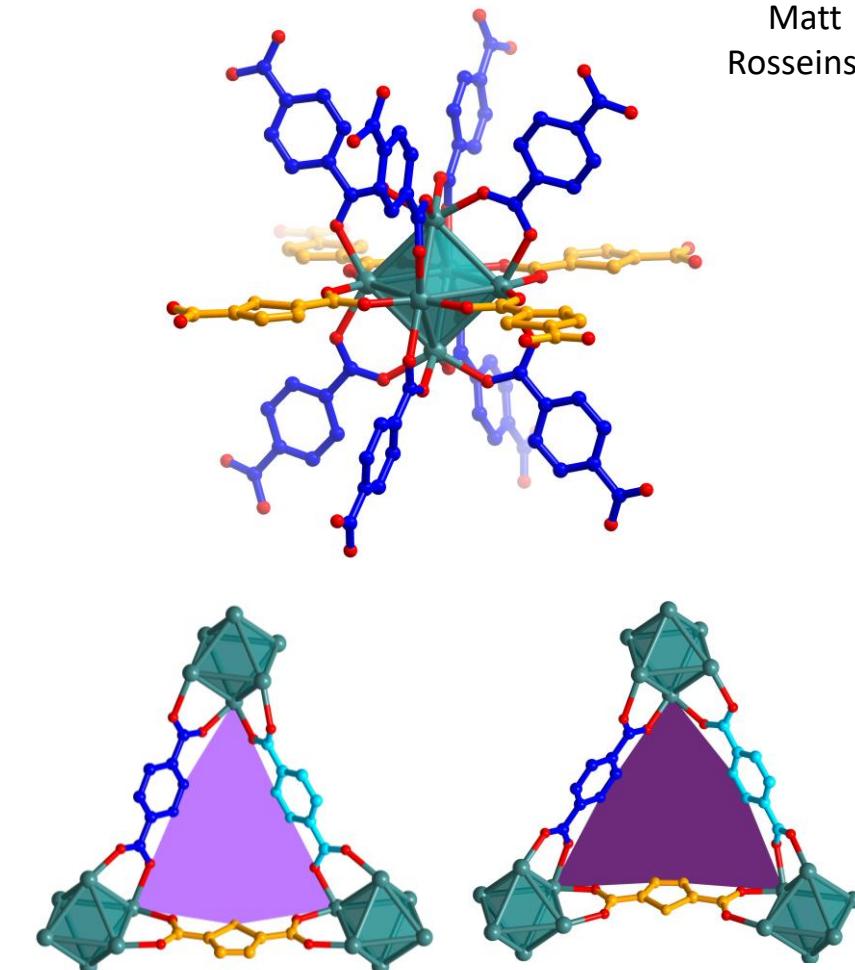
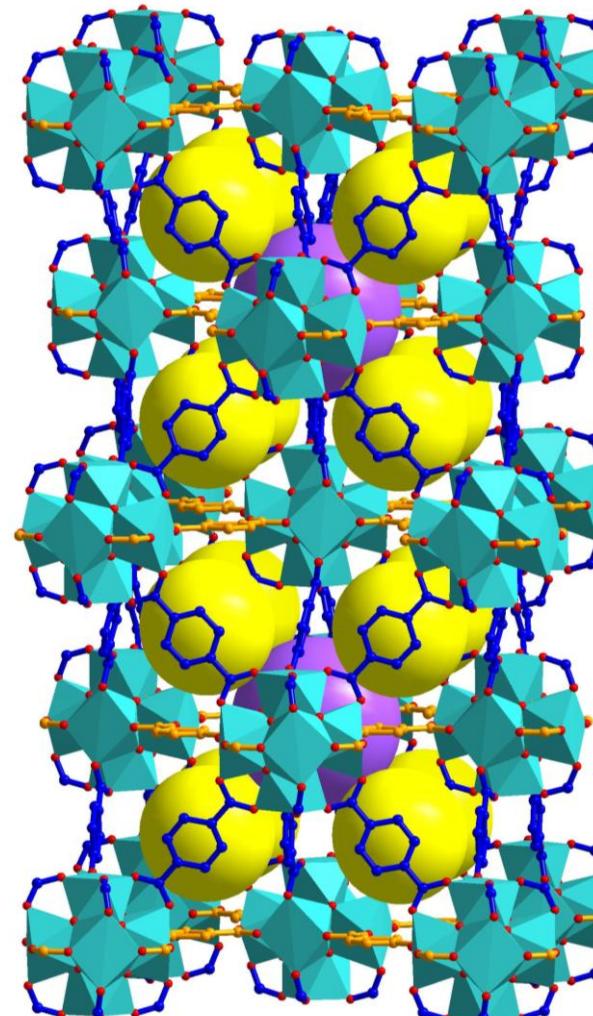
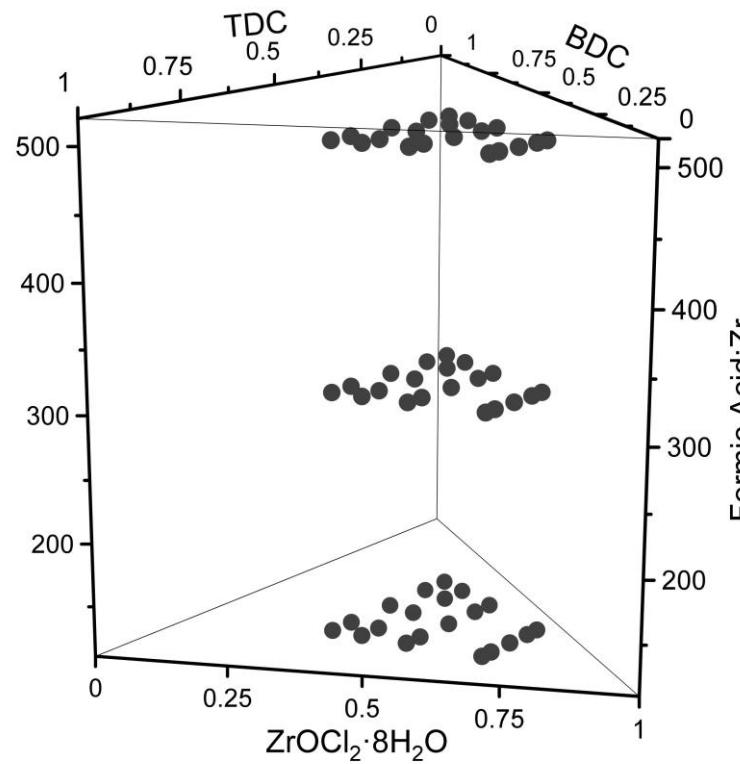
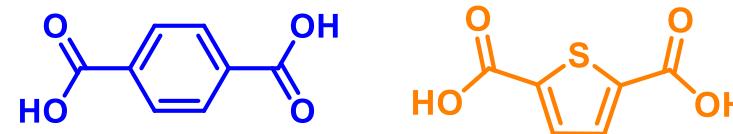


$[\text{Zn}_2(\text{L-AlaPyr})_2]$



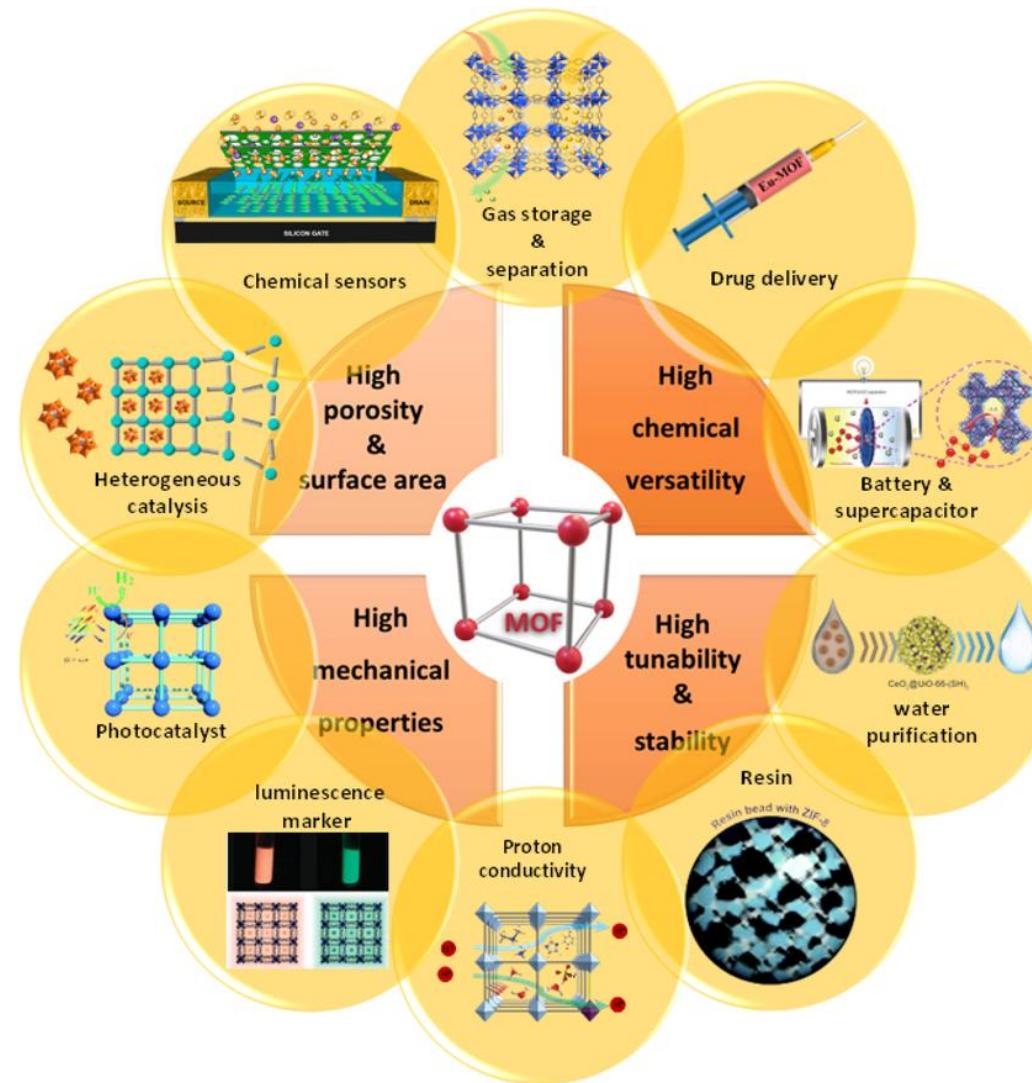
$[\text{Zn}_2(\text{L-AlaPyr})(\text{D-AlaPyr})]$

Ordered straight and bent linkers on Zirconium MOFs

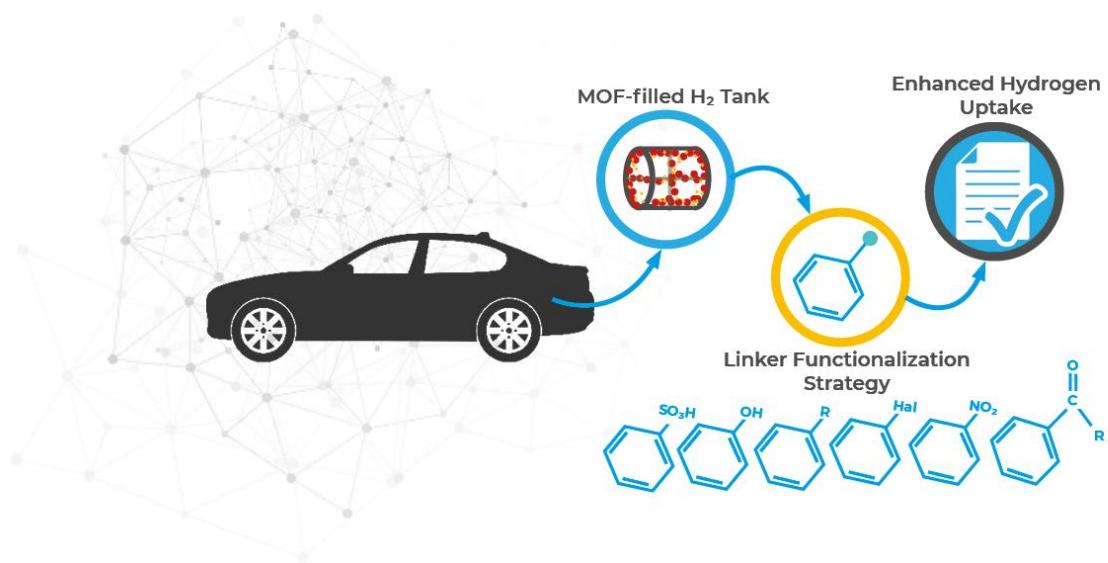


Aluminum	26.98	Phosphorus	30.97
31	33		
Gallium	69.72	As	
49	74.92		51

General applications of MOFs



Adsorptive applications of MOFs

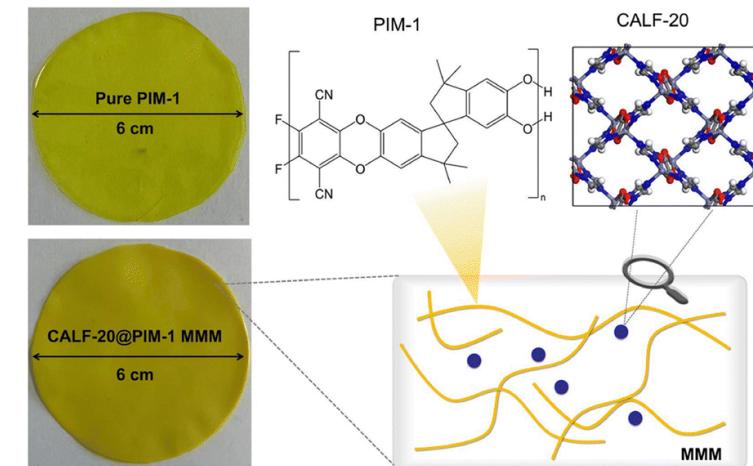
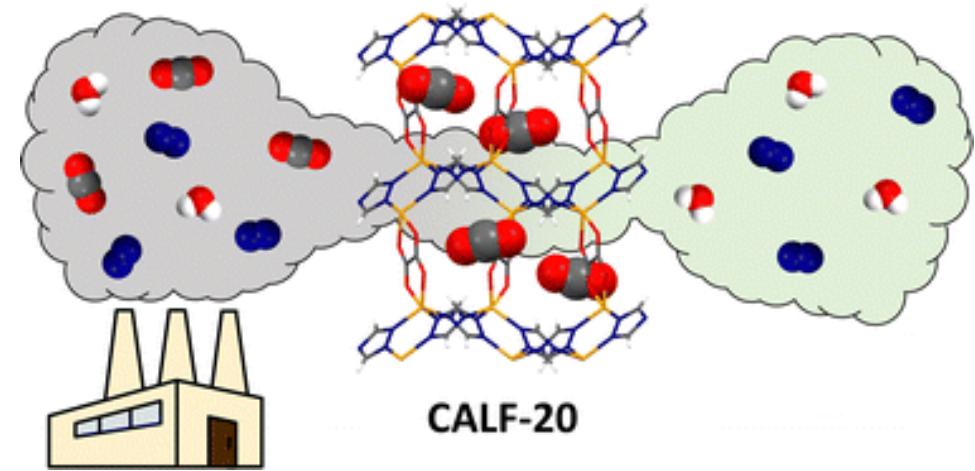
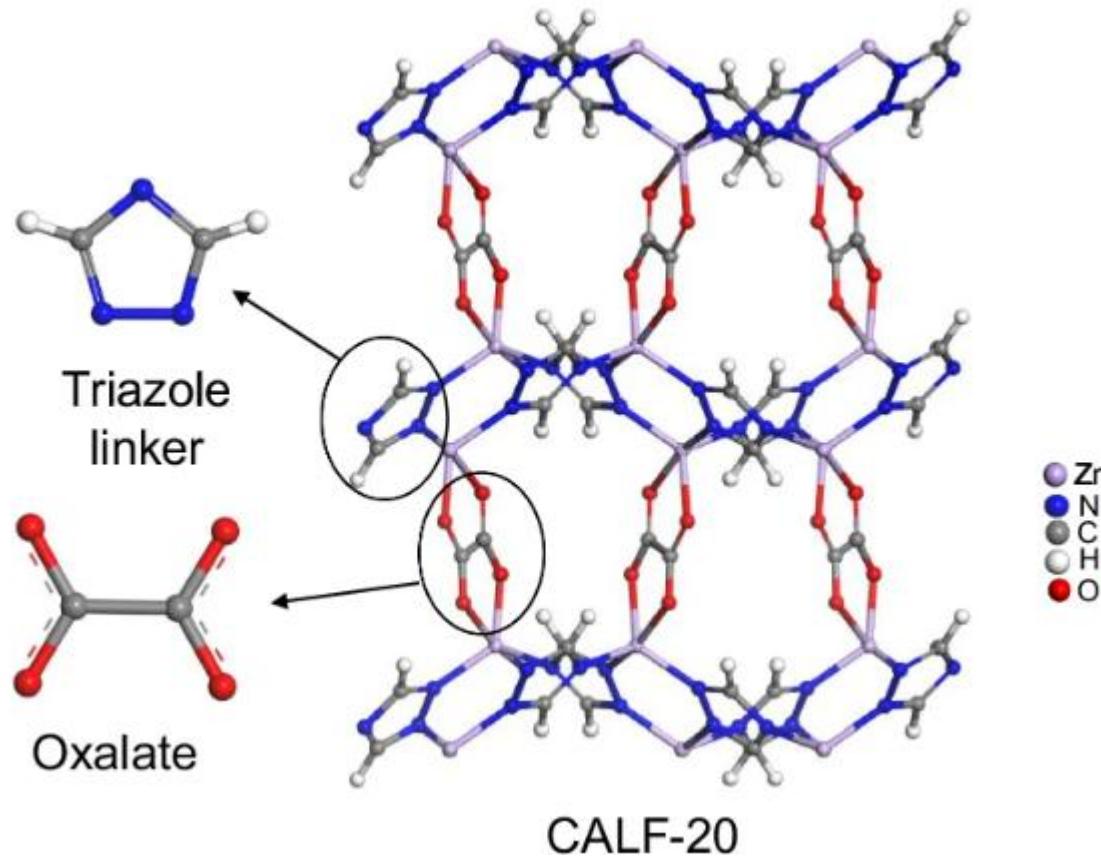


Vehicular fuel storage

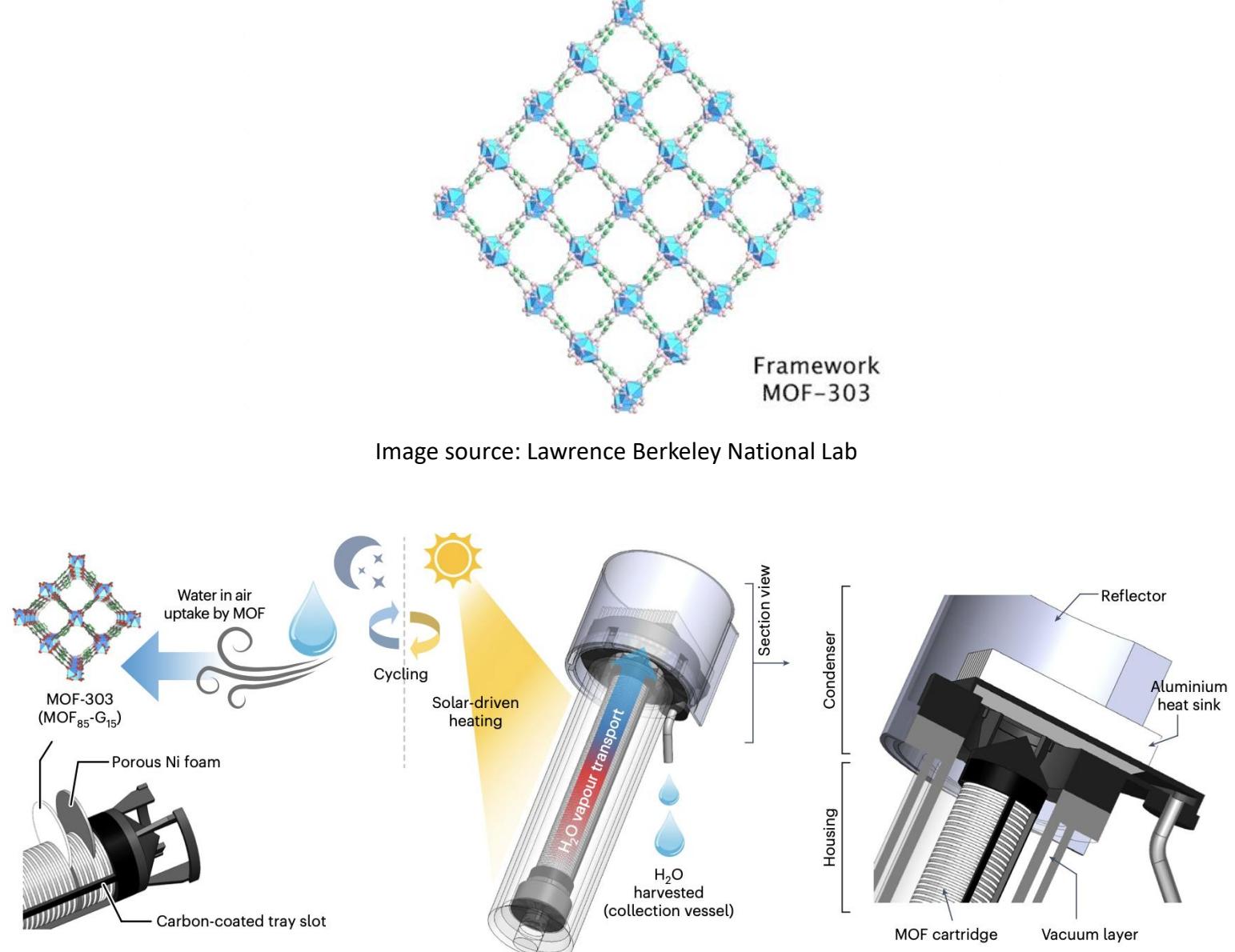
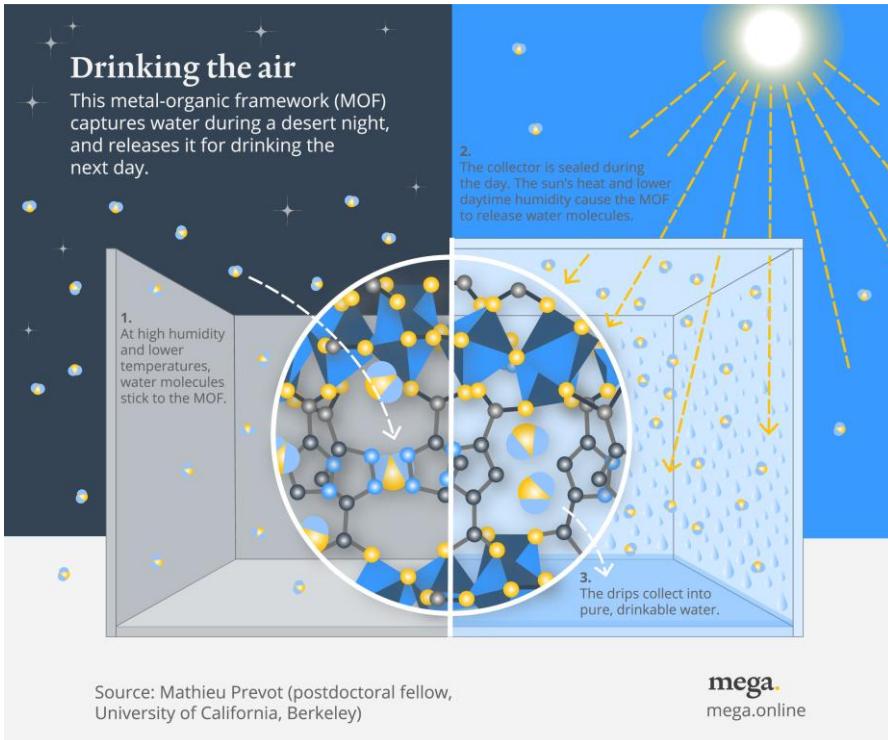


Prevention of fruit over-ripening

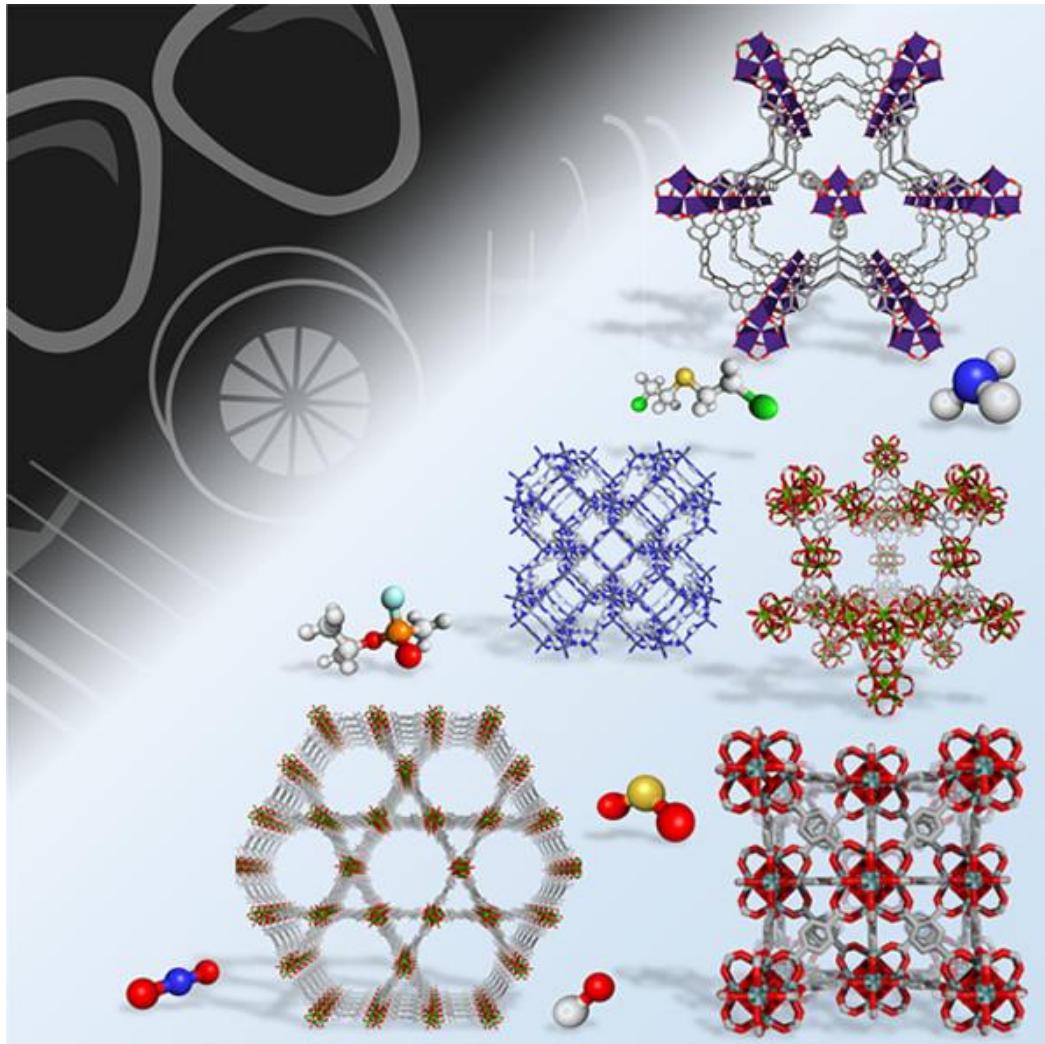
Carbon dioxide capture



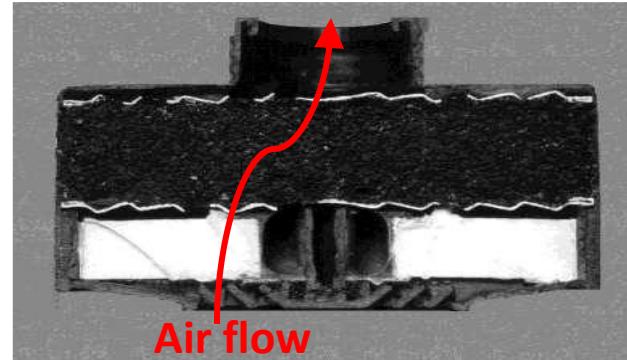
Water harvesting



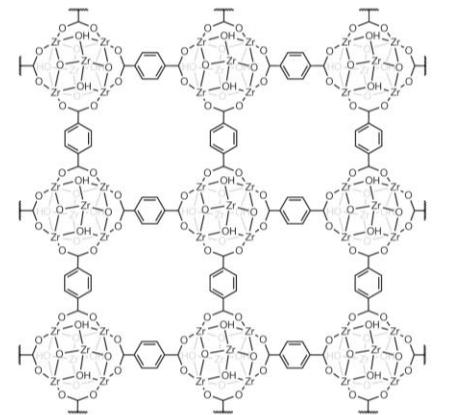
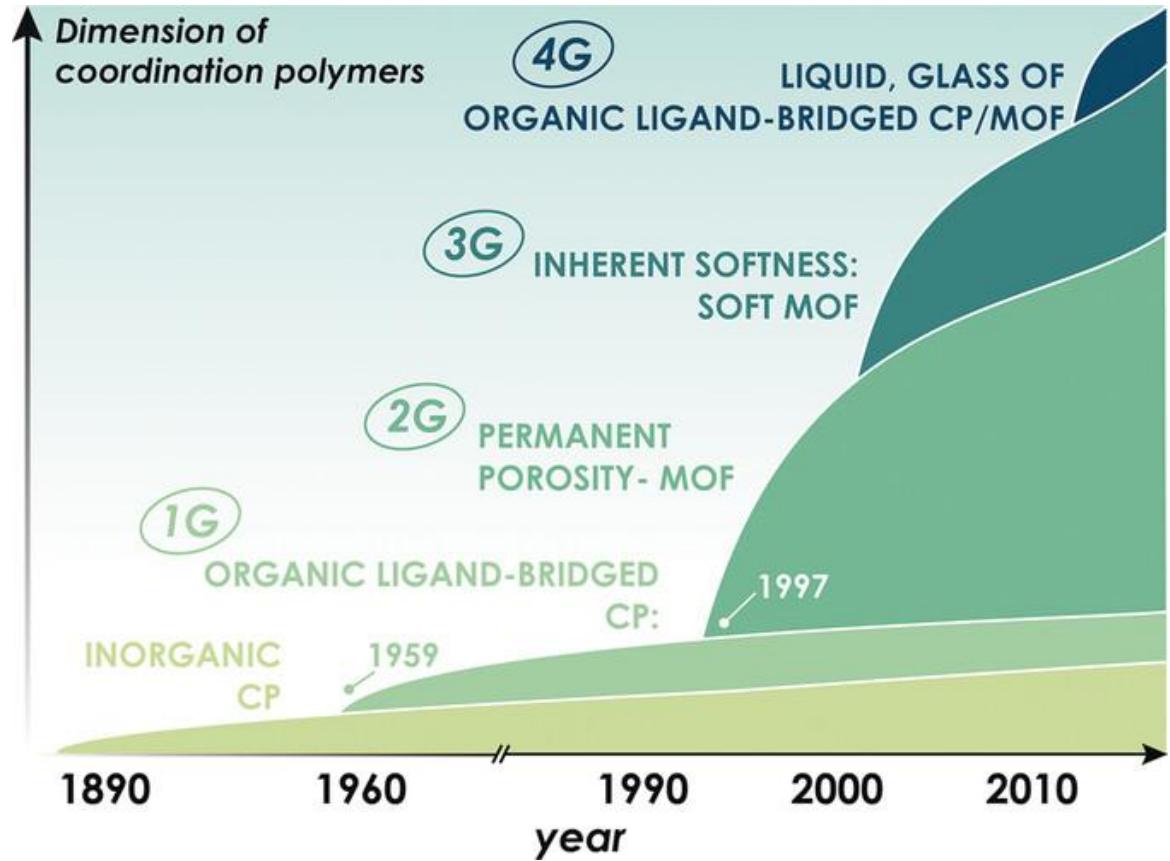
Toxic gas adsorption



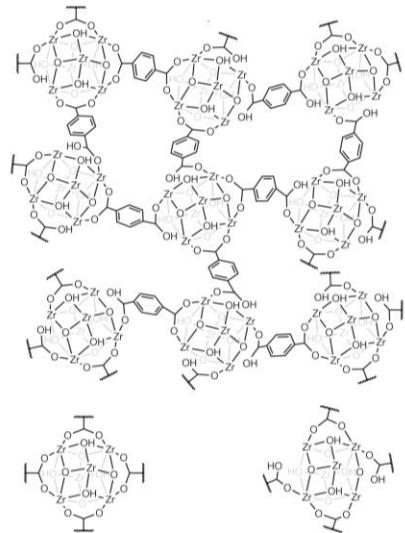
T. Islamoglu et al., *Chem. Rev.*, 2020, **120**, 8130-8160.



The changing state of MOFs

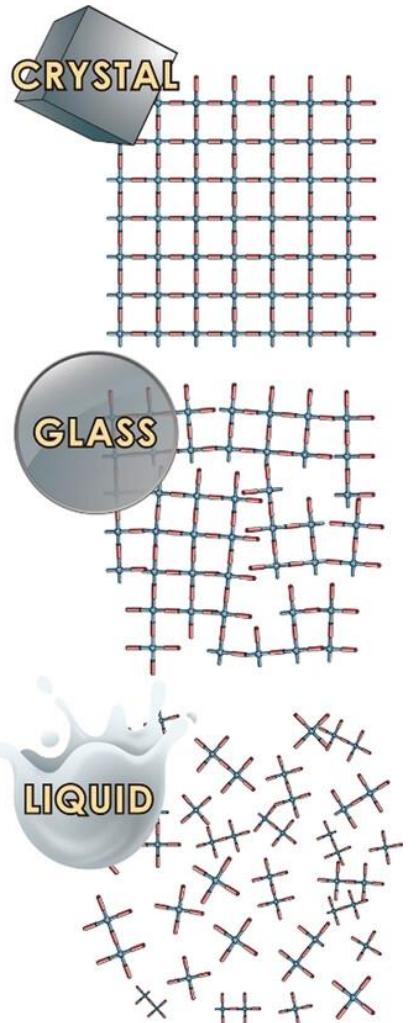


Uniaxial compression or ball milling

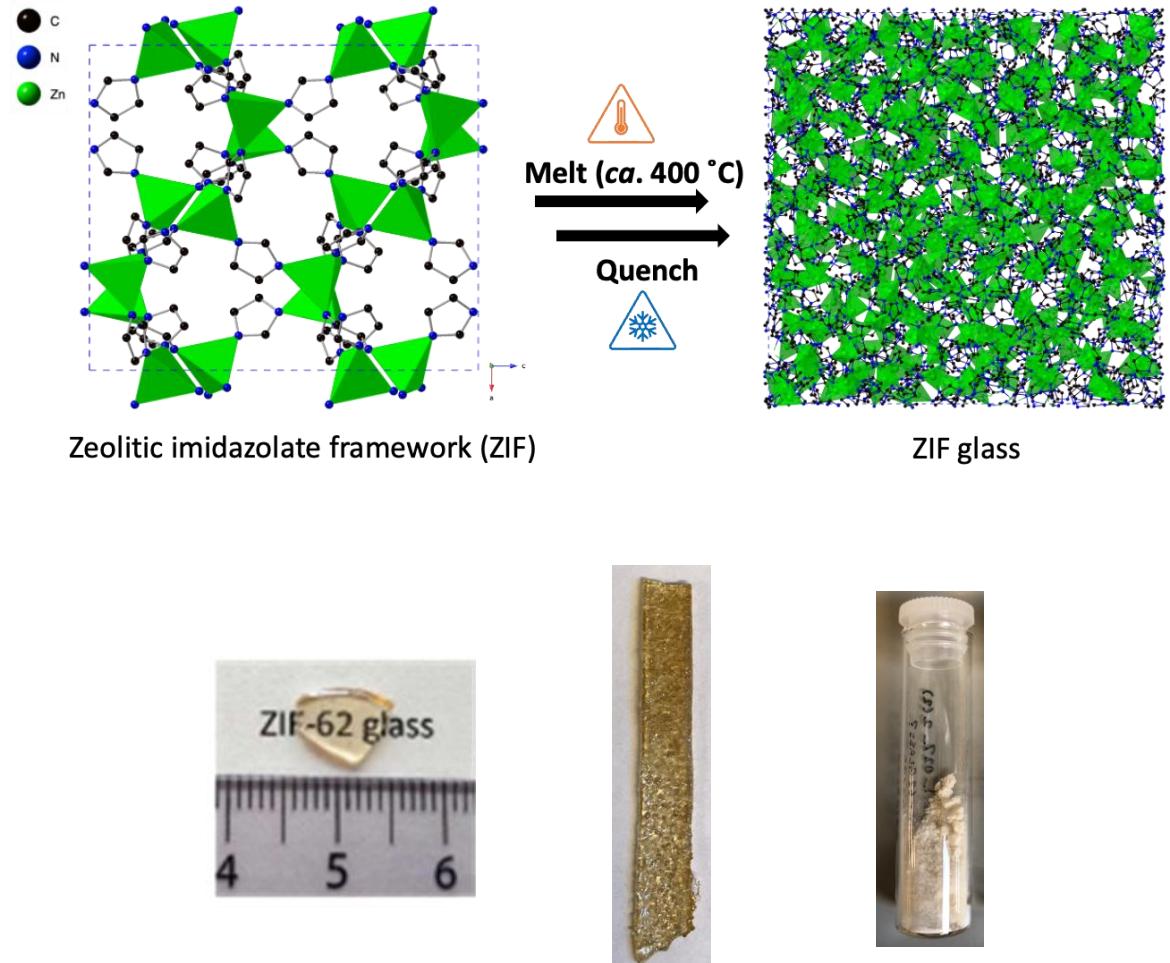
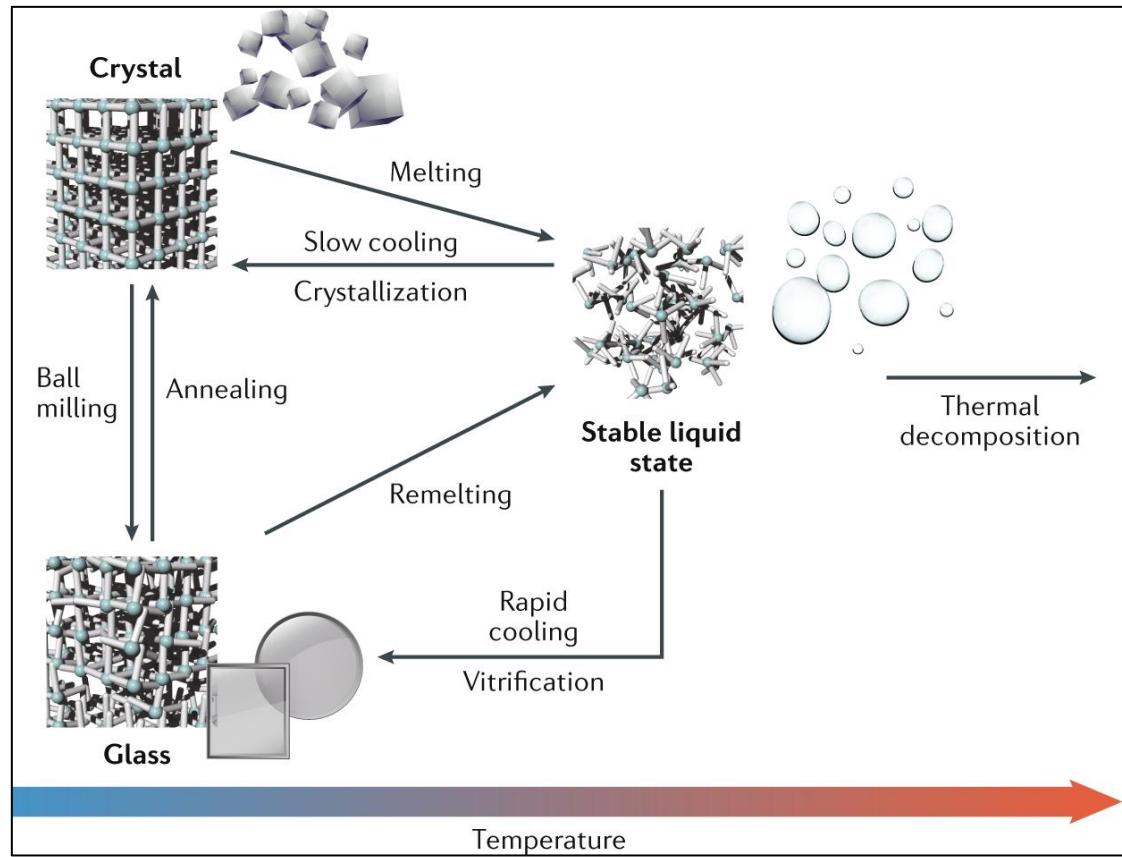


Crystalline

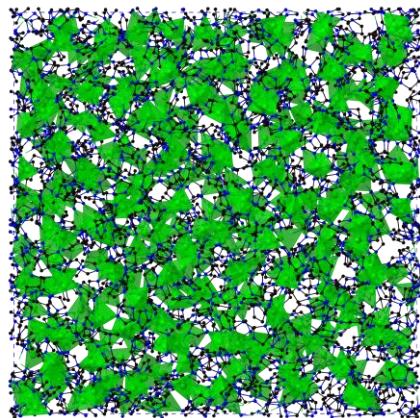
Disordered



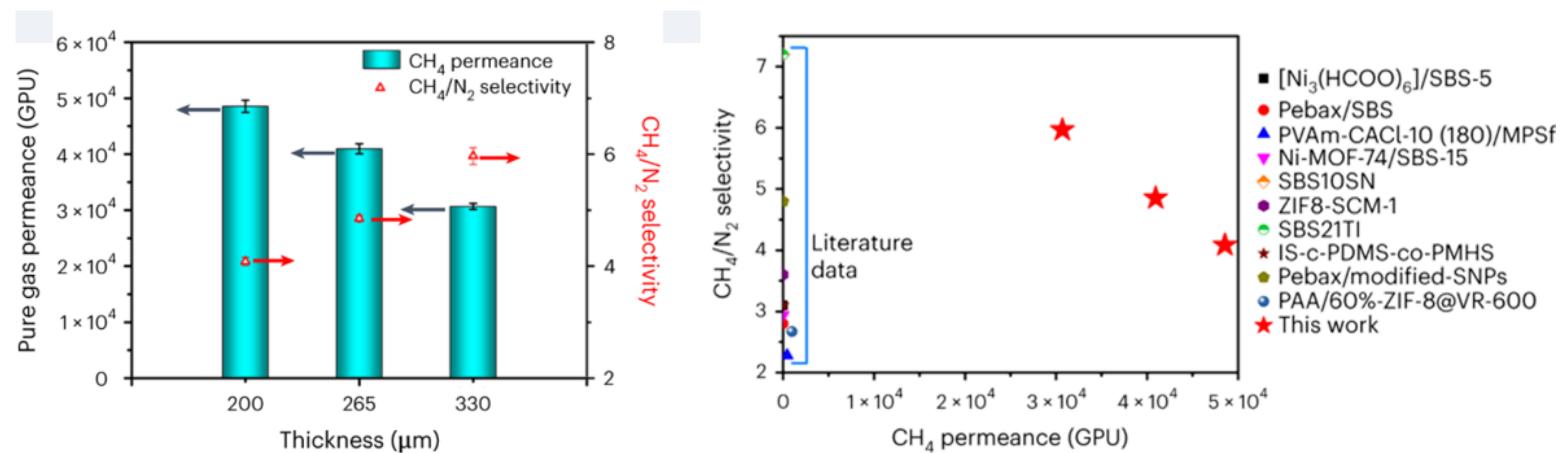
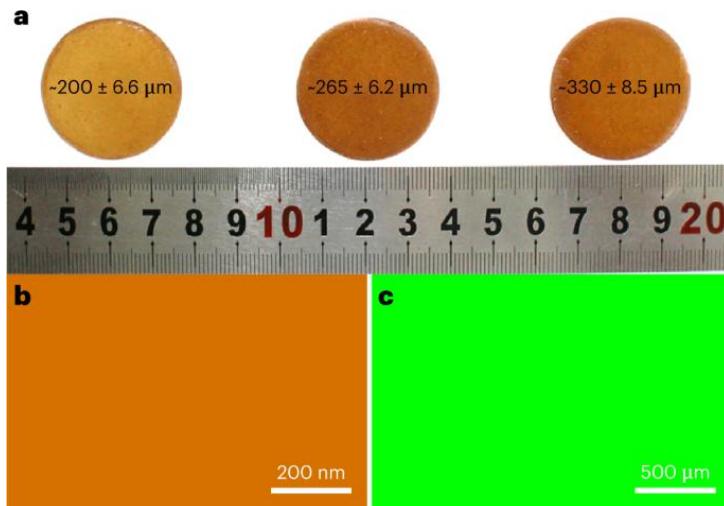
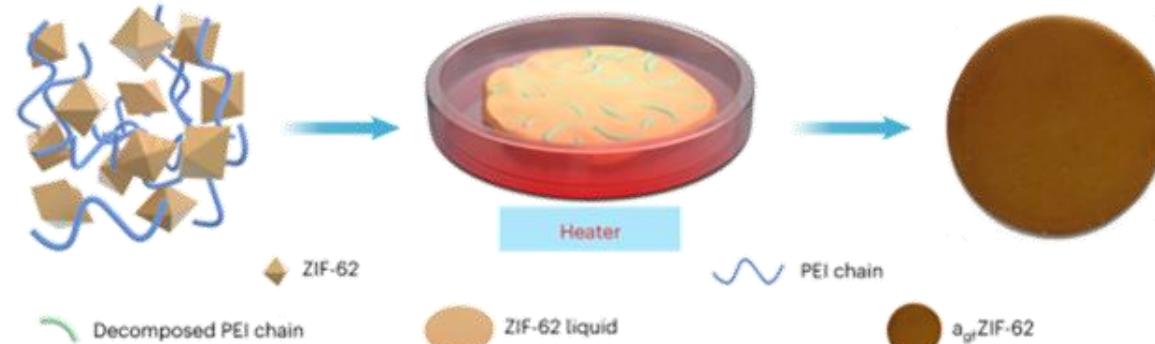
MOF glasses



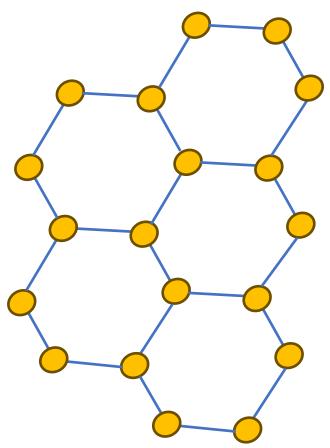
Membrane materials in chemical separations



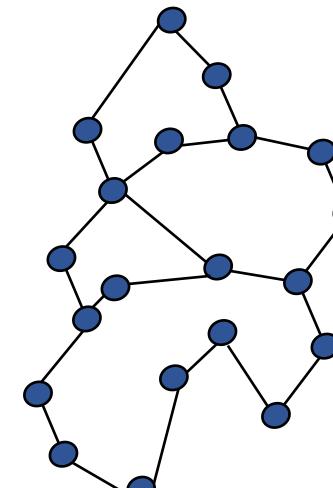
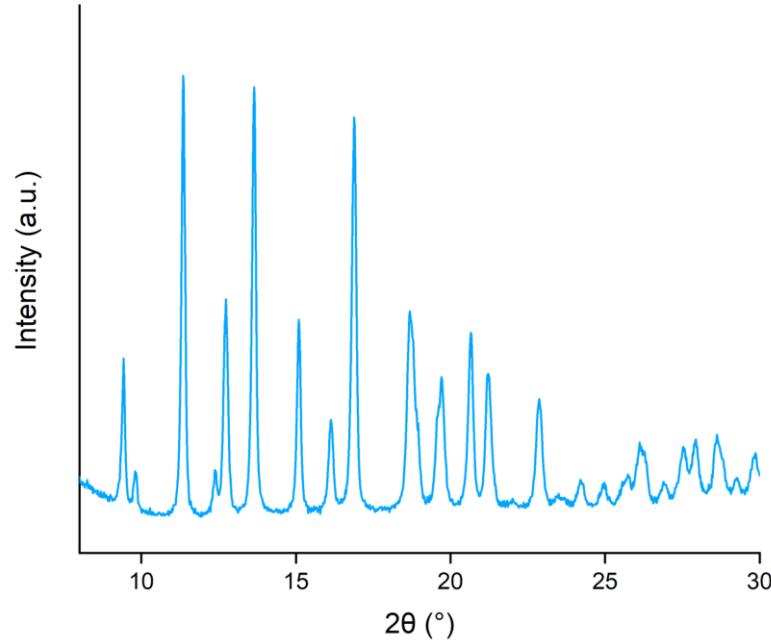
ZIF glass



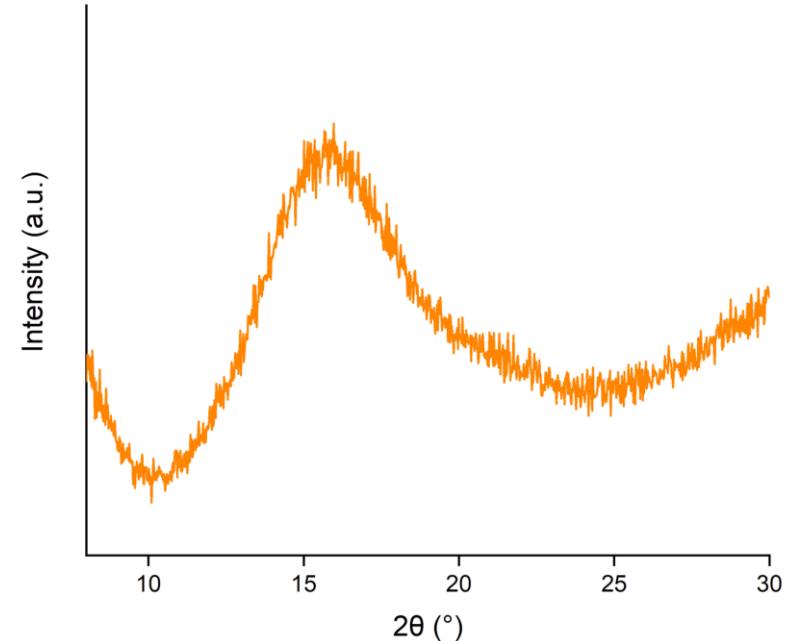
Crystalline v non-crystalline



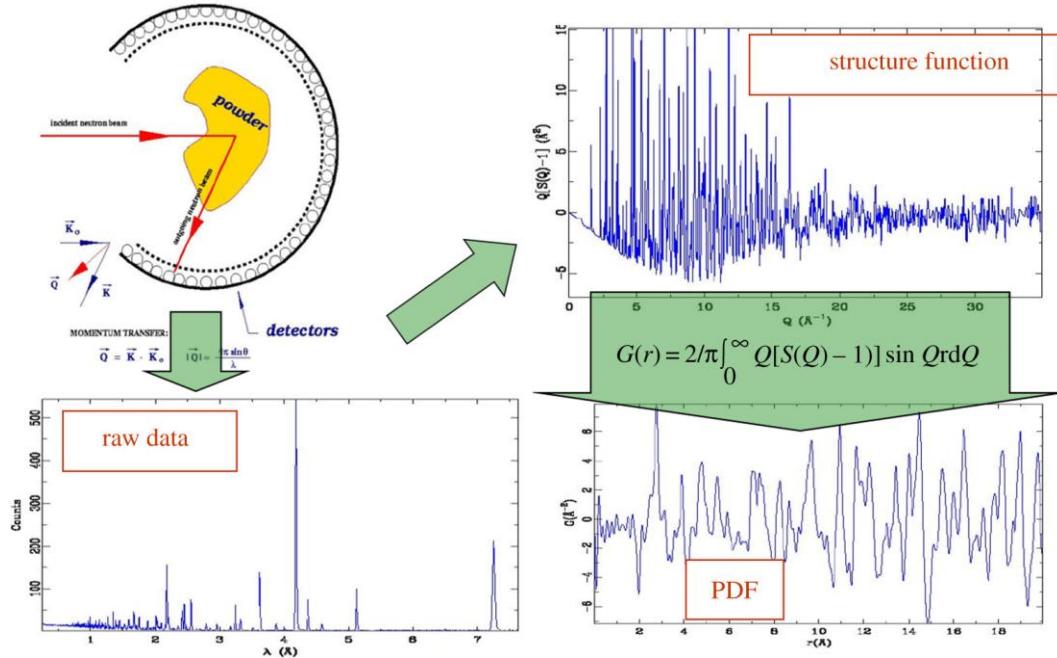
Crystalline



Non-crystalline



Total scattering and pair distribution function

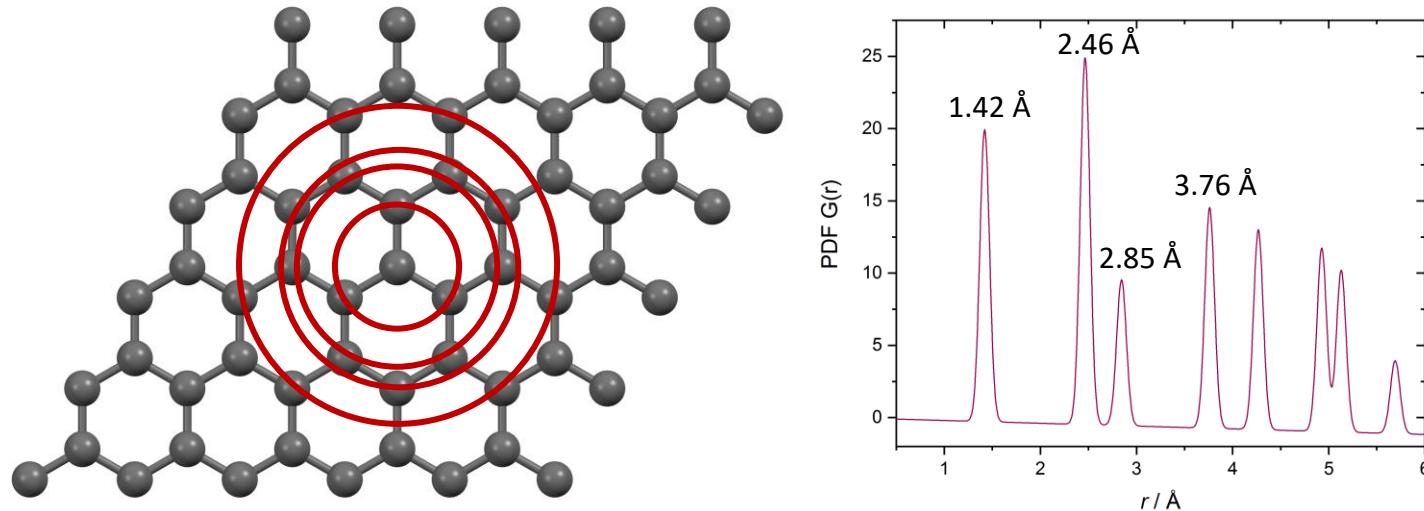


- **Total scattering** (Bragg + diffuse) experiments are similar to normal diffraction experiments.
- **Diffuse scattering provides information on the short-range structure of materials.**
- Total scattering can even be used for amorphous solids, liquids and glasses.

- For disordered crystalline materials, it helps characterise the periodic structure **and** the deviations from long-range order.

Pair distribution function (PDF)

- Total scattering is mathematically related to the PDF *via* Fourier transform (like the reciprocal and direct lattices)

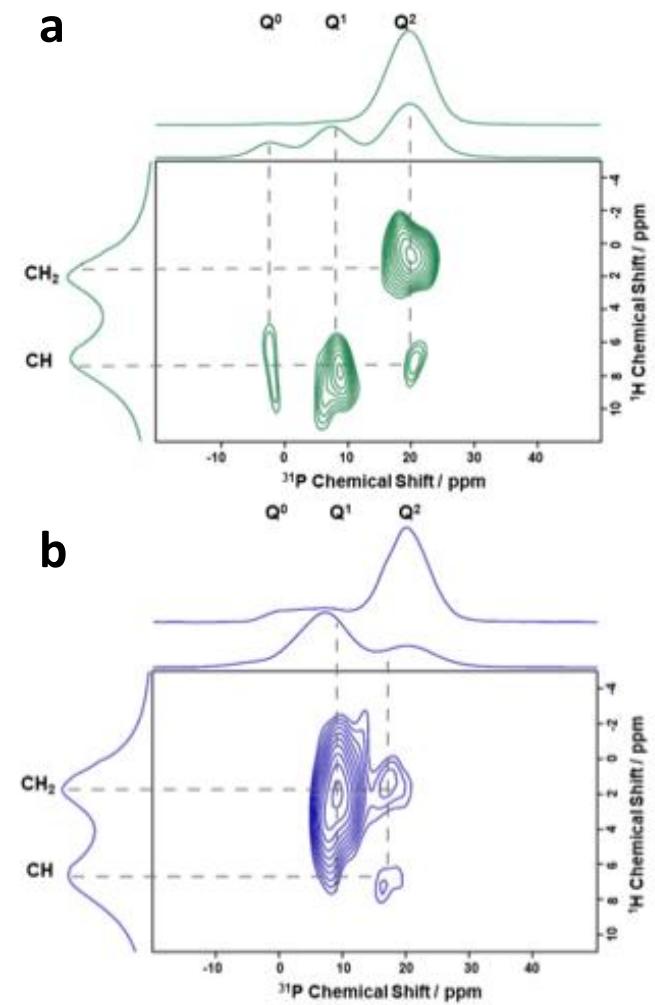
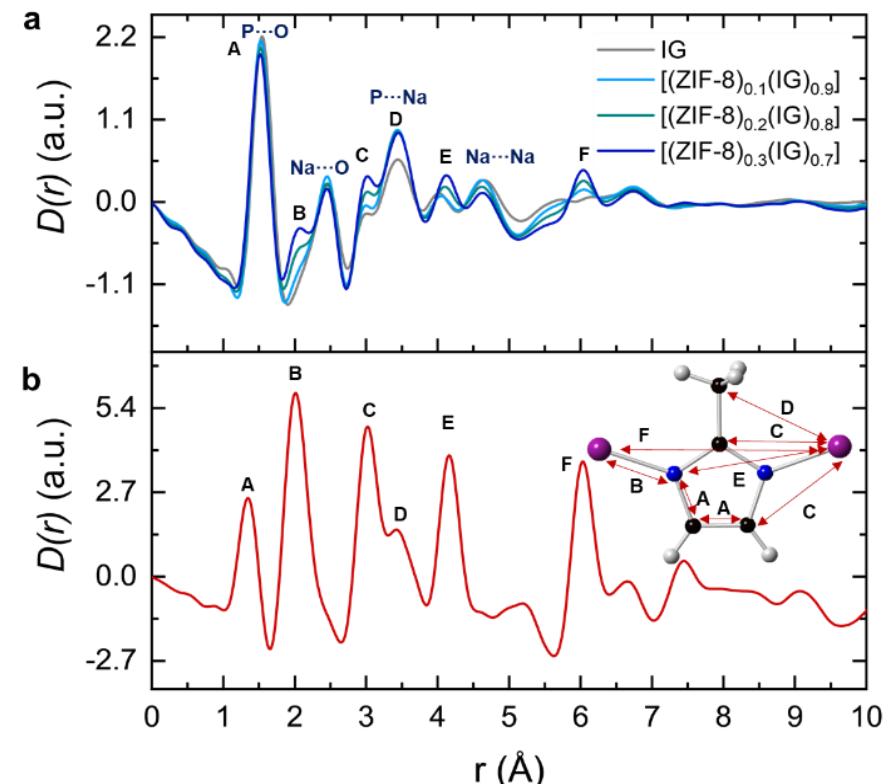
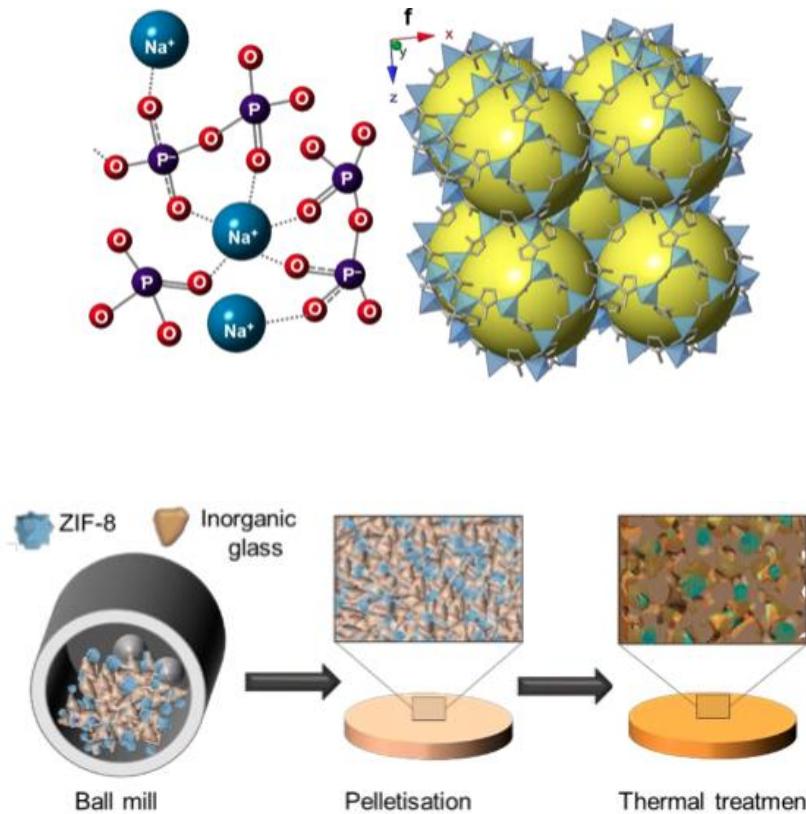


- Peak positions give interatomic separations
- Area provides information about coordination number

Interfaces in MOF crystal-glass composites



Frédéric Blanc



Summary

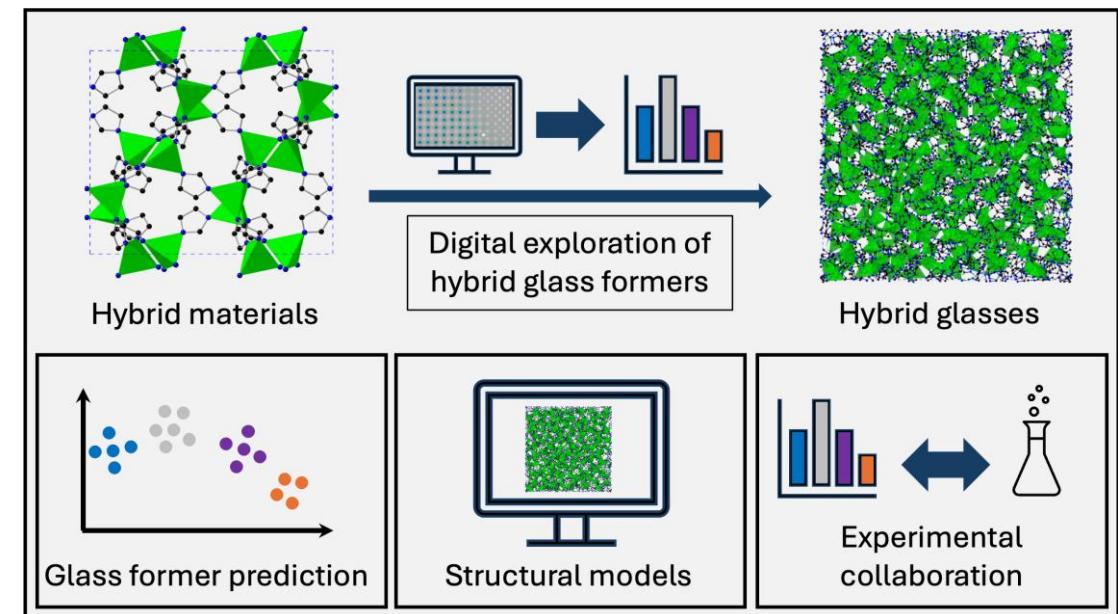
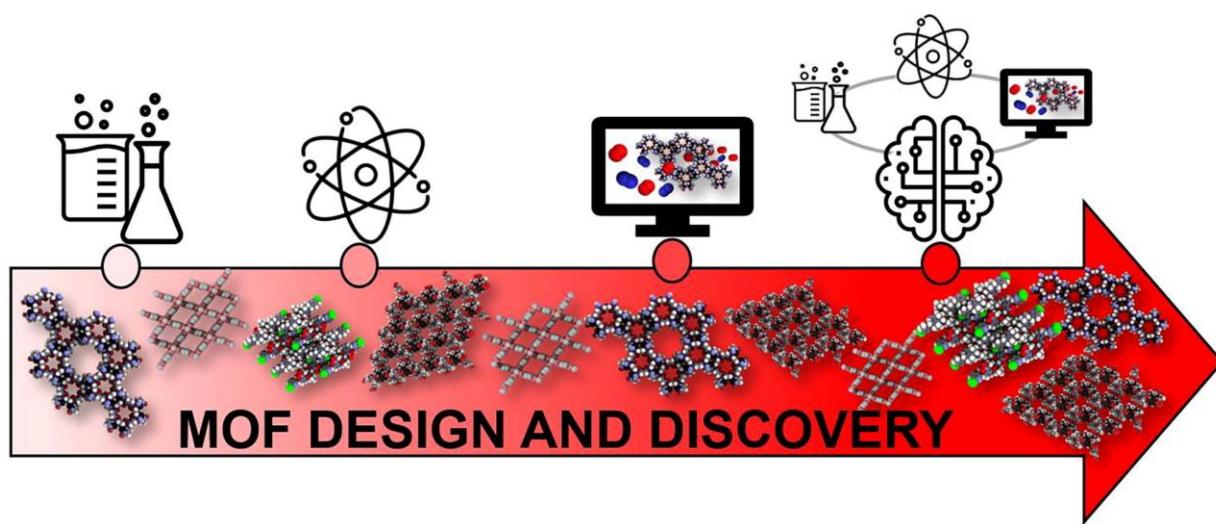
- MOFs are framework materials made from metal ions or clusters linked by organic ligands, forming porous 3D networks with exceptionally high surface areas.
- Their structure, pore size, and functionality can be precisely tuned by selecting different metals and linkers, allowing for tailored chemical and physical properties.
- MOFs have wide-ranging uses, including gas storage, separation, catalysis, drug delivery, and sensing – driven by their tuneable porosity and reactivity.

✓ Advantages: large surface areas, tuneability, and versatility.

✗ Challenges: stability (especially in moisture/heat), scalability, and cost of synthesis.

Future perspectives: digital discovery

- Digital materials chemistry goes hand in hand with experimental materials chemistry
- AI and machine learning is driving MOF commercialisation with interest from several start-ups and large companies



Thanks for listening! Questions?

